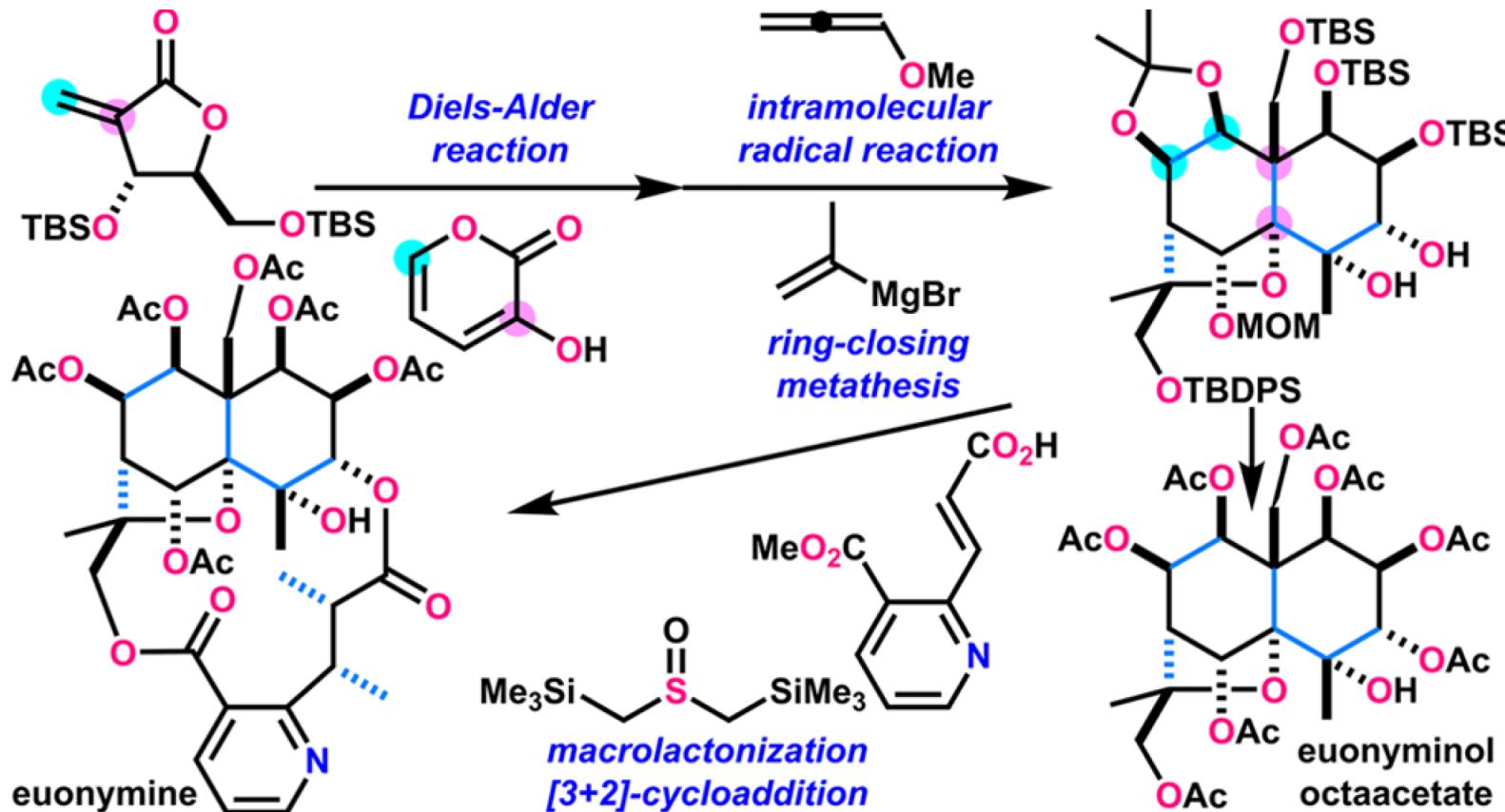


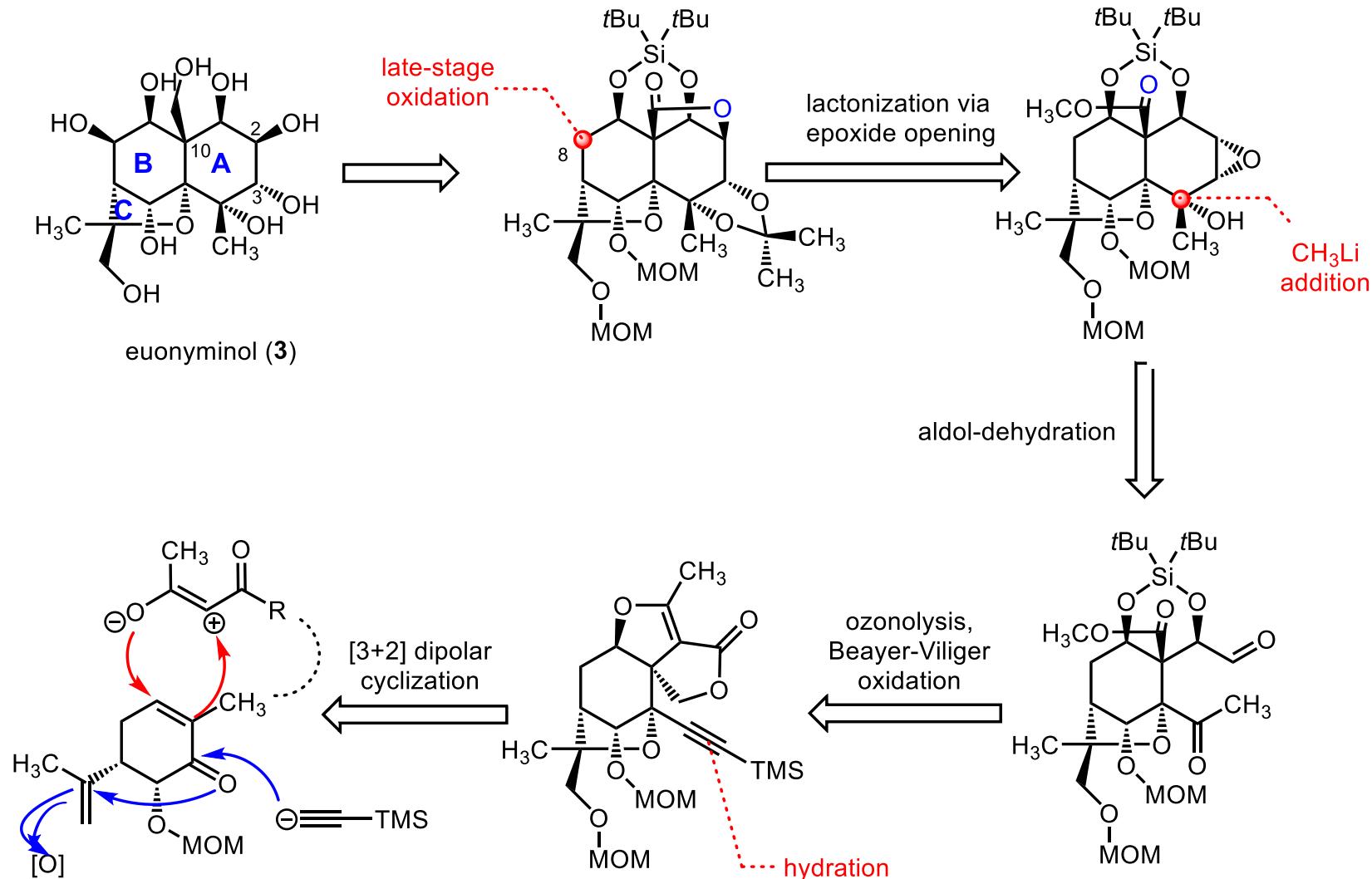
Total Synthesis of Euonymine and Euonyminol Octaacetate

Yinghua Wang,[†] Toshiya Nagai,[†] Itsuki Watanabe, Koichi Hagiwara, and Masayuki Inoue*

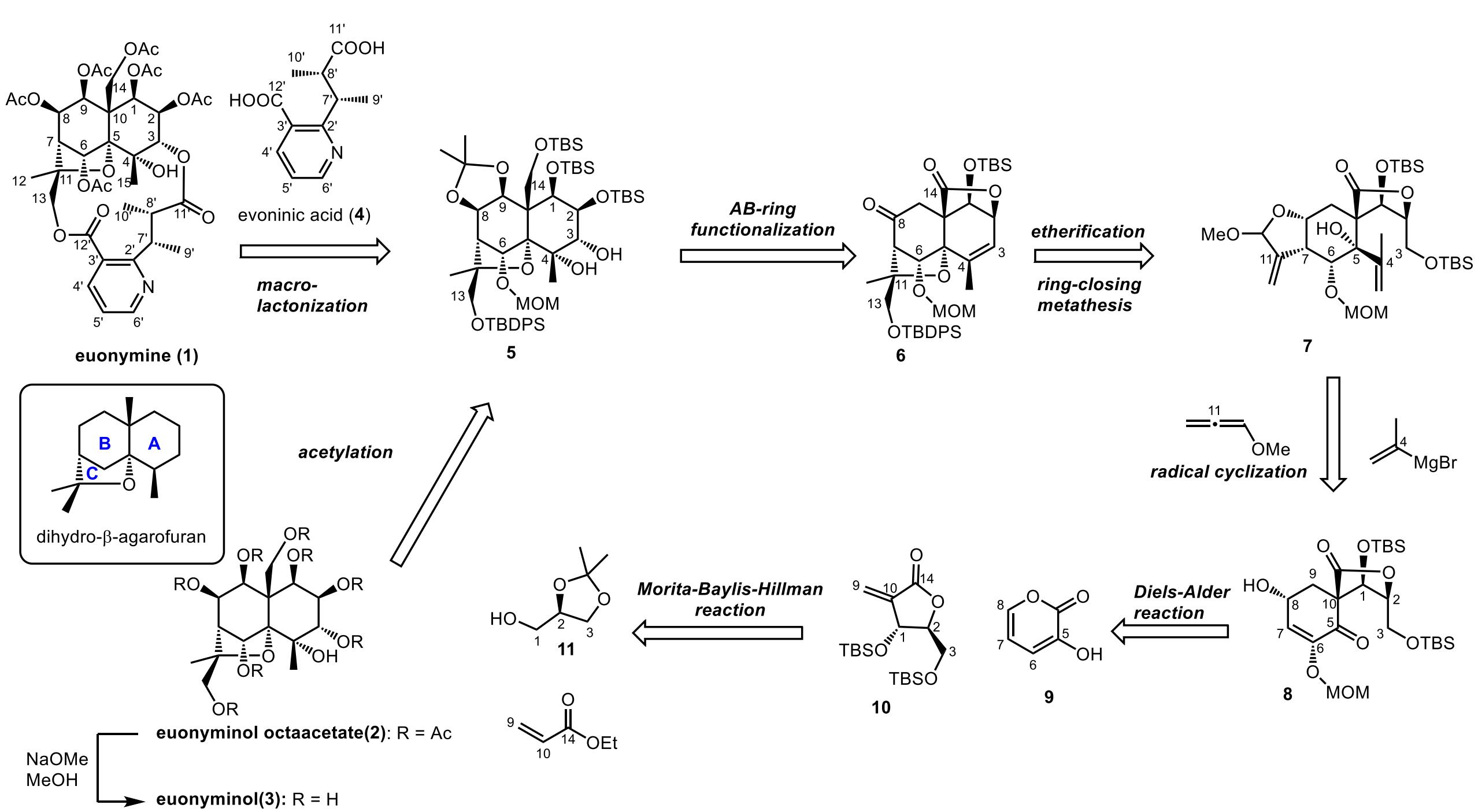


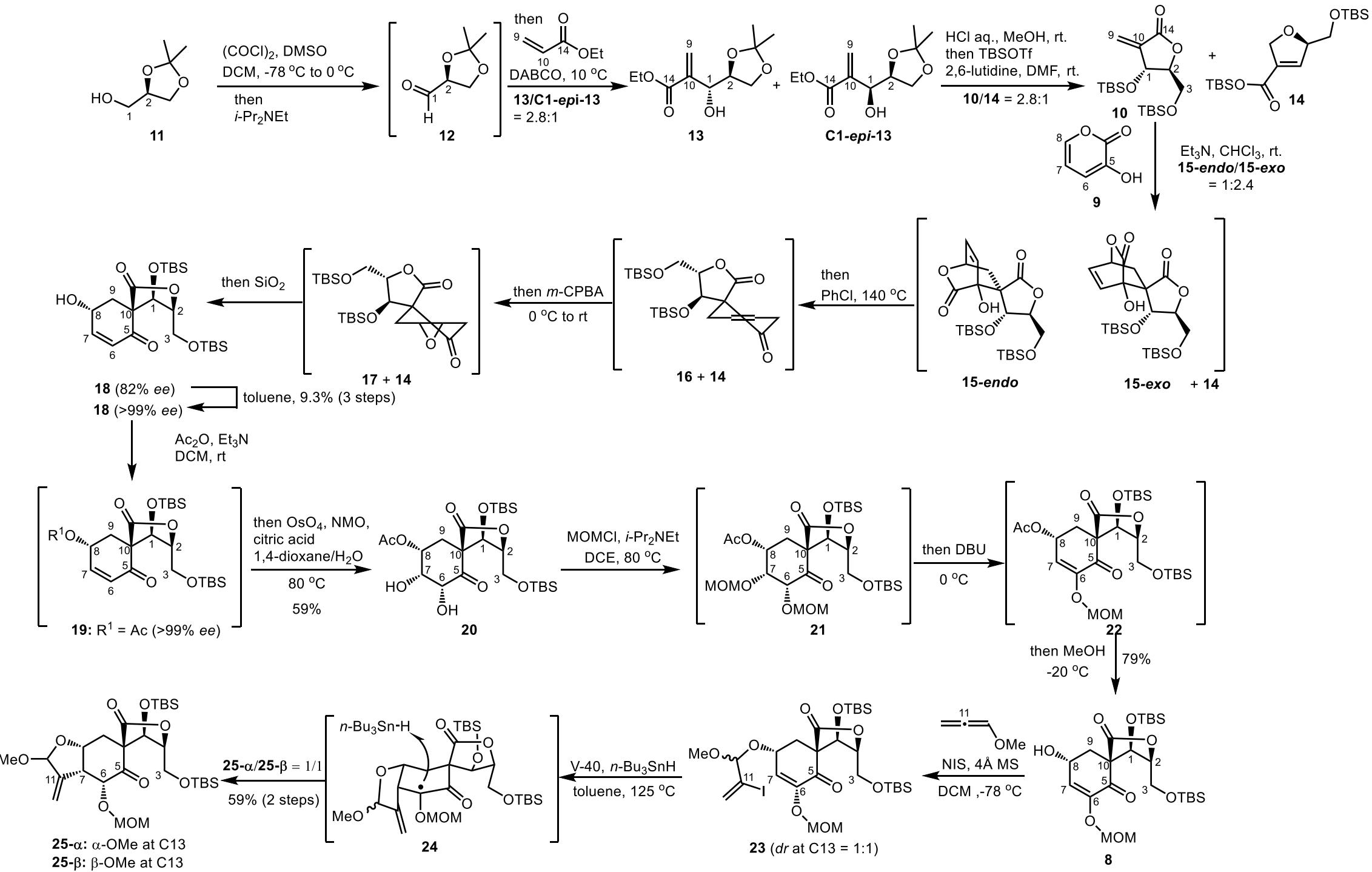
Enantioselective Synthesis of Euonyminol

Martin Tomanik, Zhi Xu, and Seth B. Herzon*



DOI: 10.1021/jacs.0c12998.





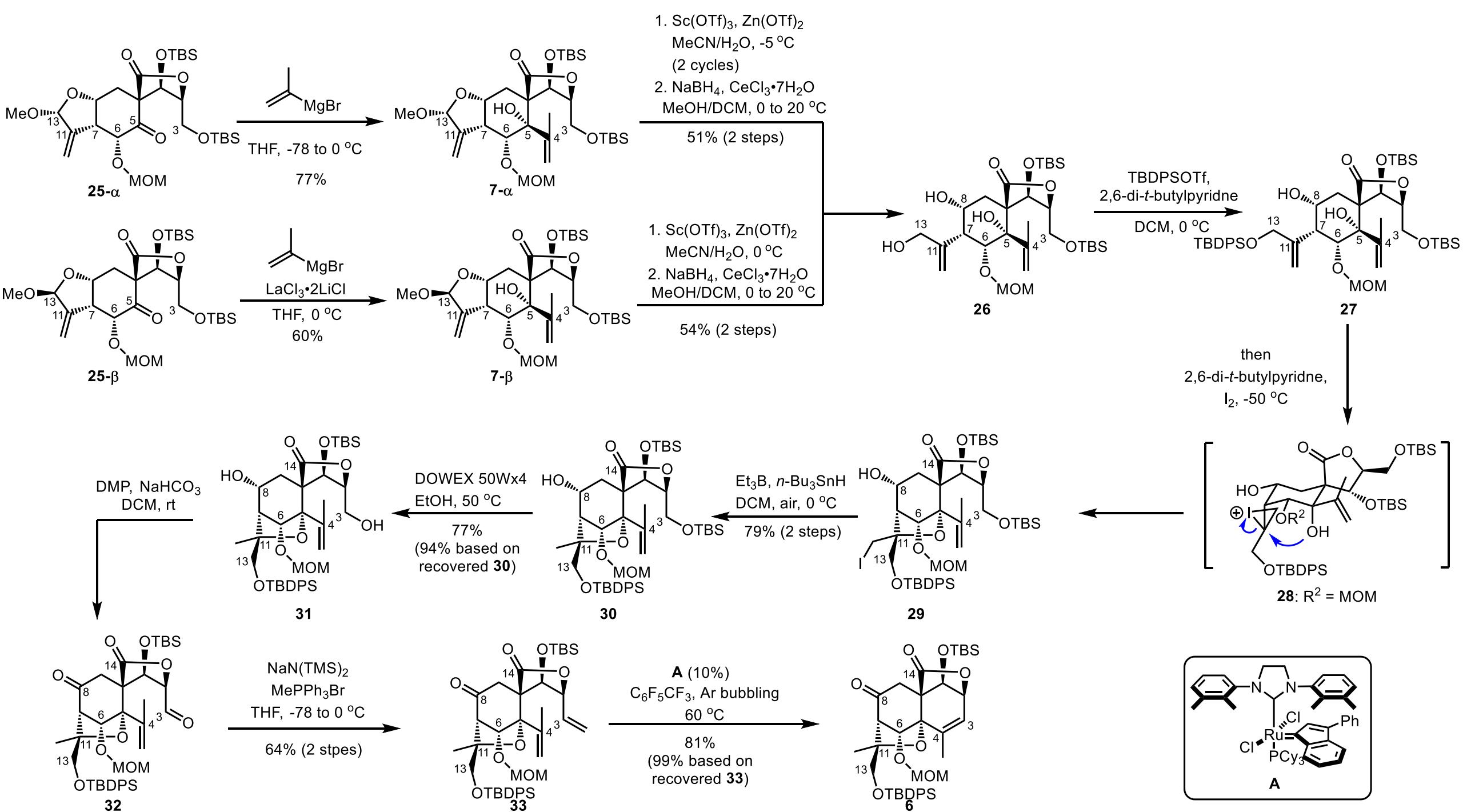
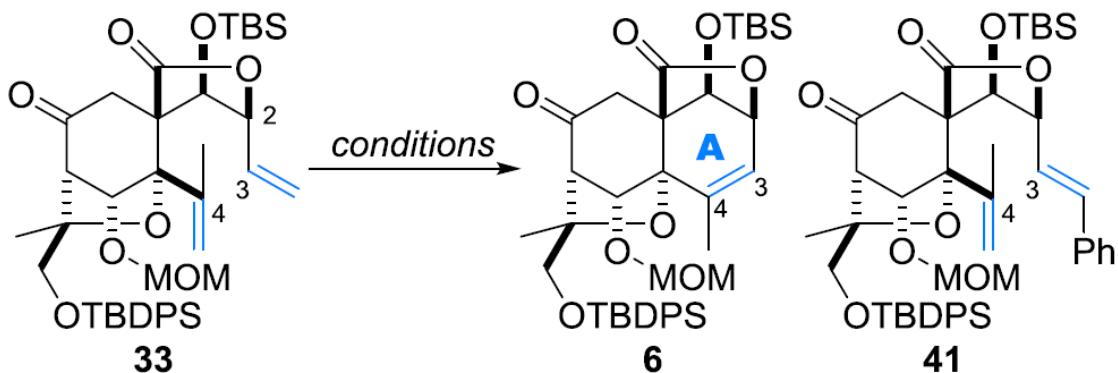
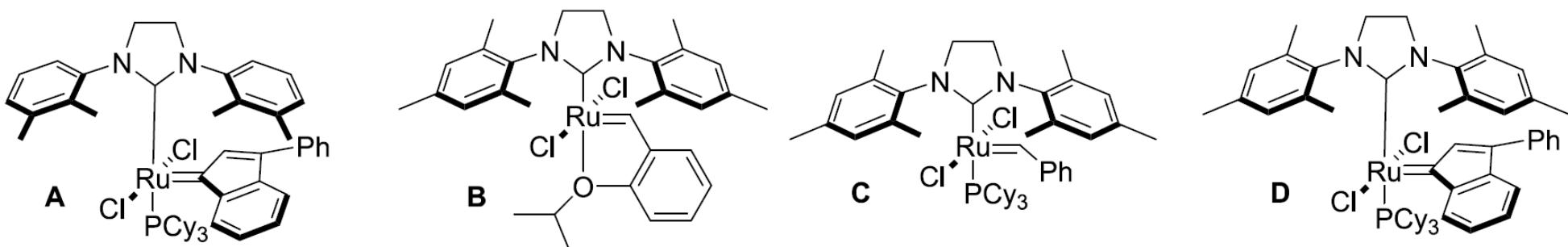


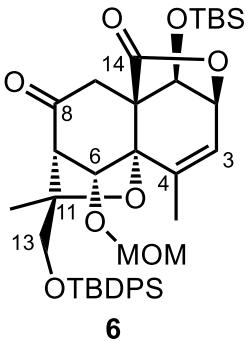
Table 1. Investigation of the A-Ring Formation



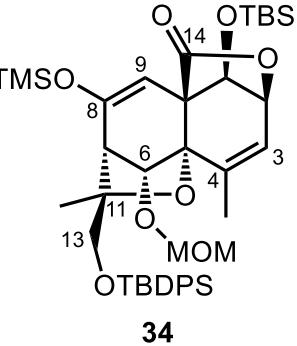
entry	conditions	results
1 ^a	B (100 mol %), toluene, 125 °C	no reaction
2 ^a	C (100 mol %), toluene, 125 °C	33/6/41 = 2:1:2
3 ^a	D (100 mol %), toluene, 125 °C	6 (59%)
4 ^a	D (50 mol %), C ₆ F ₅ CF ₃ , 120 °C	6 (58%)
5 ^a	D (50 mol %), C ₆ F ₅ CF ₃ , 80 °C	6 (83%) 33 (10%)
6	A (10 mol %), C ₆ F ₅ CF ₃ , 60 °C	6 (81%) 33 (18%)

^aDiene **33** was the mixture with C2-*epi*-**33** (**33/C2-*epi*-33** = 10:1).

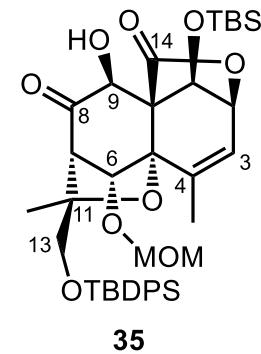




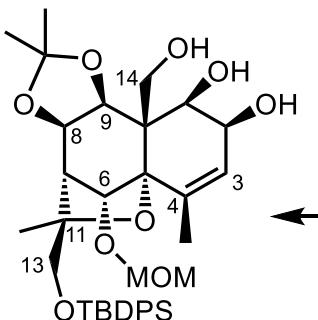
i-Pr₂NLi, THF, -78 °C
then TMSCl, -78 °C
(6/34 = 1:3.0)



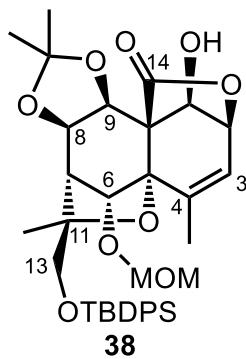
CF₃COOH
DCE, 0 °C
90% (2 steps)



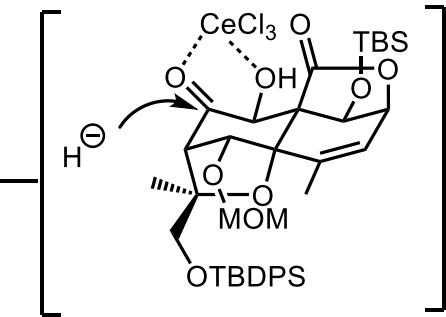
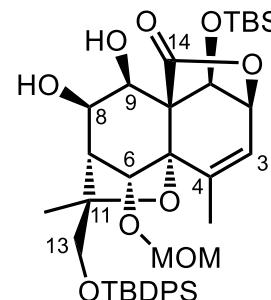
NaBH₄, CeCl₃•7H₂O
MeOH/DCM, -78 °C
95%



LiAlH₄
Et₂O, rt

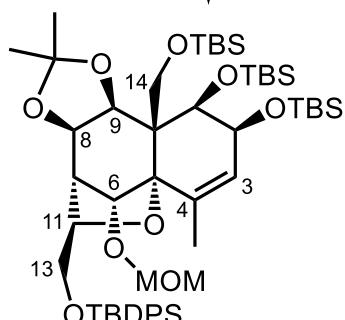


(MeO)₂CMe₂, PPTS
toluene, 80 °C
then *n*-Bu₄NF, THF, 0 °C
100%

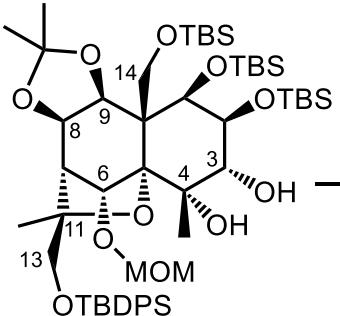


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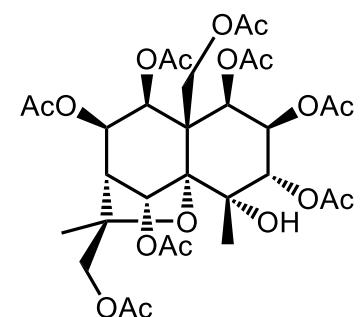
TBSOTf, 2,6-lutidine
DCM, rt
78% (2 steps)



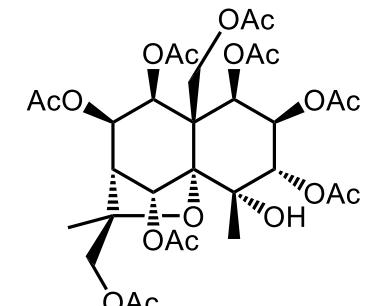
OsO₄, pyridine, 50 °C
then NaHSO₃ aq.
EtOAc, rt
88%



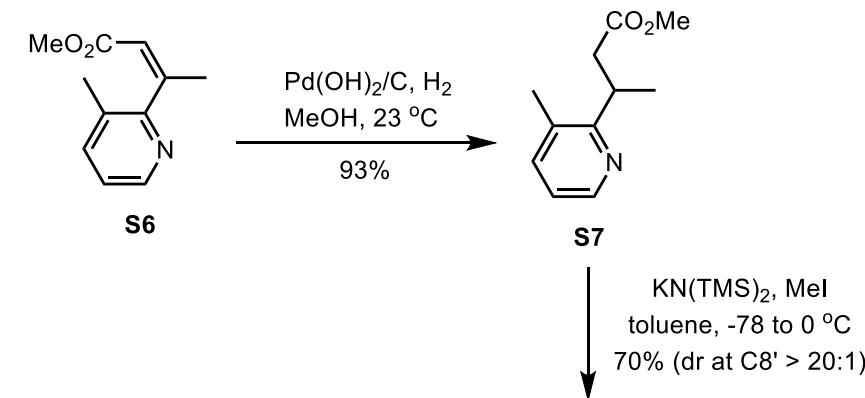
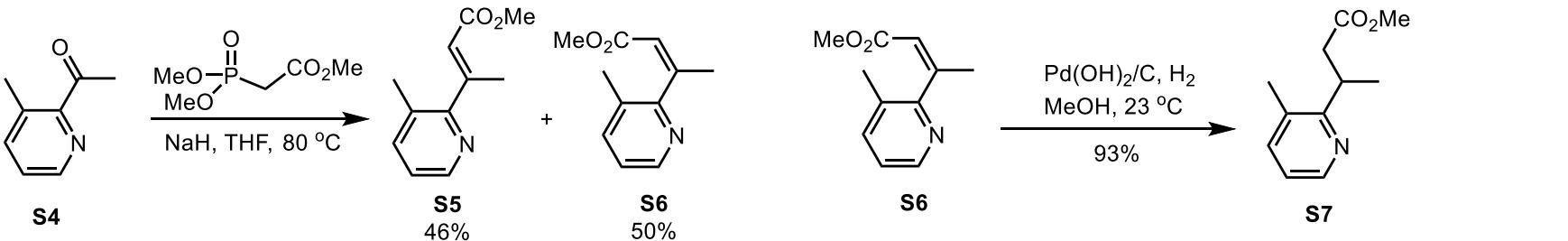
AcOH, H₂O, 100 °C
then *n*-Bu₄F
THF, 80 °C
then Ac₂O, pyridine
DMAP, rt



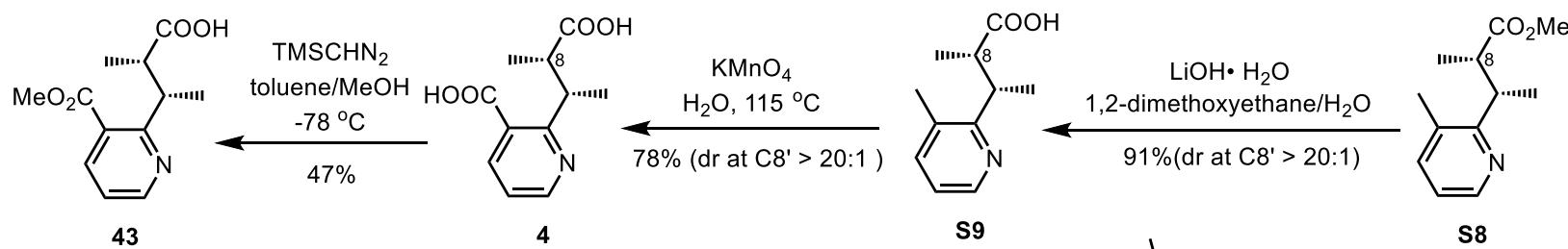
euonyminol nooctaacetate(S3)
7.2%



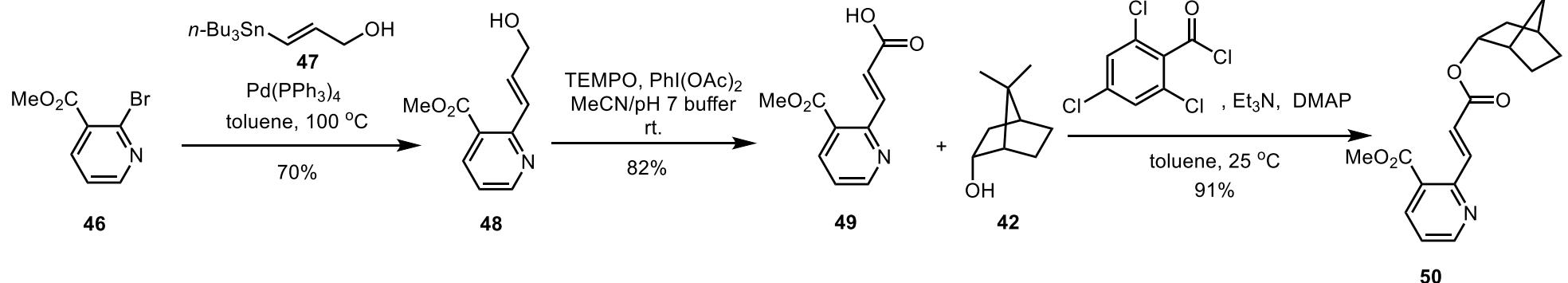
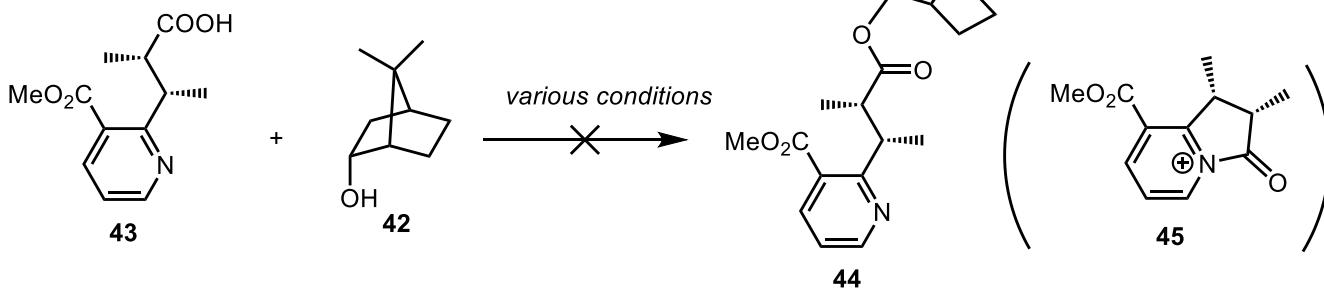
euonyminol octaacetate(2)
66%



KN(TMS)₂, MeI
toluene, -78 to 0 °C
70% (dr at C8' > 20:1)



47%



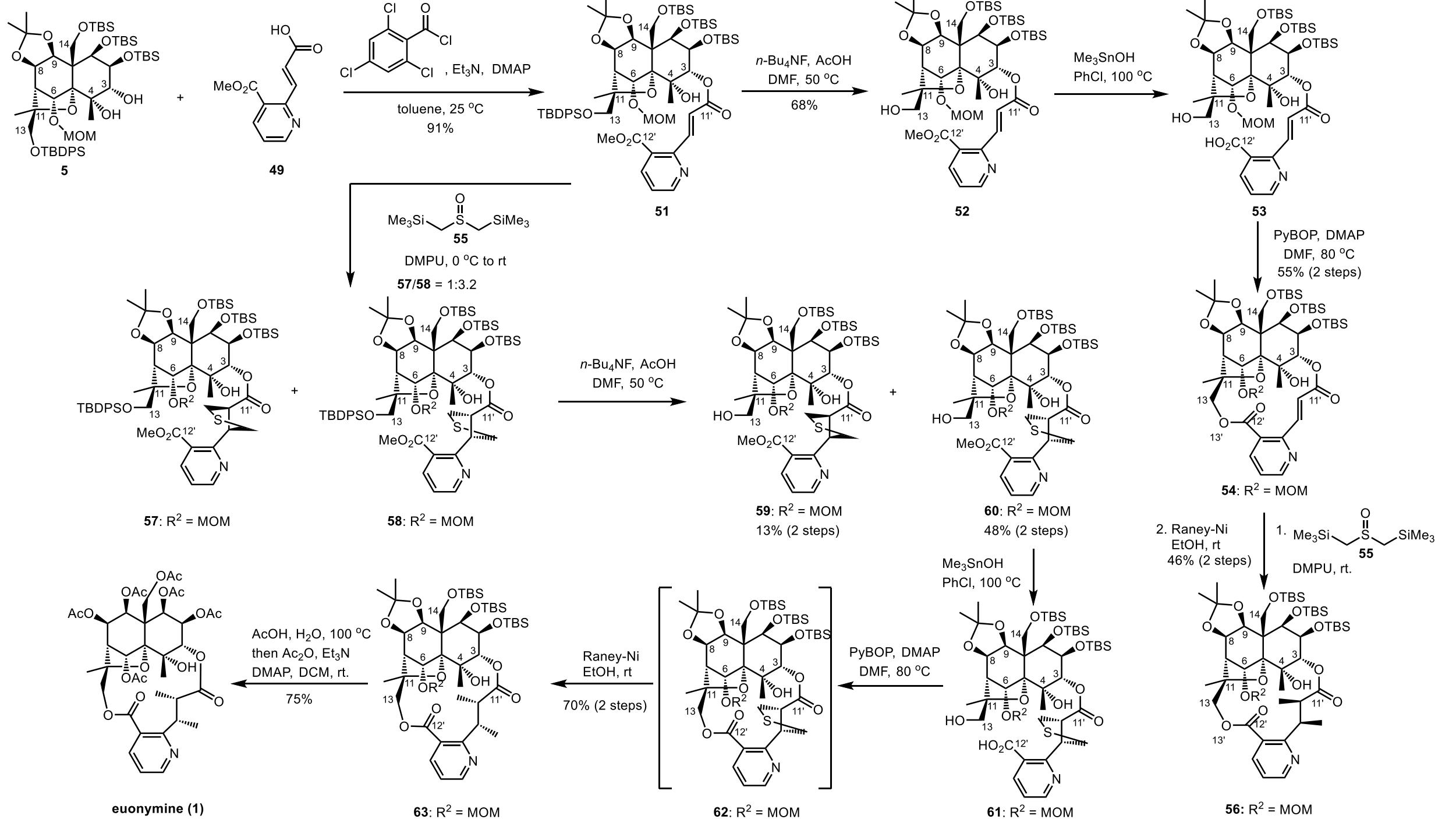
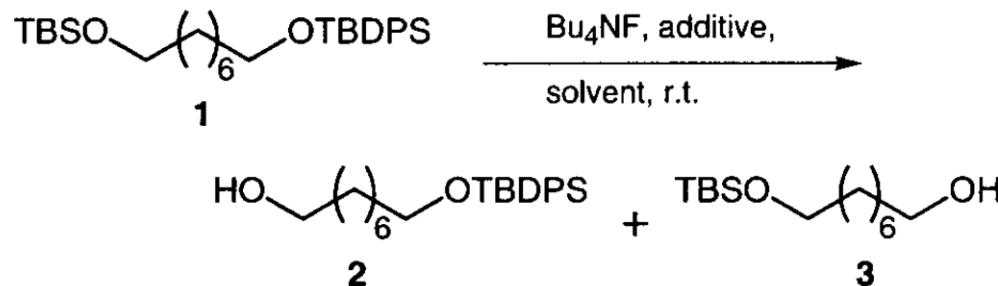


Table 1 Selective Desilylation of **1(a)**



entry	fluoride source(b)	solvent	AcOH / equiv	time / h	combined yield / %	ratio(c) 2 : 3
1	A	THF	0	2	37	6 : 94
2	A	THF	1	17	76	2 : 98
3	A	THF	2	51(d)	43	55 : 45
4	A	THF	5	130(d)	66	>99 : <1
5	B	DMF	0	0.5	25	<1 : >99
6	B	DMF	1	13	79	<1 : >99
7	B	DMF	2	36	22	32 : 68
8	B	DMF	5	64(d)	65	97 : 3
9	A	DMF	1	14	82	<1 : >99

(a) The reaction was carried out in 0.05 M solution of **1** in the specified solvent at r.t. by using 1.0 equiv of the fluoride source. (b) A: 1M Bu₄NF in THF. B: Bu₄NF•xH₂O. (c) Determined by ¹H NMR (270 or 300 MHz) analysis of the chromatographically pure mixture of **2** and **3**. (d) The starting material **1** remained.

Entry	Ester	Product(s)	T [°C]	t [h]	Yield [%]	Selectivity ^[b]							
1	5	5a	80	3	98	—	11	15	15a	70	9	70	≈10:1
2	6	6a	70	1	85	—	12	16	16a	70	10	70	≈3:1
3	7	7a	80	2	88	—	13	17	17a	80	2	67	—
4	8	8a	80	1	87	—	14	18	18a	80	5	100	—
5	9	9a	80	2	100	—	15	19	19a	80	2	80	—
6	10	10a	80	1	77	—	16	20	20a	80	9	70	≈7:1
7	11	11a + 11b	80	7	84	—	17	21	21a	80	5	75	—
8	12	12a	80	2	100	—	[a] Reactions were carried out in 1,2-DCE on a 0.04–0.15-mmol scale and worked up as described in the general procedure. Bn = benzyl; Fmoc = 9-fluorenylmethoxycarbonyl. [b] For the methyl ester.						
9	13	13a	80	5	80	≈90:1							
10	14	14a	80	7	82	≈10:1							

[a] Reactions were carried out in 1,2-DCE on a 0.04–0.15-mmol scale and worked up as described in the general procedure. Bn=benzyl; Fmoc=9-fluorenylmethoxycarbonyl. [b] For the methyl ester.