

# Stereocontrolled Total Synthesis of (–)-Isocelorbicol and Its Elaboration to Natural Dihydro- $\beta$ -agarofuran Esters

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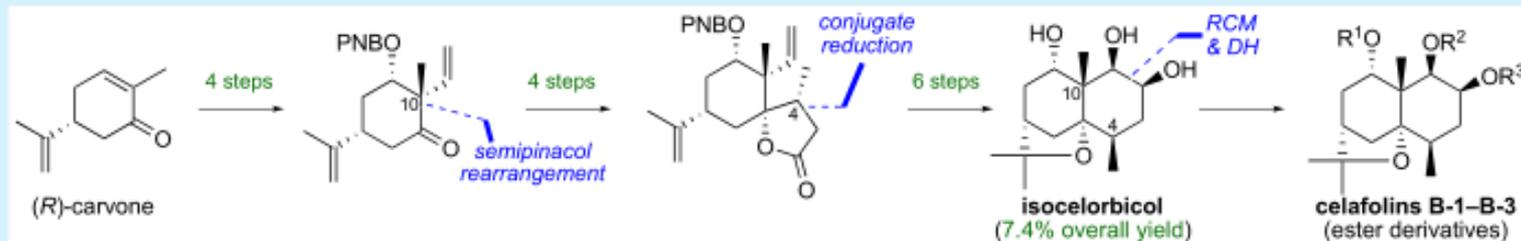
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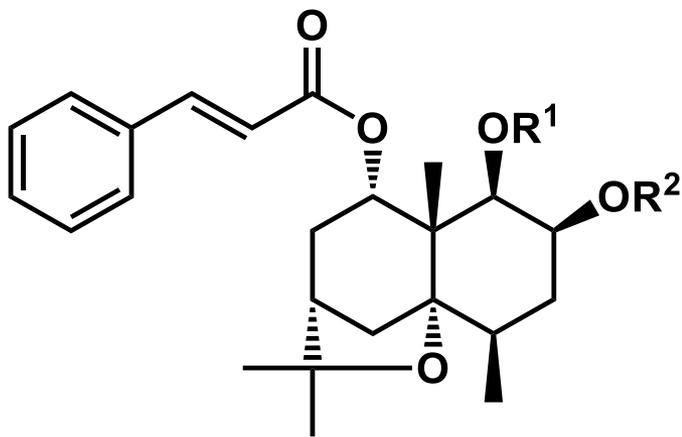
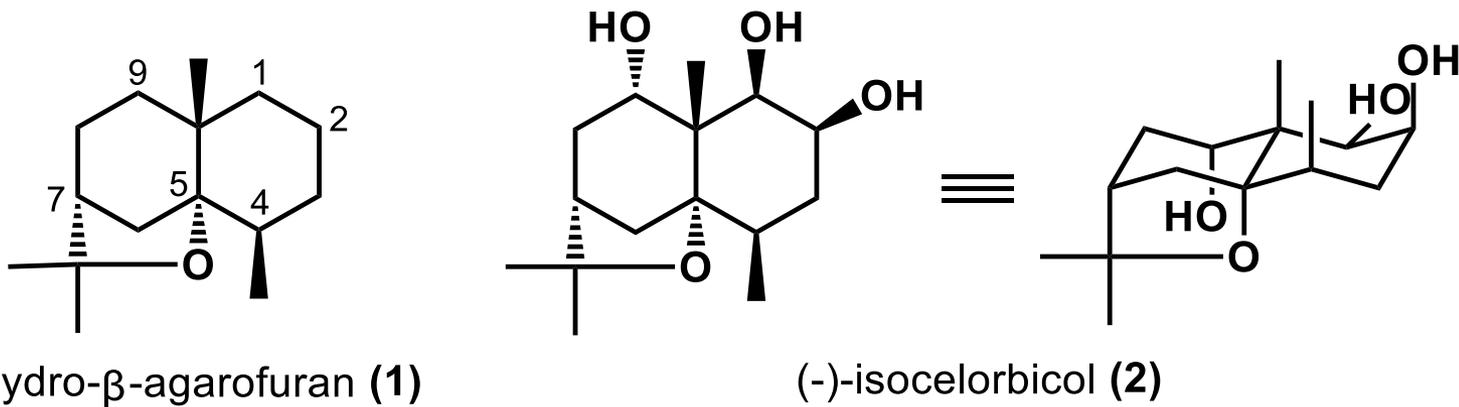
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Article Recommendations

Supporting Information

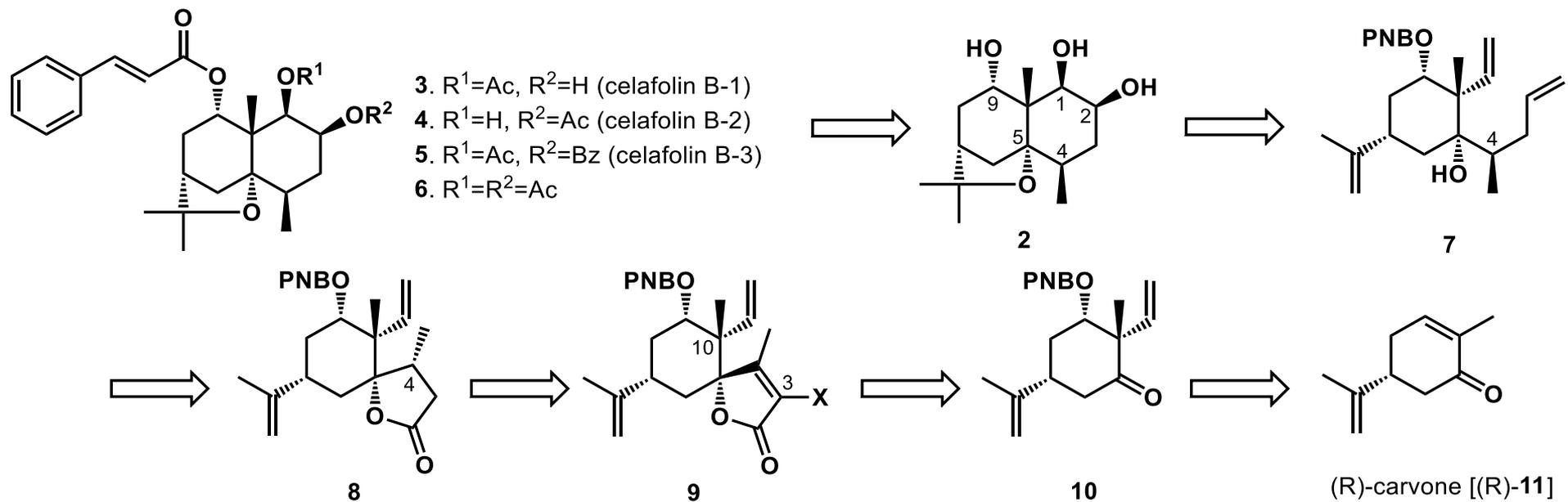


# 背景介绍

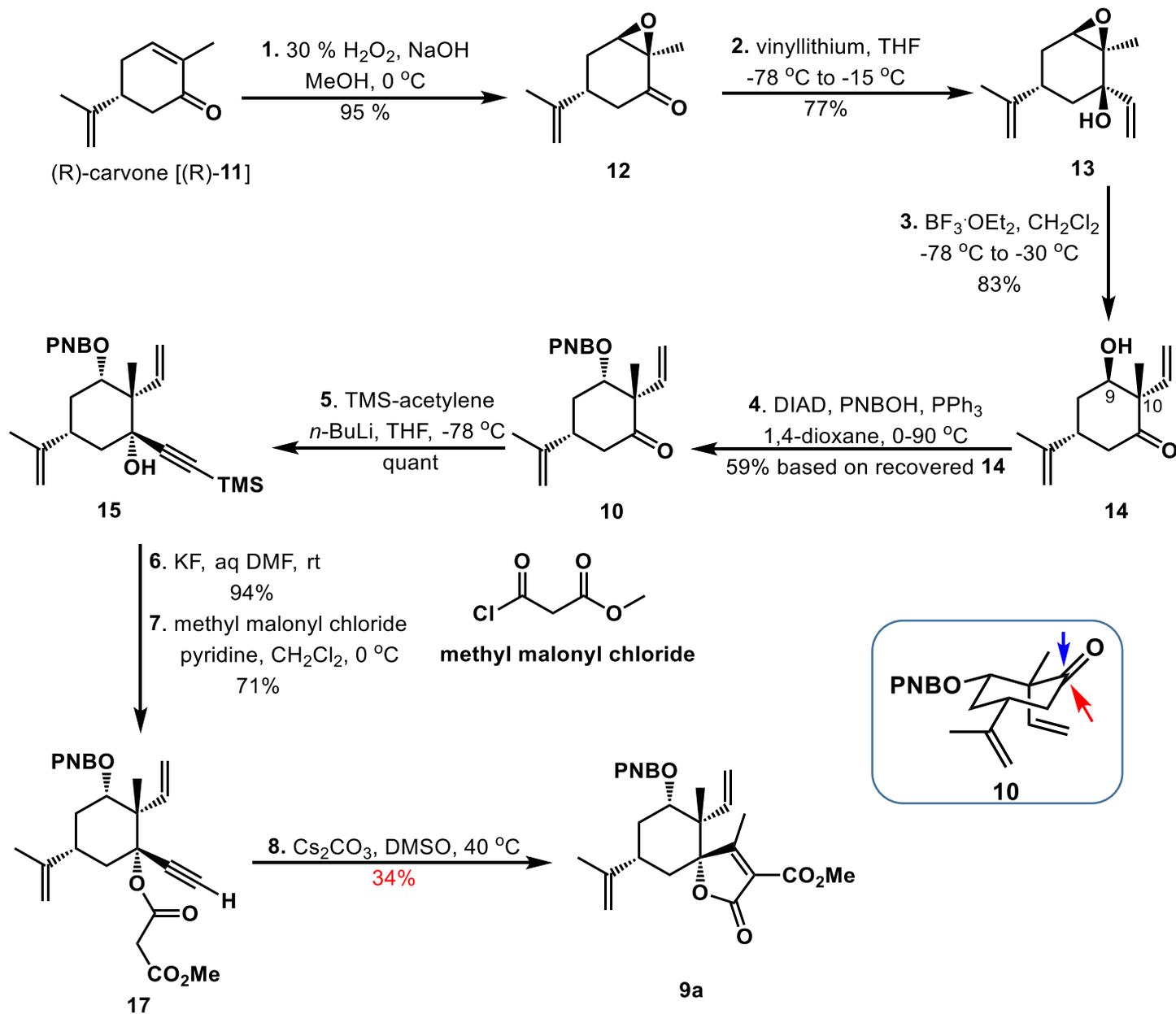


3.  $R^1=Ac$ ,  $R^2=H$  (celafolin B-1)
4.  $R^1=H$ ,  $R^2=Ac$  (celafolin B-2)
5.  $R^1=Ac$ ,  $R^2=Bz$  (celafolin B-3)
6.  $R^1=R^2=Ac$

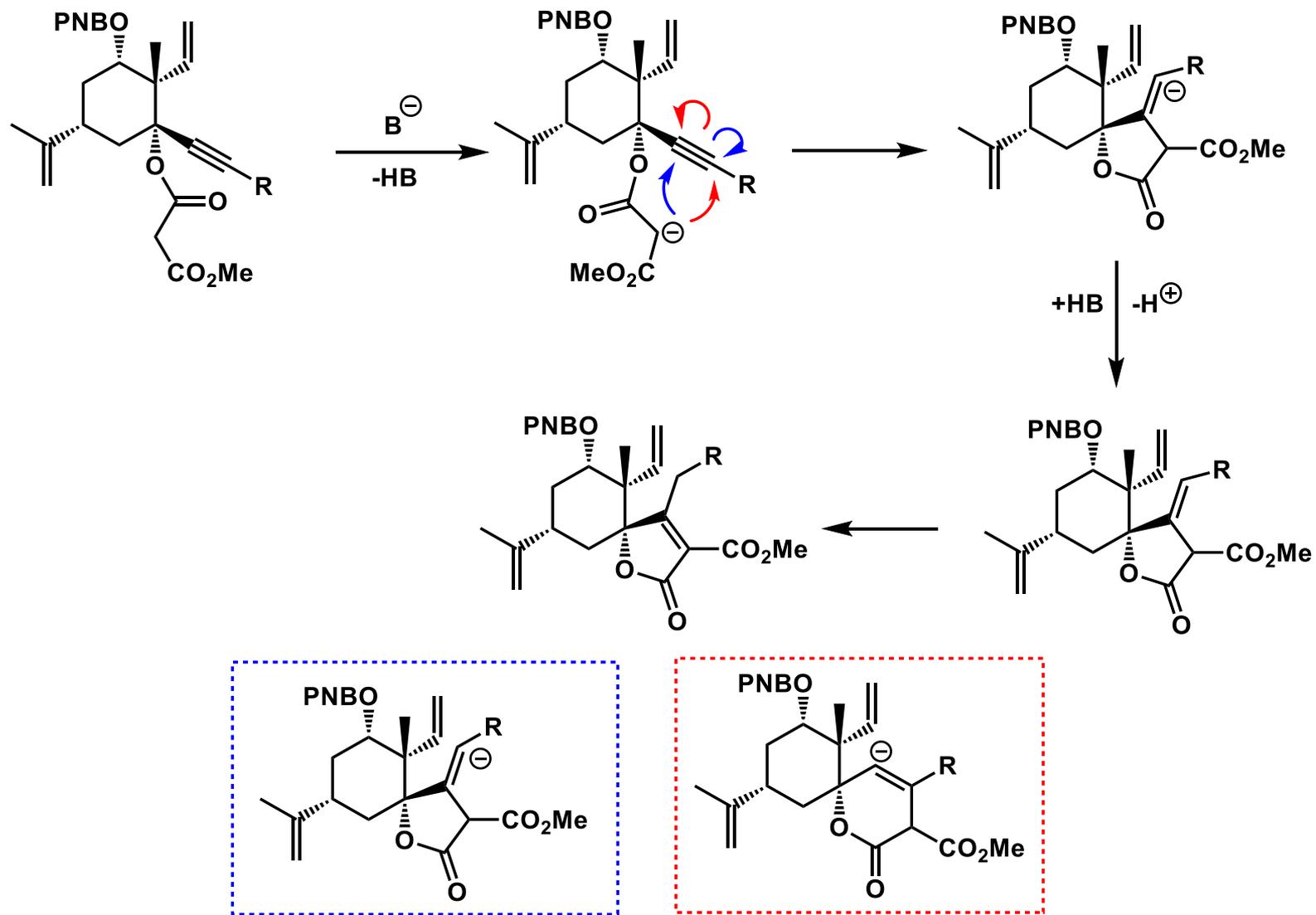
# 逆合成分析



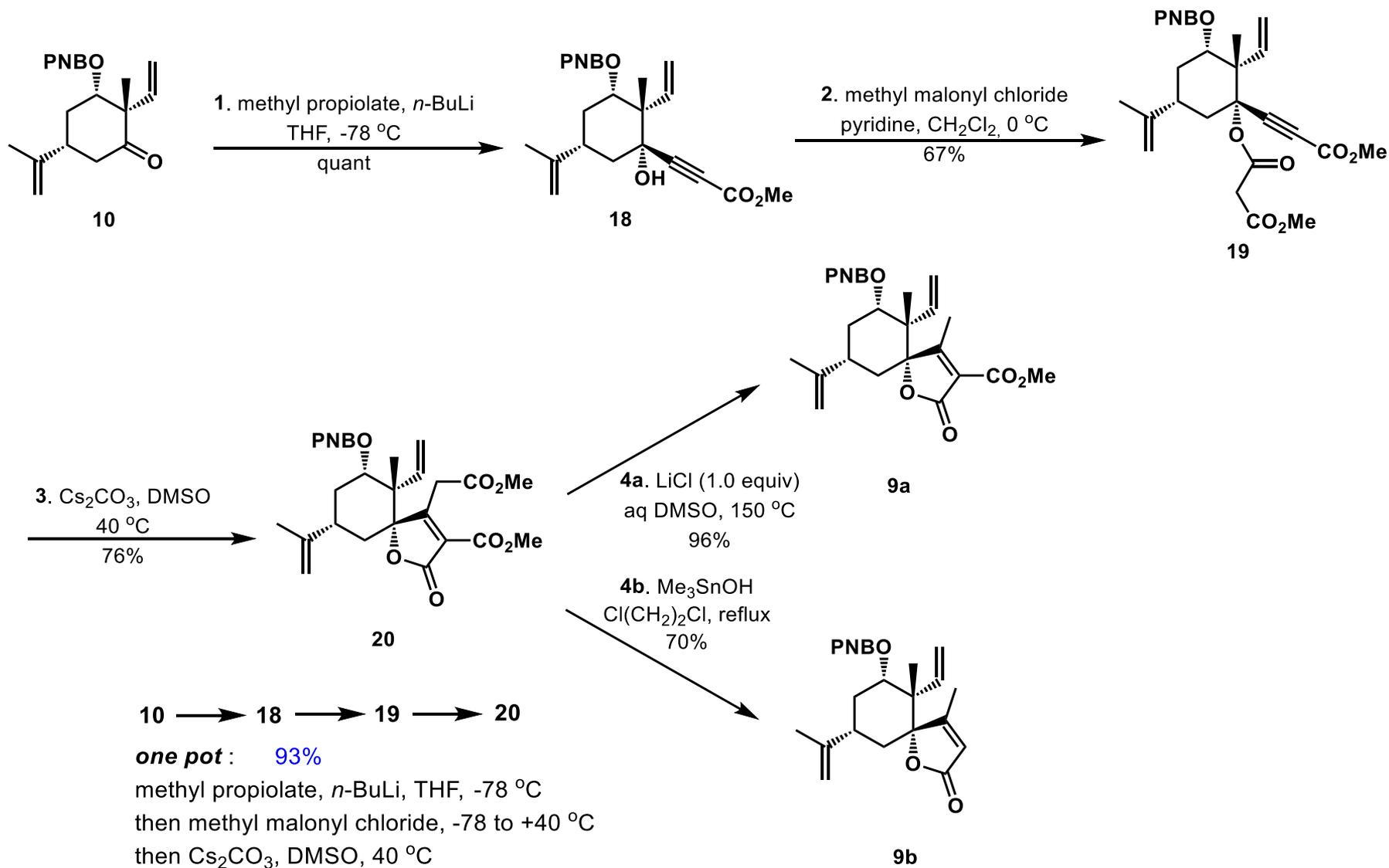
# 螺环中间体9的制备



# 螺环关环机理



# 螺环中间体9的制备



# Krapcho dealkoxycarbonylation

Krapcho dealkoxycarbonylation:

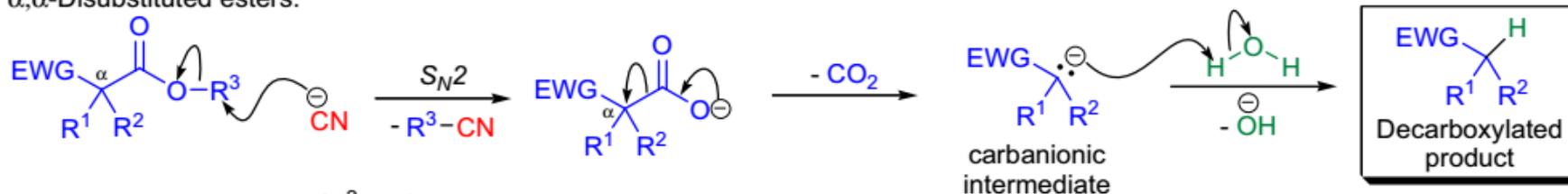


EWG = CO<sub>2</sub>-alkyl, CO<sub>2</sub>-aryl, CN, CO-alkyl, SO<sub>2</sub>-alkyl, SO<sub>2</sub>-aryl; R<sup>1-2</sup> = H, alkyl, aryl; R<sup>3</sup> = Me, Et; MX = NaCN, KCN, LiCl, NaCl, NaBr, NaI, LiI·H<sub>2</sub>O, Na<sub>2</sub>CO<sub>3</sub>·H<sub>2</sub>O, Na<sub>3</sub>PO<sub>4</sub>·12H<sub>2</sub>O, Me<sub>4</sub>NOAc; solvent: DMSO, DMF, DMA, HMPT

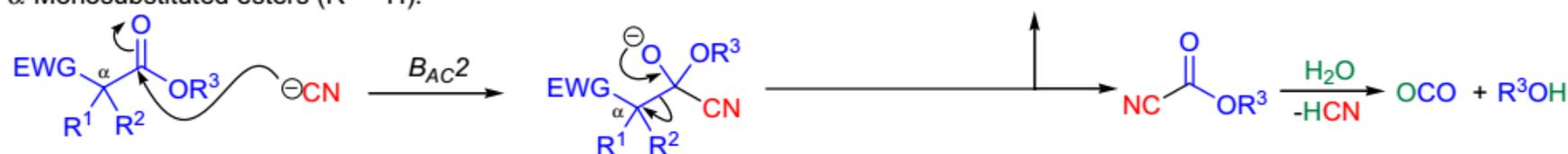
**Mechanism:** 16,17,9,18,19

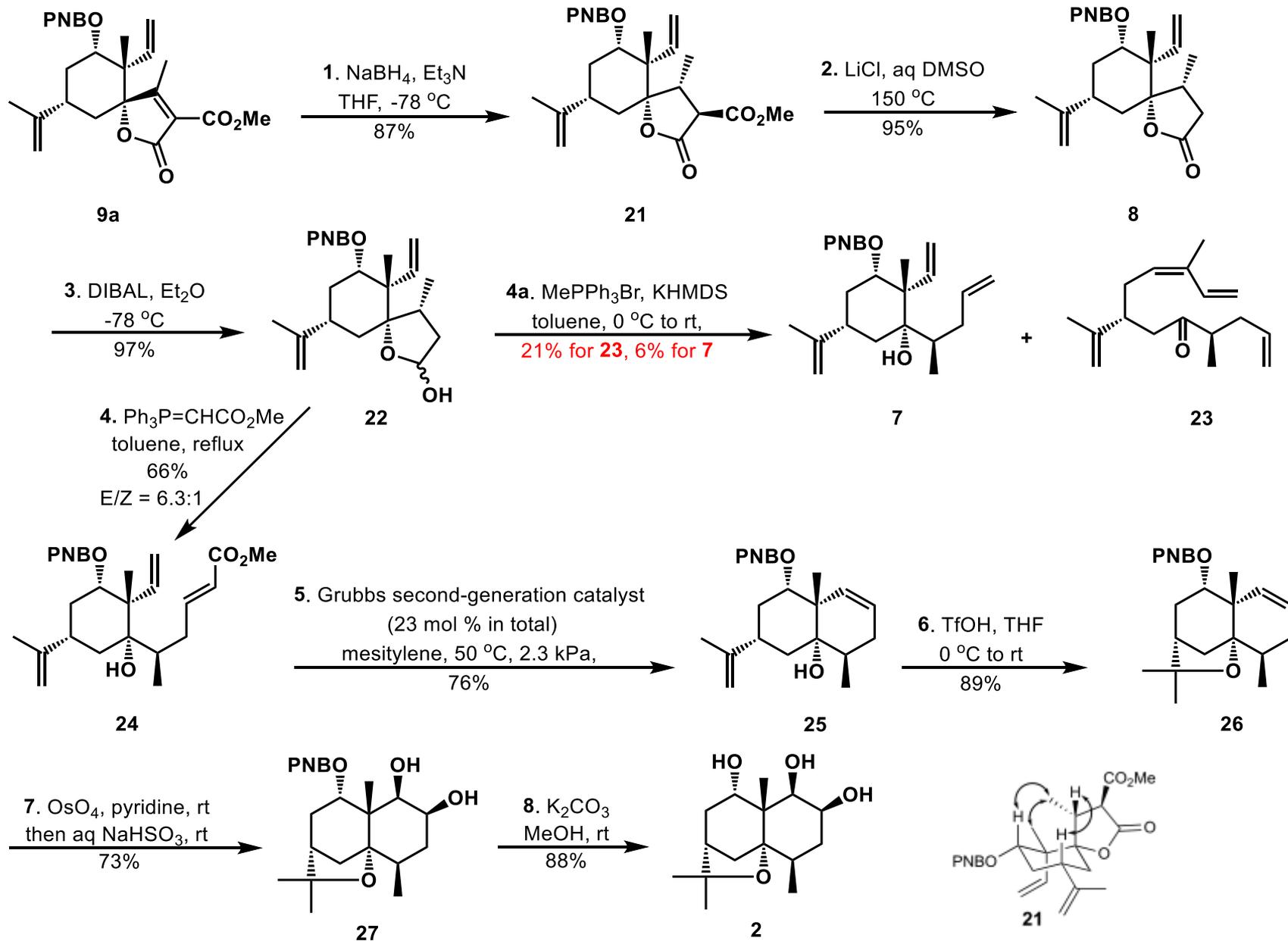
The mechanism of the *Krapcho dealkoxycarbonylation* is dependent on the structure of the substrate ester and the type of anion used. In the case of  $\alpha,\alpha$ -disubstituted diesters (especially the methyl esters), the anion from the salt (cyanide ion in the scheme) attacks the alkyl group of the ester in an S<sub>N</sub>2 fashion and the decarboxylation results in the formation of a carbanionic intermediate that is quenched by the water. In the case of  $\alpha$ -monosubstituted diesters the cyanide attacks the carbonyl group to form a tetrahedral intermediate, which breaks down to give the same carbanionic intermediate and a cyanofornate, which is hydrolyzed to give carbon dioxide and an alcohol.

$\alpha,\alpha$ -Disubstituted esters:

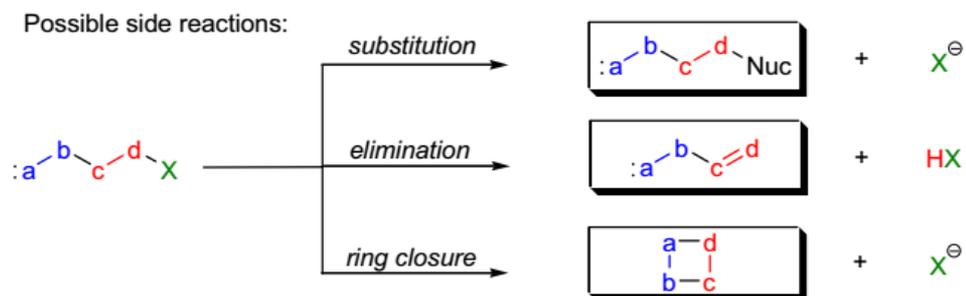
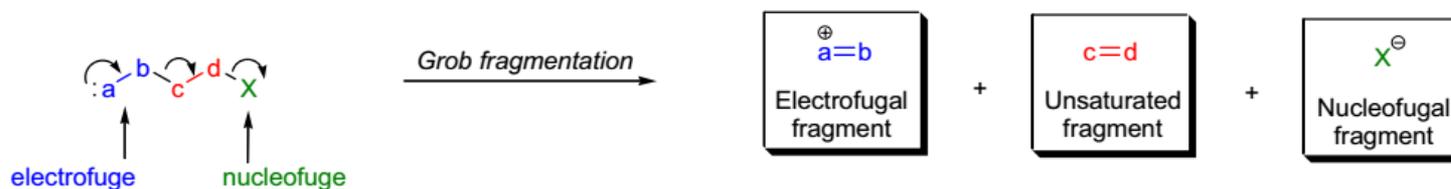


$\alpha$ -Monosubstituted esters (R<sup>2</sup> = H):





# Grob碎片化反应



- $\text{:a-b-c-d-X} \longrightarrow \text{a}=\text{b}^{\oplus} + \text{c}=\text{d} + \text{X}^{\ominus}$ 

one-step synchronous
- $\text{:a-b-c-d-X} \xrightarrow{-\text{X}^{\ominus}} \left[ \text{:a-b-c-d}^{\oplus} \right] \longrightarrow \text{a}=\text{b}^{\oplus} + \text{c}=\text{d}$ 

two-step cationic
- $\text{:a-b-c-d-X} \xrightarrow{-\text{a}=\text{b}^{\oplus}} \left[ \text{:c-d-X}^{\ominus} \right] \longrightarrow \text{c}=\text{d} + \text{X}^{\ominus}$ 

two-step anionic

