

Total Synthesis of (+)-Cyclobutastelletolide B

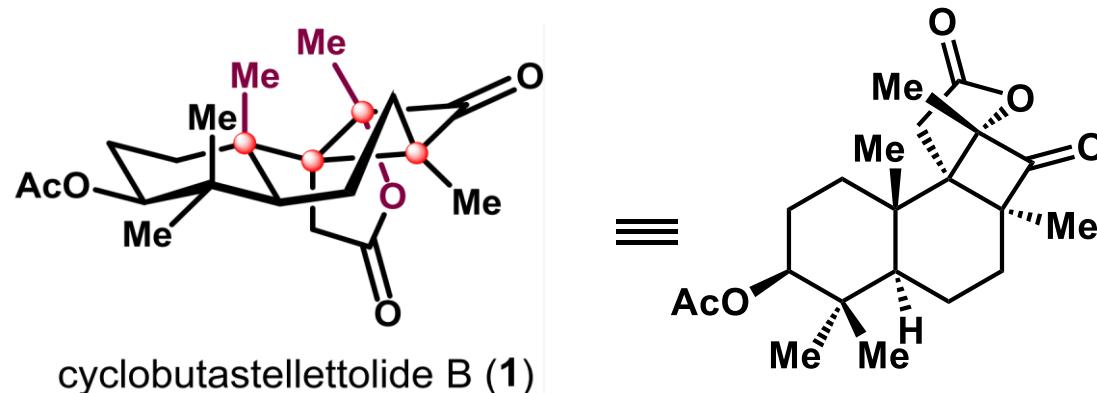
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Cite This: <https://doi.org/10.1021/jacs.1c08880>



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isolated from a *Stelletta* sp. By Stonik et al. in 2019
an unusual 6/6/4-fused tricyclic core
six stereocenters (three contiguous quaternary stereocenters)

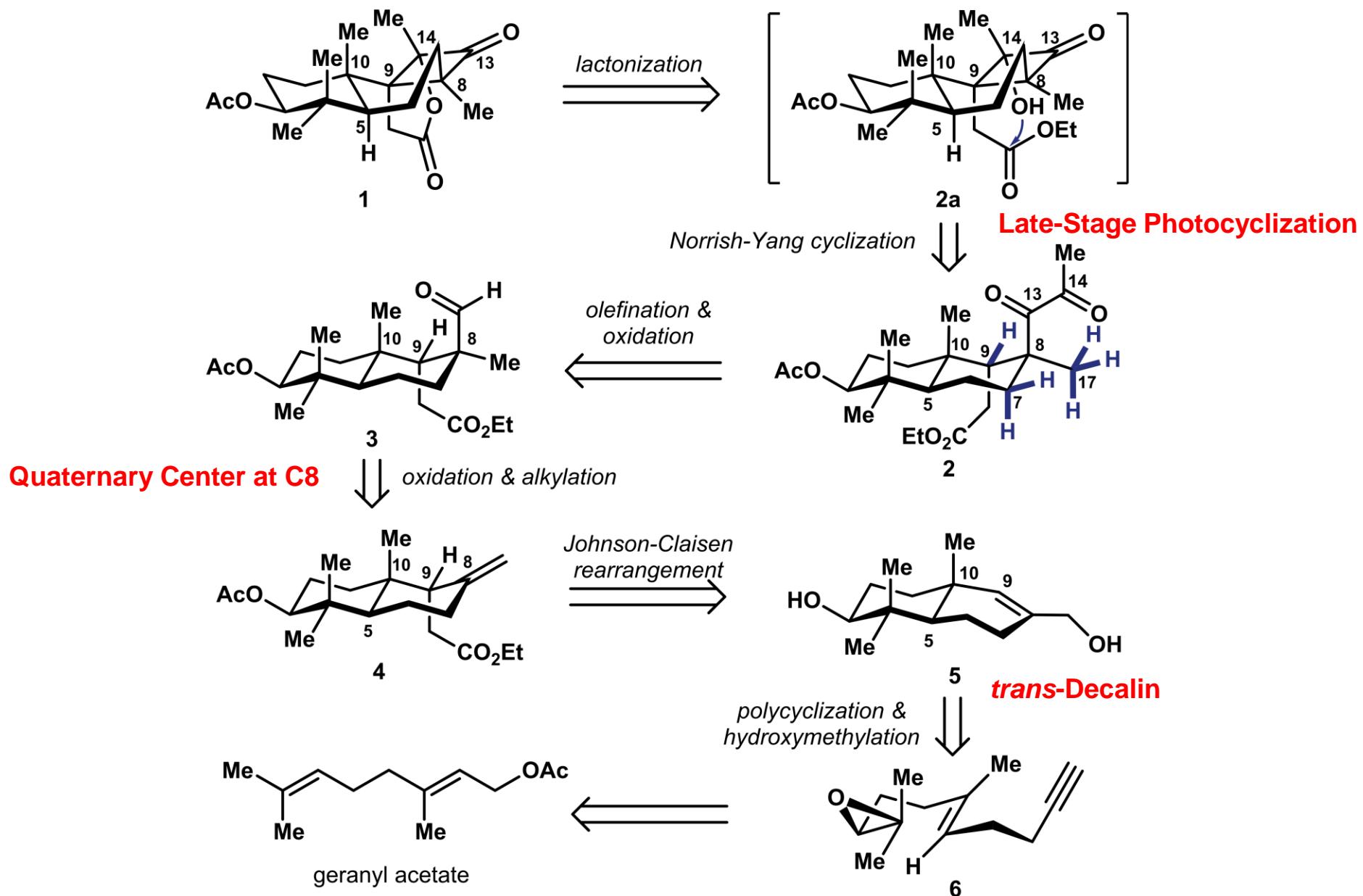
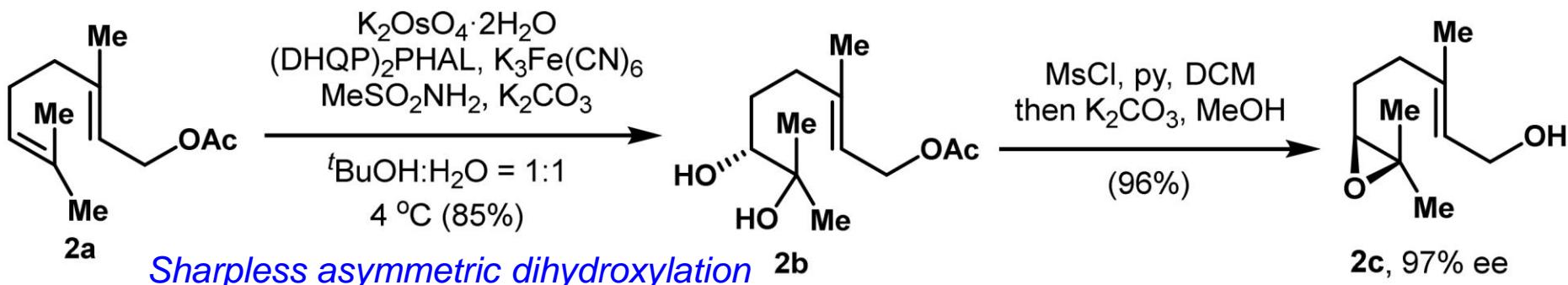
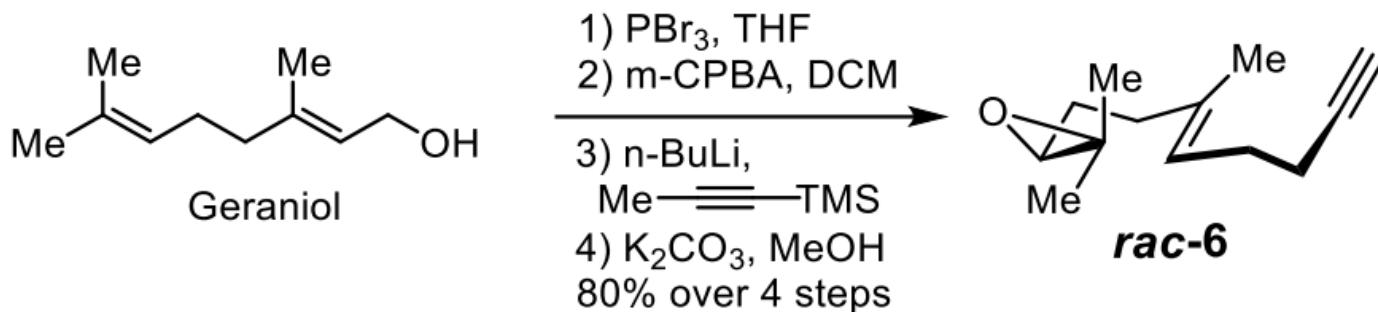
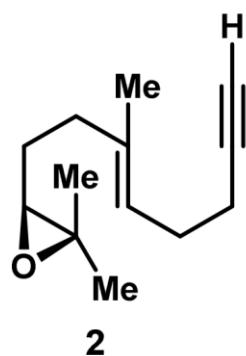


Figure 1. Retrosynthetic analysis of cyclobutastellettolide B (1).

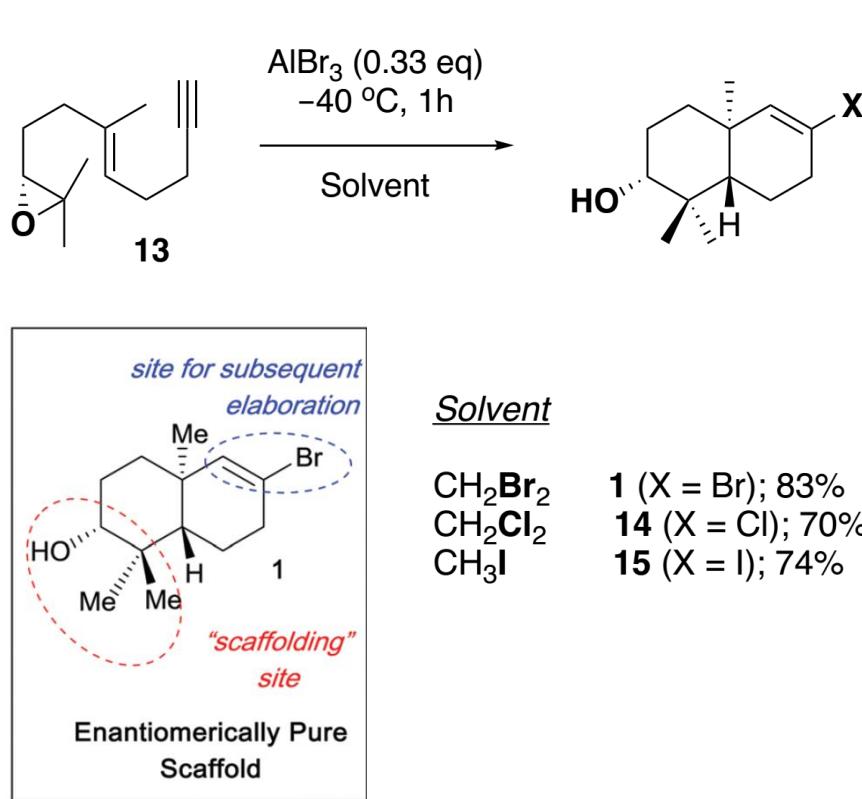
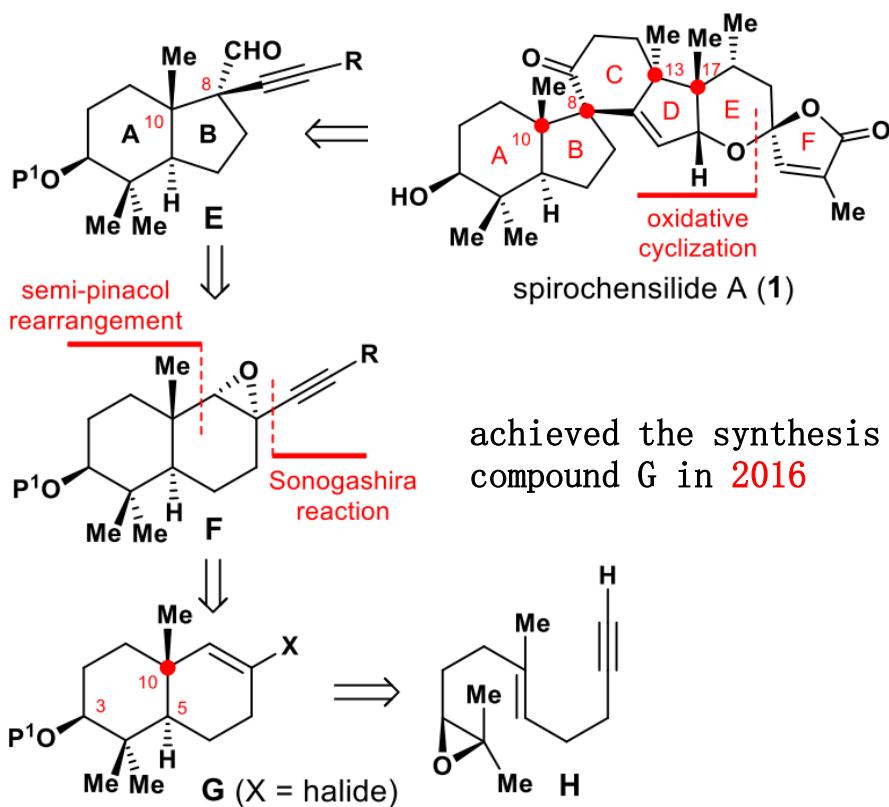
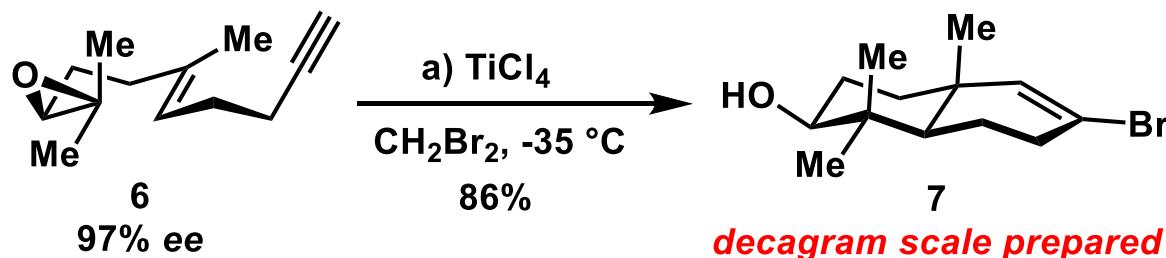
Synthetic route of the known epoxide **2** in asymmetric form:



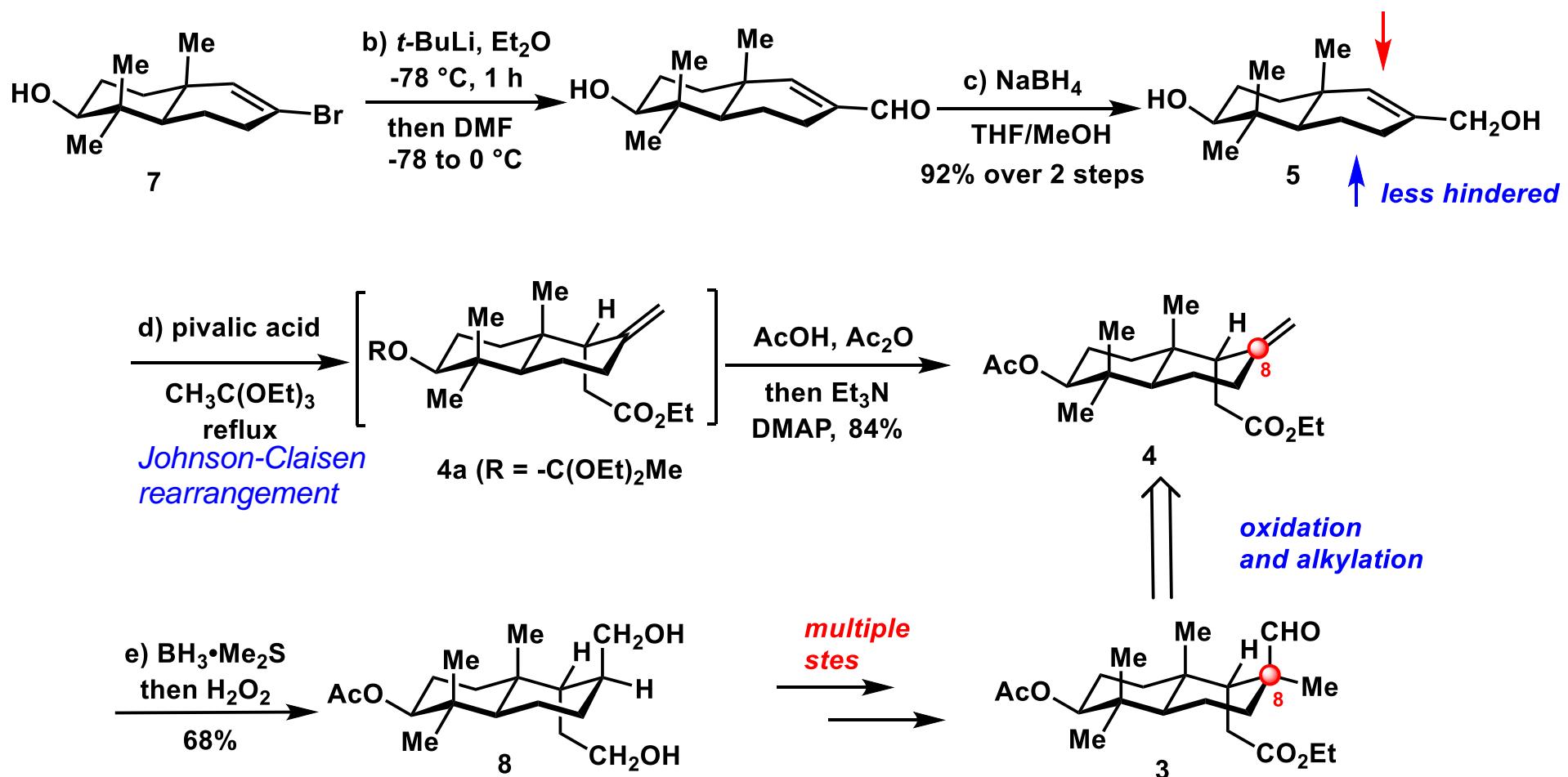
- 1) MsCl , Et_3N , LiCl
THF, -40°C to rt
- 2) $\text{LiCH}_2\text{C}\equiv\text{CTMS}$
THF, 0°C
- 3) K_2CO_3 , MeOH
(76%, 3 steps)

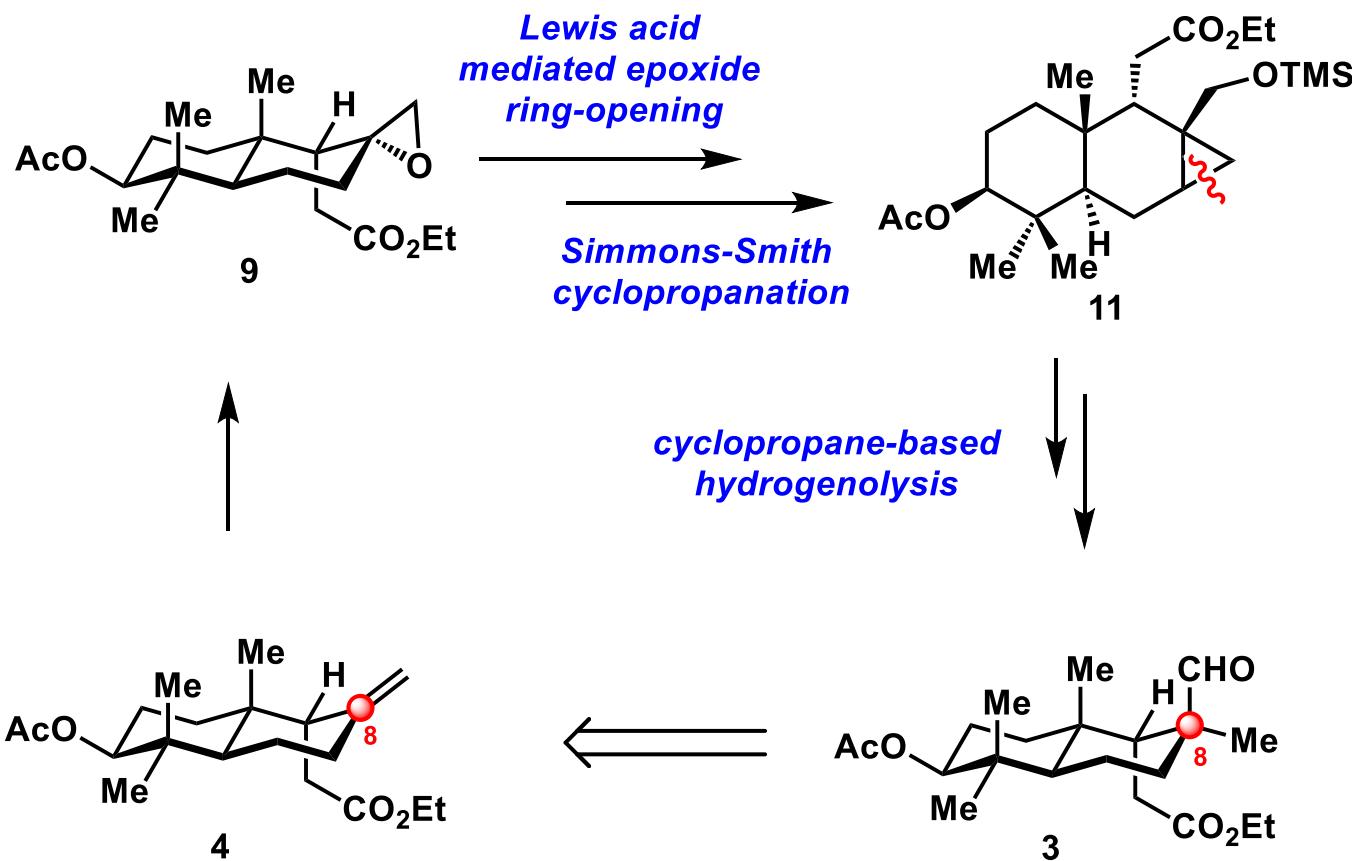


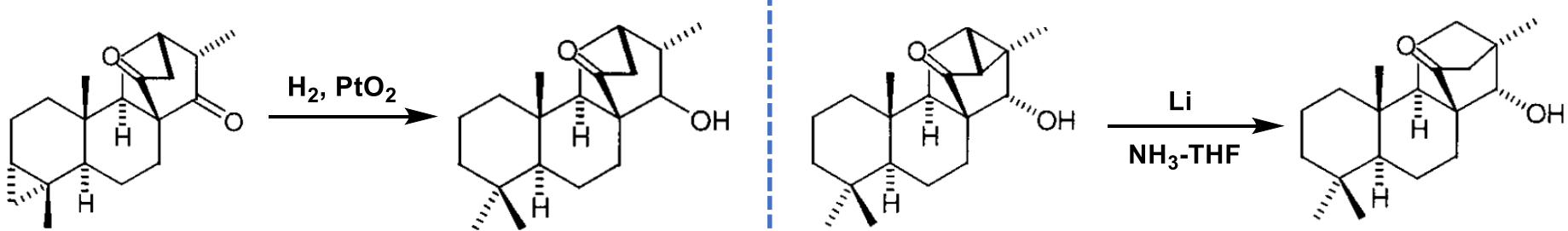
Synthesis of *trans*-Decalin



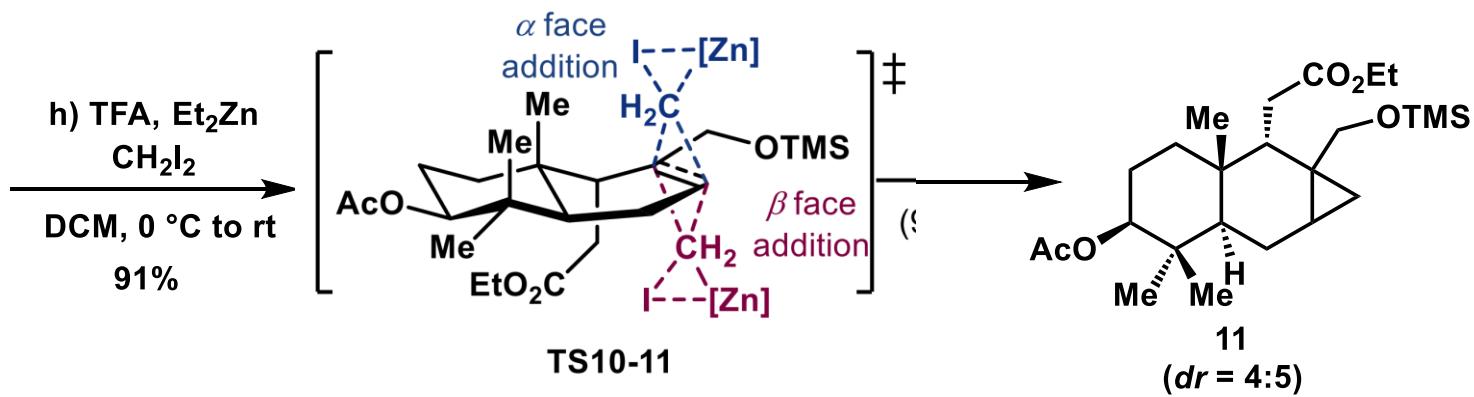
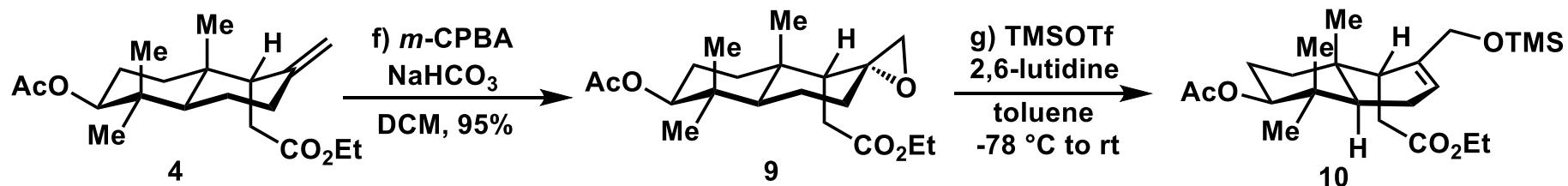
Efforts to Install a Quaternary Center at C8

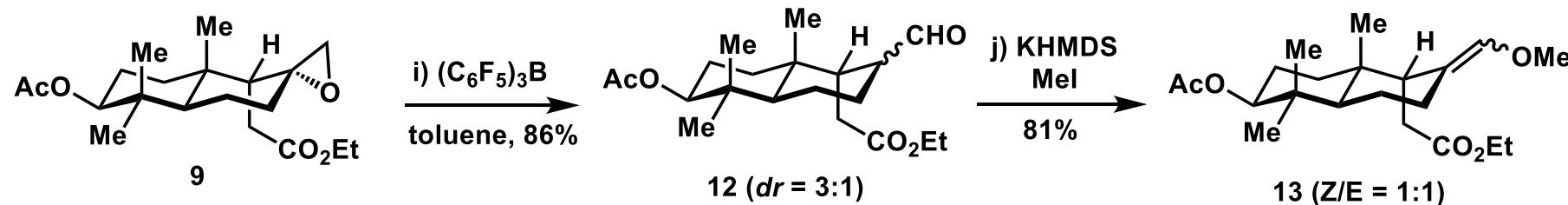






Synlett 2001, 349





12 was reacted with a variety of methylating agents under different basic conditions

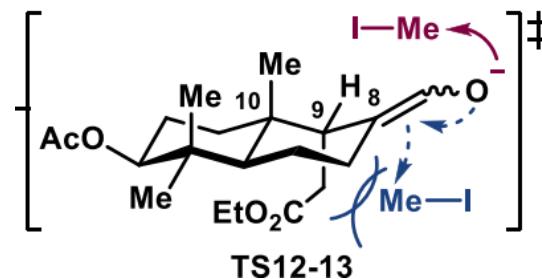
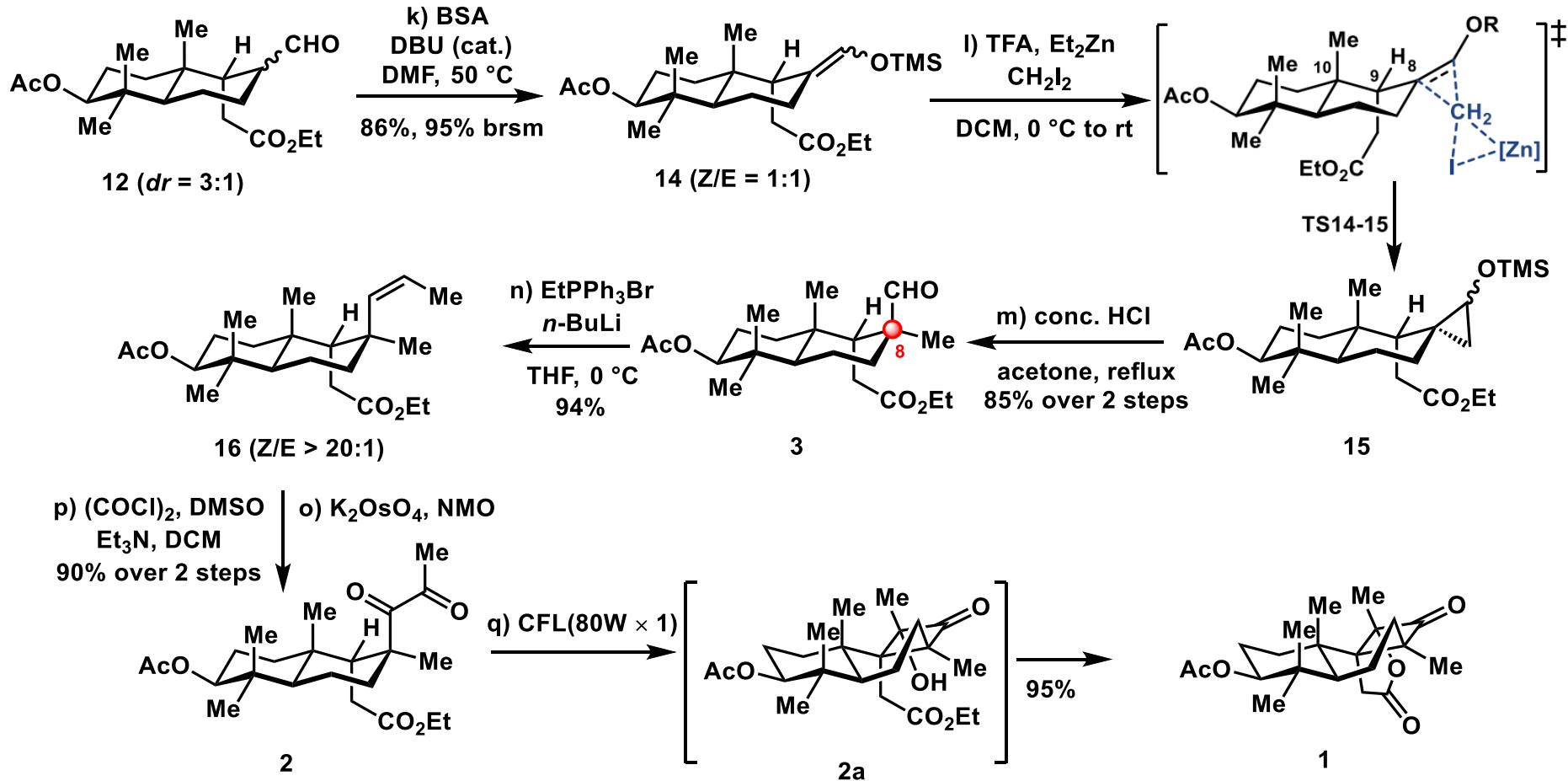
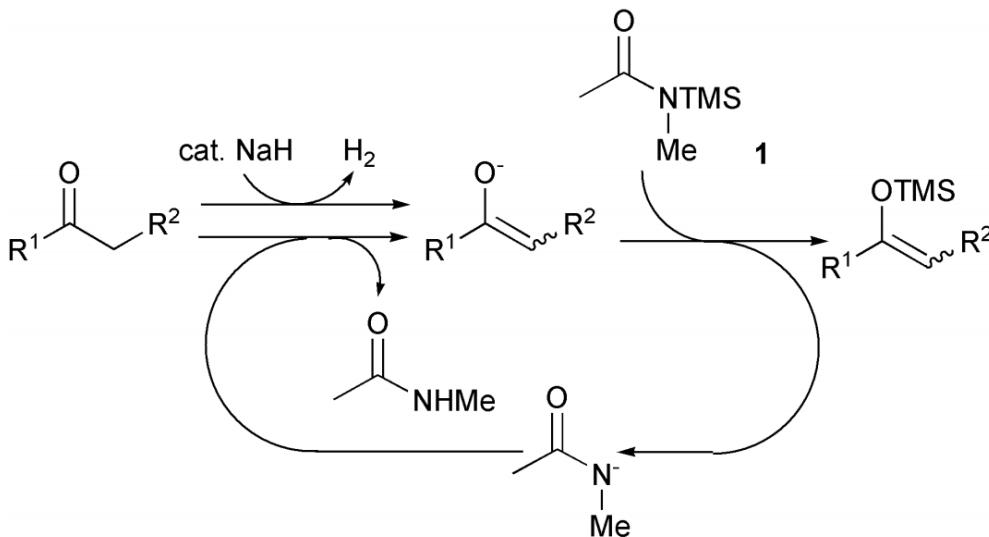


Table S1. Conditions screening of semi-pinacol rearrangement.

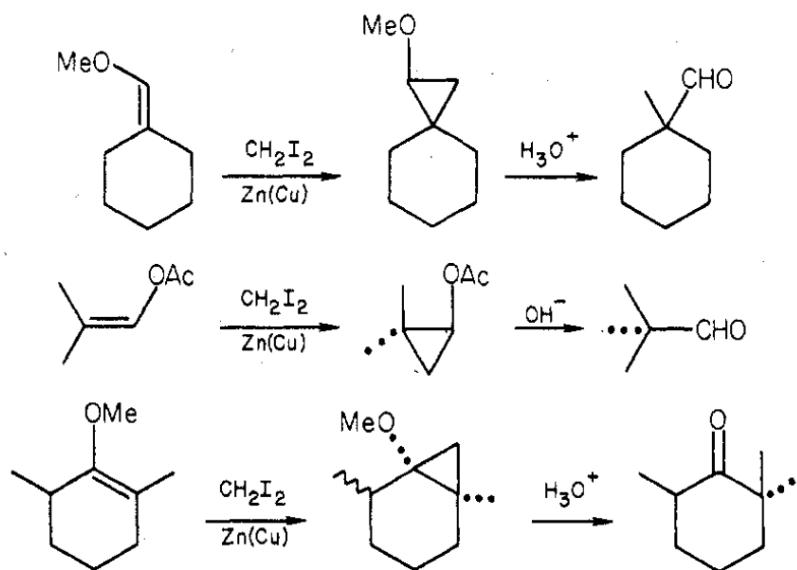
Entry	LA	equiv.	sol.	Temp	Time	yield
1	$BF_3 \cdot OEt_2$	2.0 eq.	DCM	0 °C to r.t.	0.5 h	48 %
2	$BF_3 \cdot OEt_2$	2.0 eq.	toluene	0 °C to r.t.	0.5 h	61 %
3	$BF_3 \cdot OEt_2$	6.0 eq.	Et_2O	0 °C to r.t.	3.0 h	64 %
4	$(C_6F_5)_3B$	0.5 eq.	DCM	r.t.	4.0 h	71 %
5	$(C_6F_5)_3B$	0.5 eq.	toluene	r.t.	4.0 h	92 %

Cyclopropanation/oxycyclopropane ring-opening reaction strategy





Chem. Commun. **2002**, 1628.

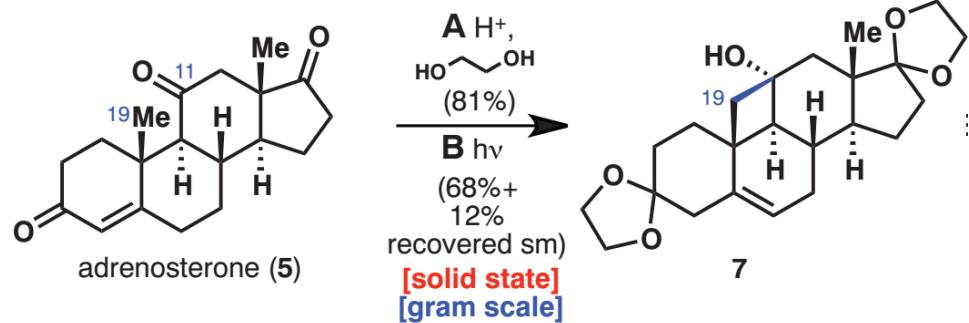
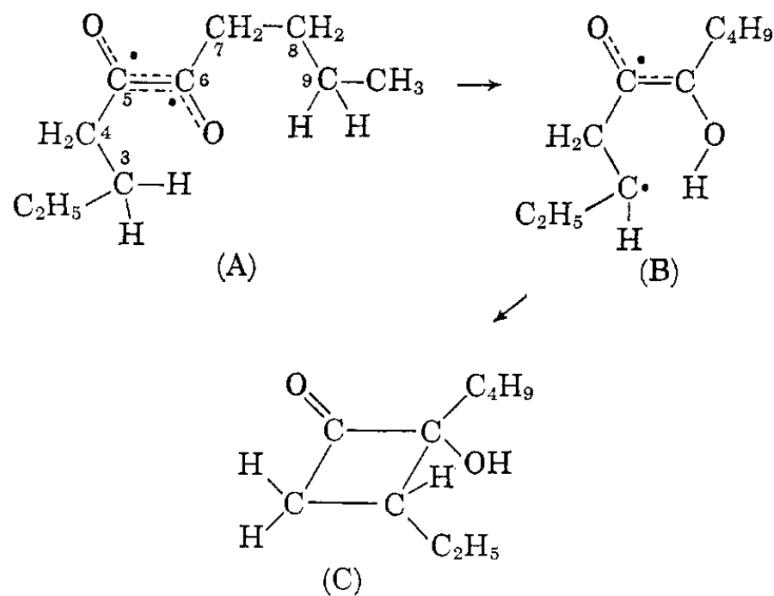


masked keto systems, the combination of oxycyclopropane synthesis and acid-induced unravelling represents an interesting three-step equivalent of the **α -alkylation of carbonyl compounds** (*vide infra*), especially of relevance to the formation of α -keto quaternary centers and to **angular methylation** of terpenes and steroids.



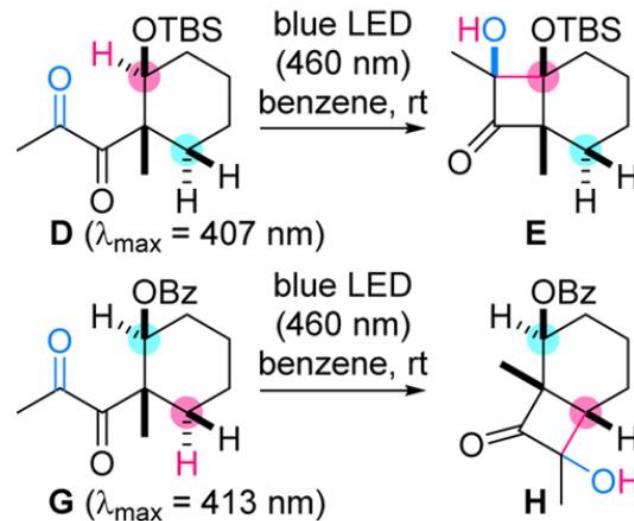
Acc. Chem. Res. **1980**, 13, 27.

Norrish-Yang photocyclization

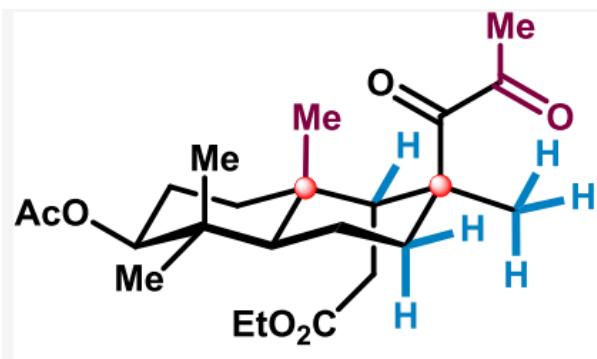


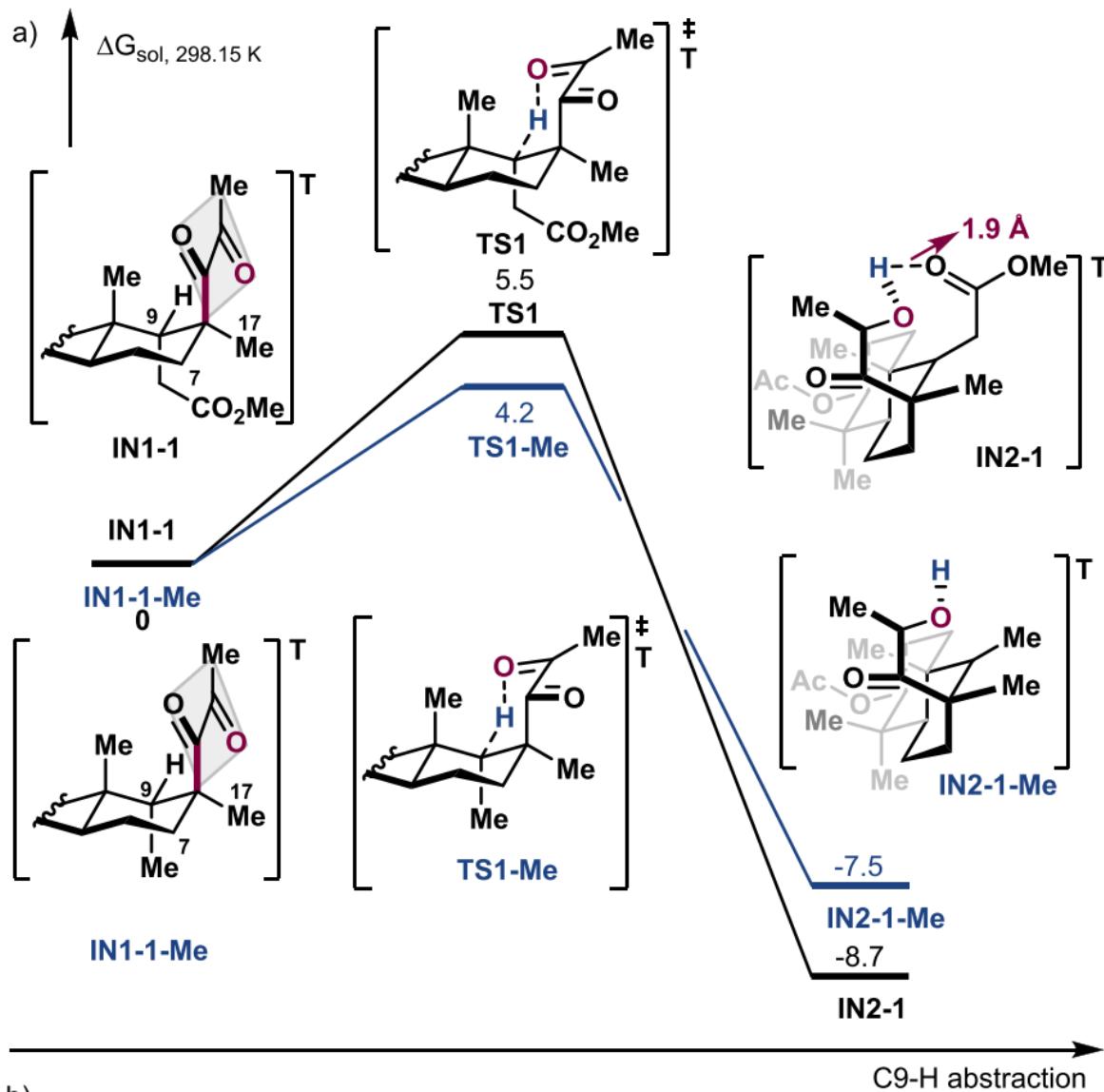
Science 2013, 339, 59.

J. Am. Chem. Soc. 1962, 84, 118.



J. Am. Chem. Soc. 2017, 139, 1814.





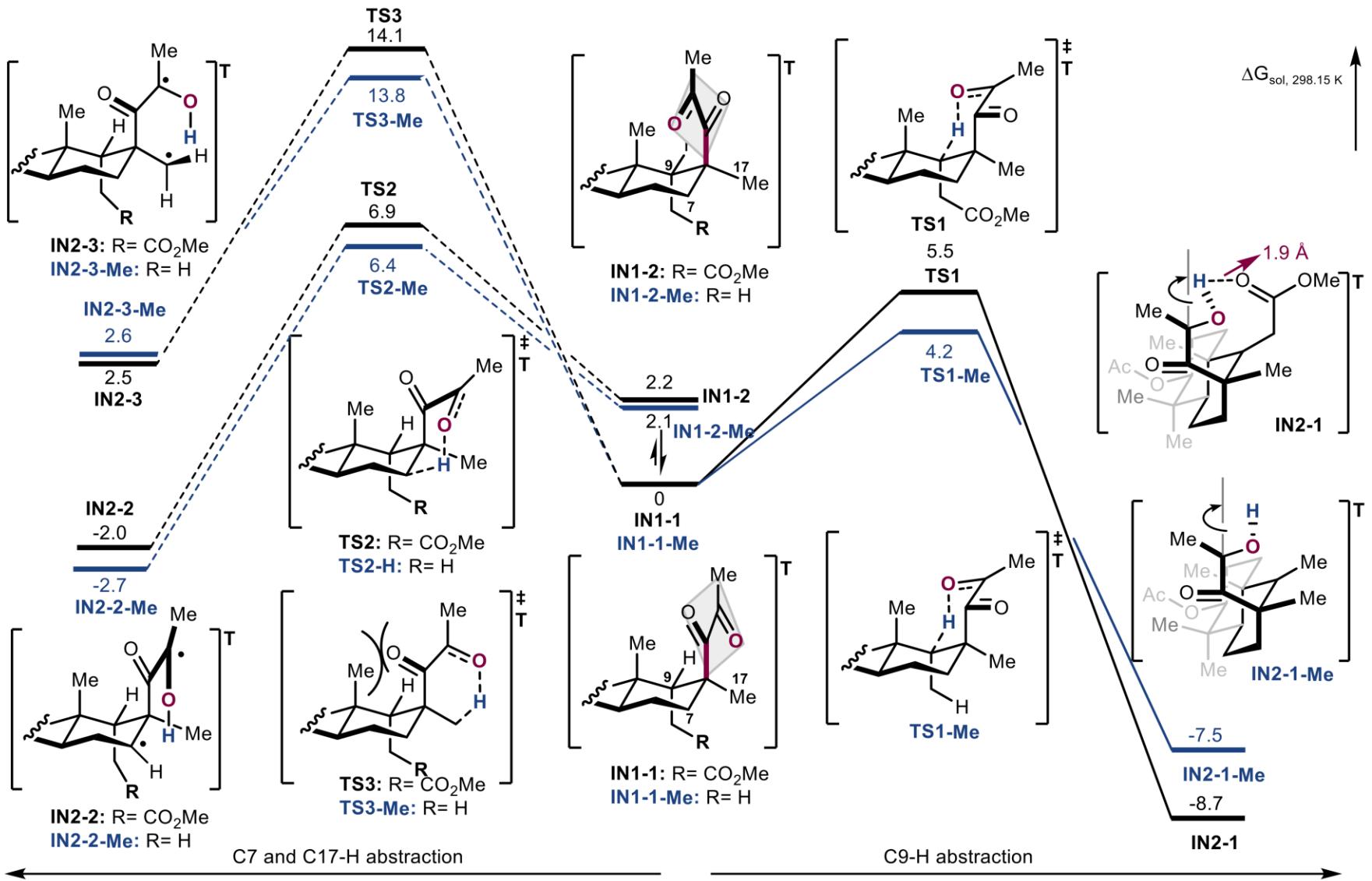
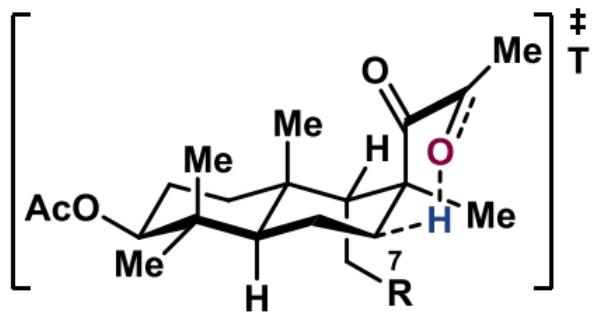


Figure S2. Energy profiles of 1,5-hydrogen shift of **IN1-1** and **IN1-1-Me** via C9-H, C7-H and C17-H abstraction.

b)



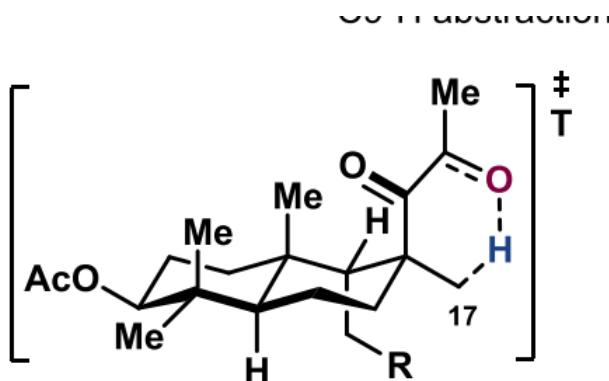
TS2 ($R = CO_2Me$) **TS2-Me** ($R = H$)

$$\Delta G^\ddagger = 6.9$$

$$\Delta G^\circ = -2.0$$

$$\Delta G^\ddagger = 6.4$$

$$\Delta G^\circ = -2.7$$



TS3 ($R = CO_2Me$) **TS3-Me** ($R = H$)

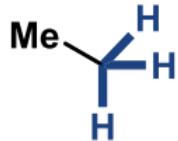
$$\Delta G^\ddagger = 14.1$$

$$\Delta G^\circ = +2.6$$

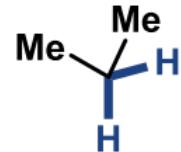
$$\Delta G^\ddagger = 13.8$$

$$\Delta G^\circ = +2.5$$

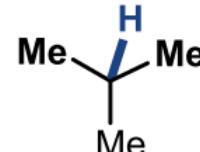
c)



BDE = 100.5 kcal/mol

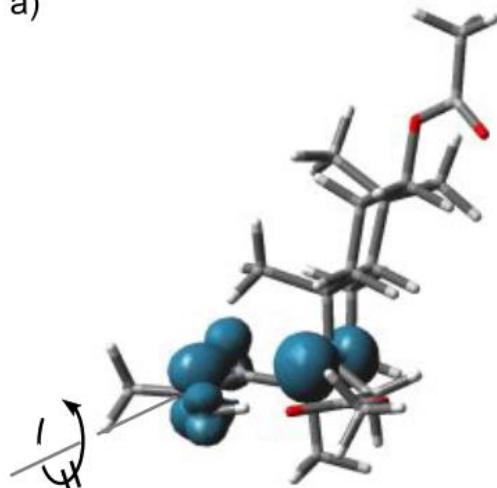
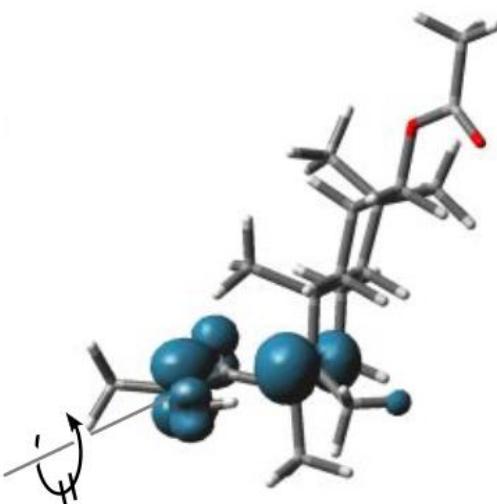


BDE = 98.1 kcal/mol



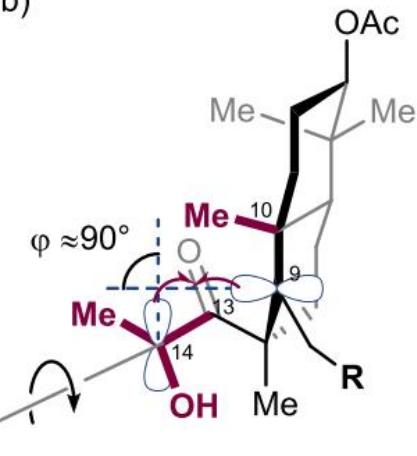
BDE = 95.7 kcal/mol

a)

**IN2-1****IN2-1-Me**

anticlockwise rotation - restricted

b)



IN2-1 ($R = CO_2Me$)
IN2-1-Me ($R = H$)



clockwise rotation - 90° rule

Figure 4. (a) Spin density surfaces of biradical **IN2-1** and **IN2-1-Me** (isovalue = 0.01); (b) schematic representation of the 90° rule.

