

A Convergent Total Synthesis of (+)-Ineleganolide

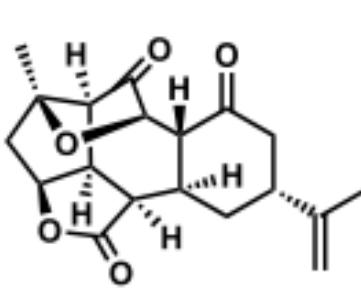
Benjamin M. Gross, Seo-Jung Han, Scott C. Virgil, and Brian M. Stoltz*



Cite This: <https://doi.org/10.1021/jacs.3c02142>

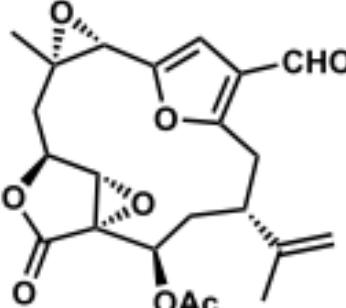


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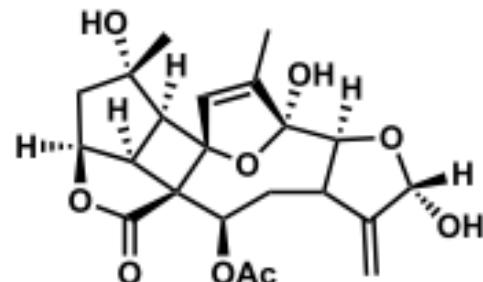
Ineleganolide (1)

cytotoxic against
P-388 murine
leukemia cells
($ED_{50} = 3.82 \mu\text{g/mL}$)



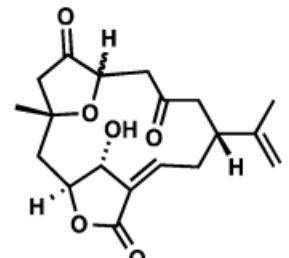
Lophotoxin (2)

potent neurotoxin
irreversible inhibitor of
nicotinic
acetylcholine receptor

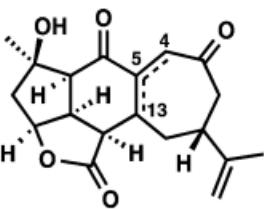


Bielschowskysin (3)

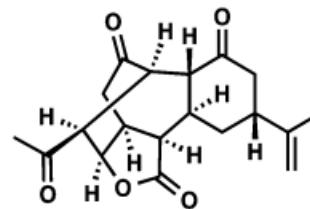
cytotoxic against
EKVX nonsmall cell lung cancer
($GI_{50} < 0.01 \mu\text{M}$)
CAKI-1 renal cancer
($GI_{50} = 0.51 \mu\text{M}$)



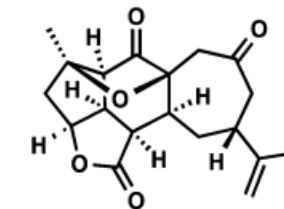
α -H: sinuleptolide (2)
 β -H: 5-episinuleptolide (3)



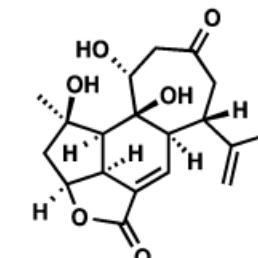
Δ 5-13: scabrolide A (4)
 Δ 4-5: scabrolide B (5)



holiolide (6)

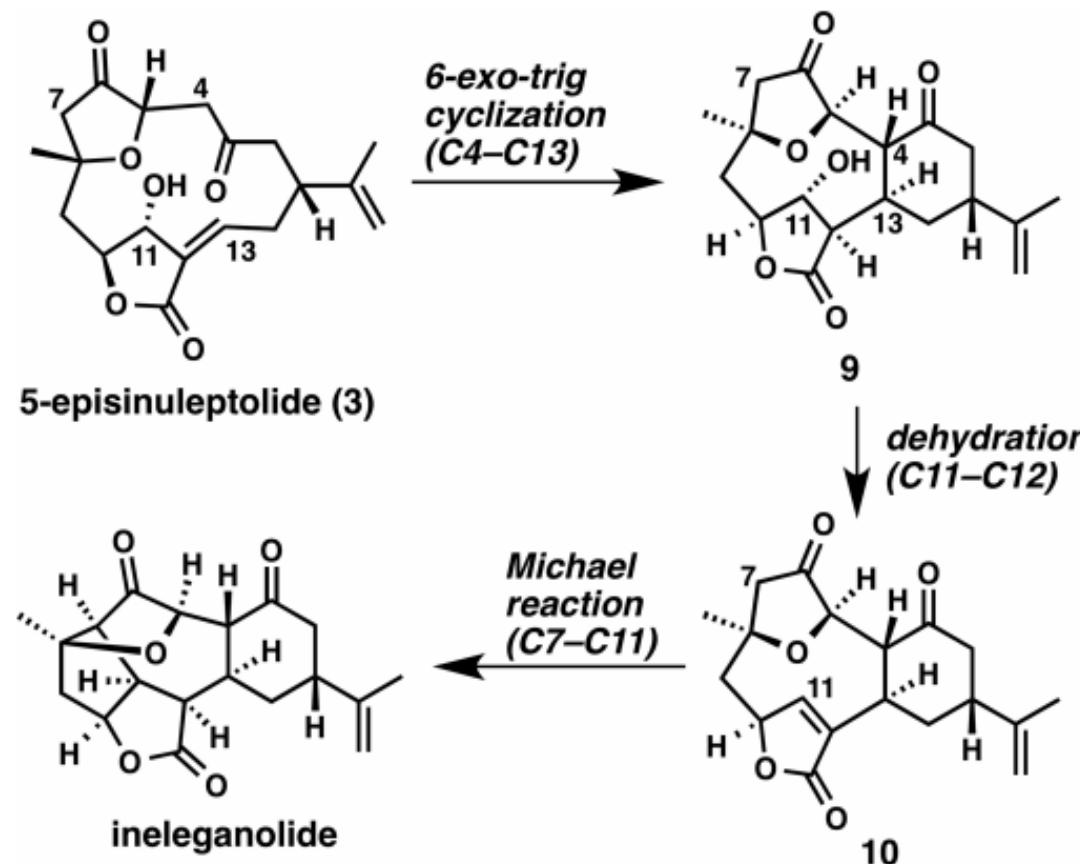


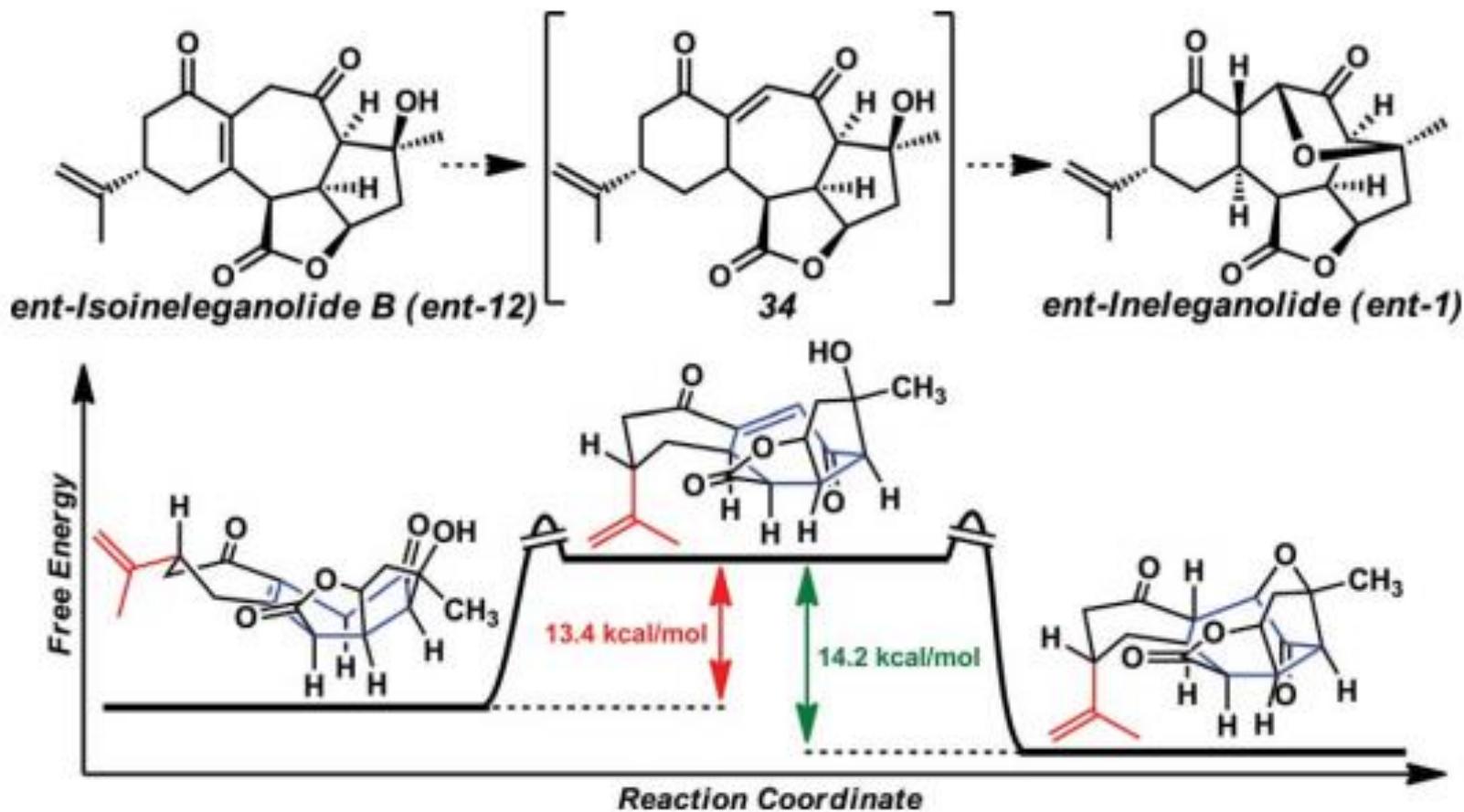
sinulochmodin C (7)



dissectolide (8)

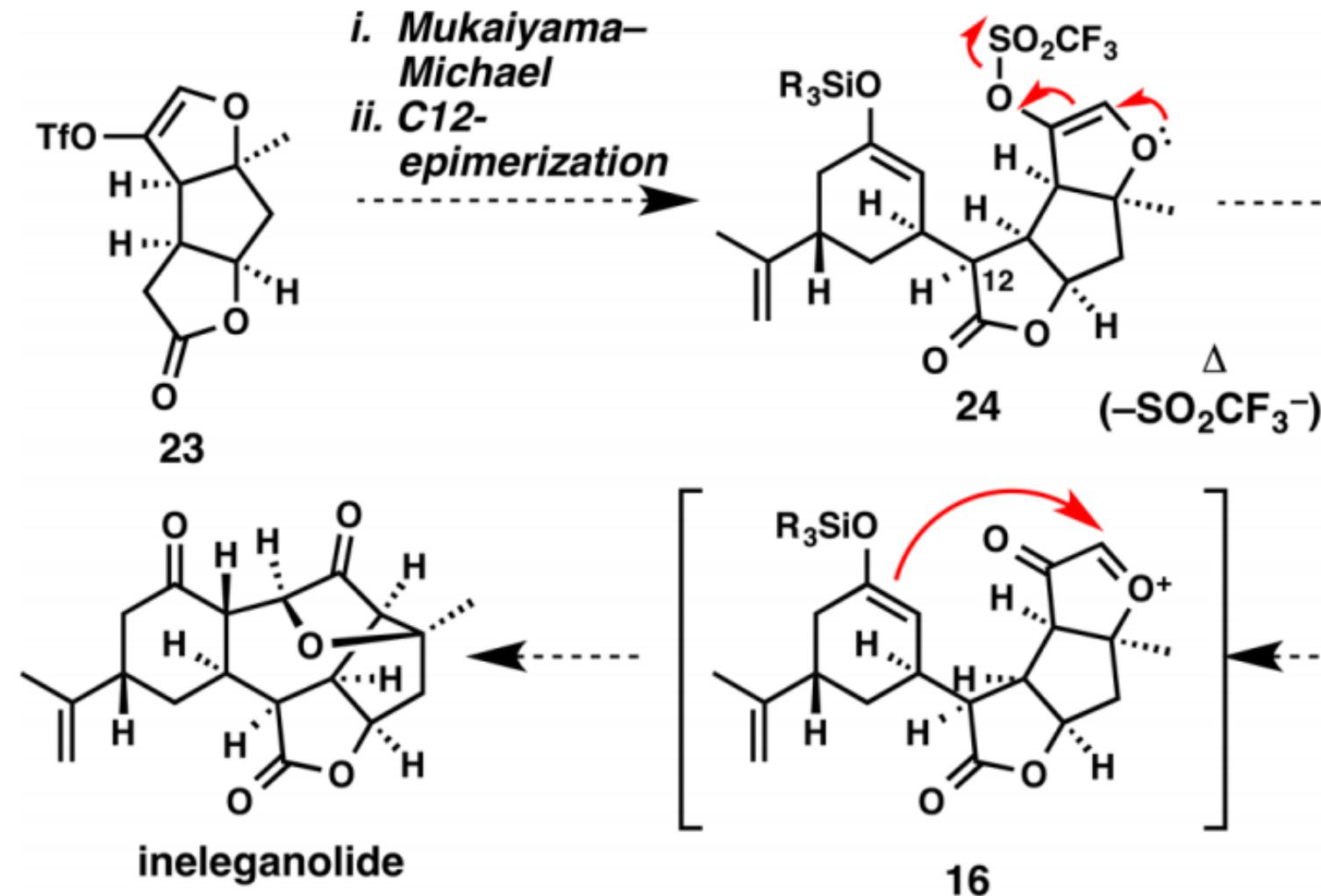
Scheme 1. Proposed Biosynthesis of Ineleganolide and Sinulariadiolide from 5-Episinuleptolide



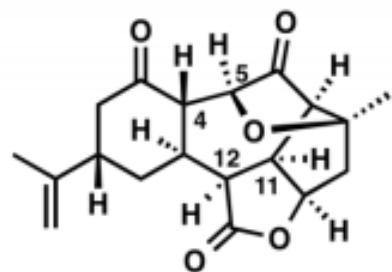


Scheme 6 Conformational assessment and relative ground state energies.

Scheme 4. Proposed Implementation of Triflate Fragmentation in Our Synthesis of Ineleganolide

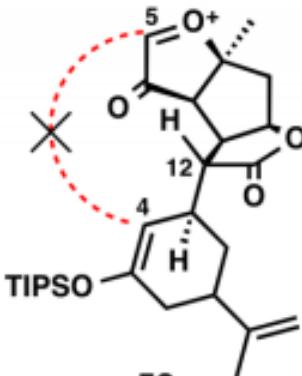


a.



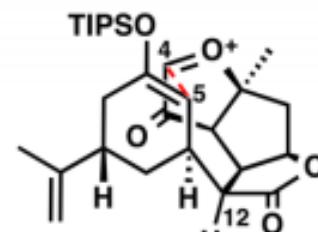
ineleganolide

C4 and C5 cannot reach each other owing to rigidity of tricycle



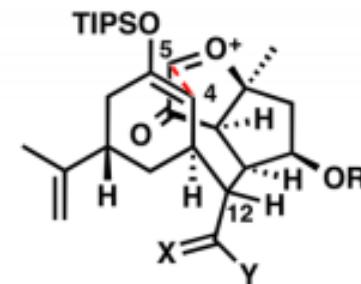
53
(epimeric C12 configuration to ineleganolide)

C4 and C5 held in close proximity to each other



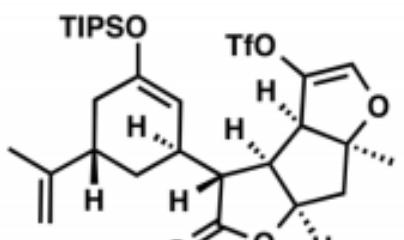
16
(same C12 configuration as ineleganolide)

opening the lactone ring allows C11–C12 bond rotation; C4 and C5 can reach each other regardless of C12 configuration



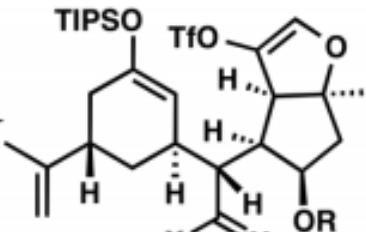
54
(epimeric C12 configuration to ineleganolide)

b.



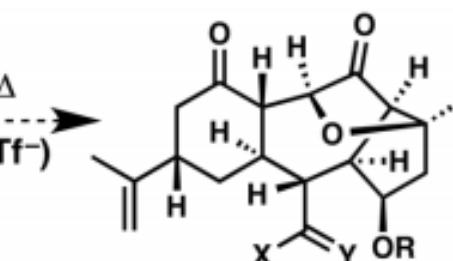
38

open lactone



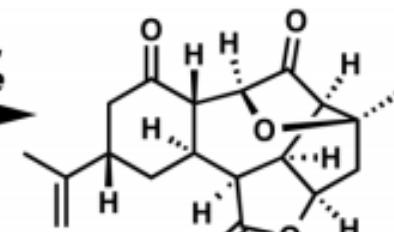
55

Δ

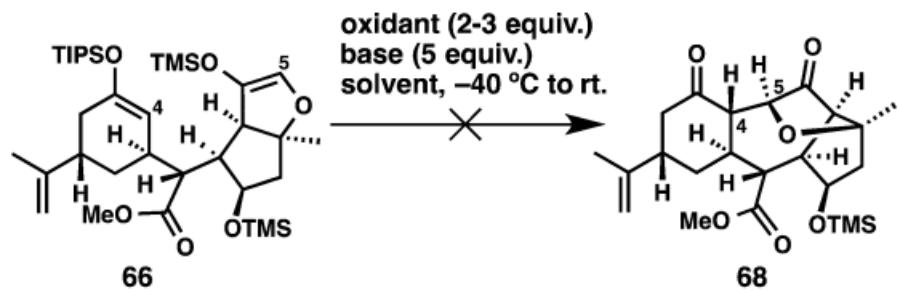


56

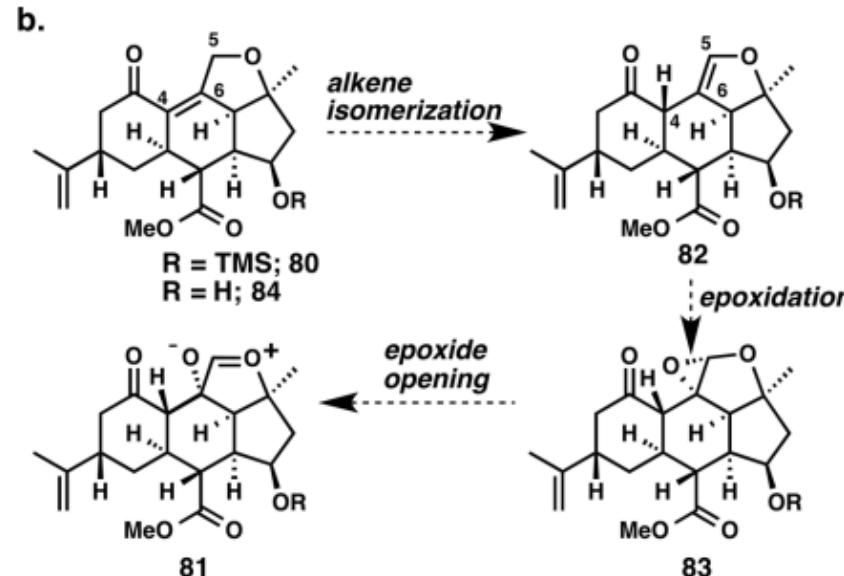
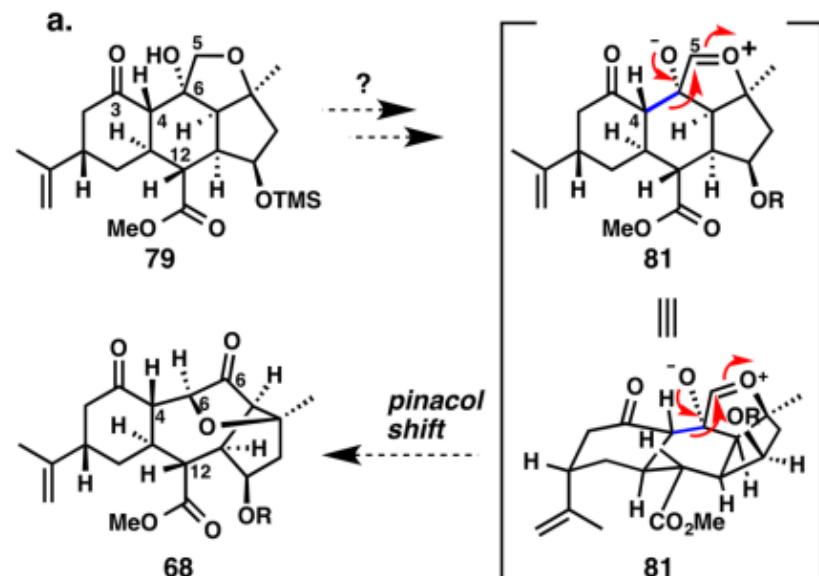
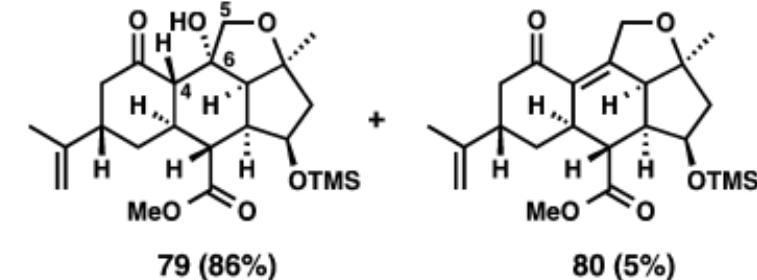
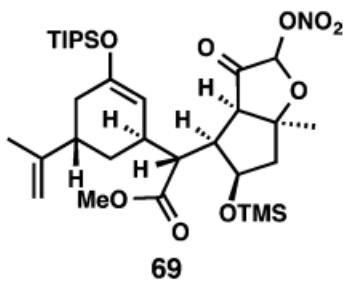
epimerize C12, reform lactone

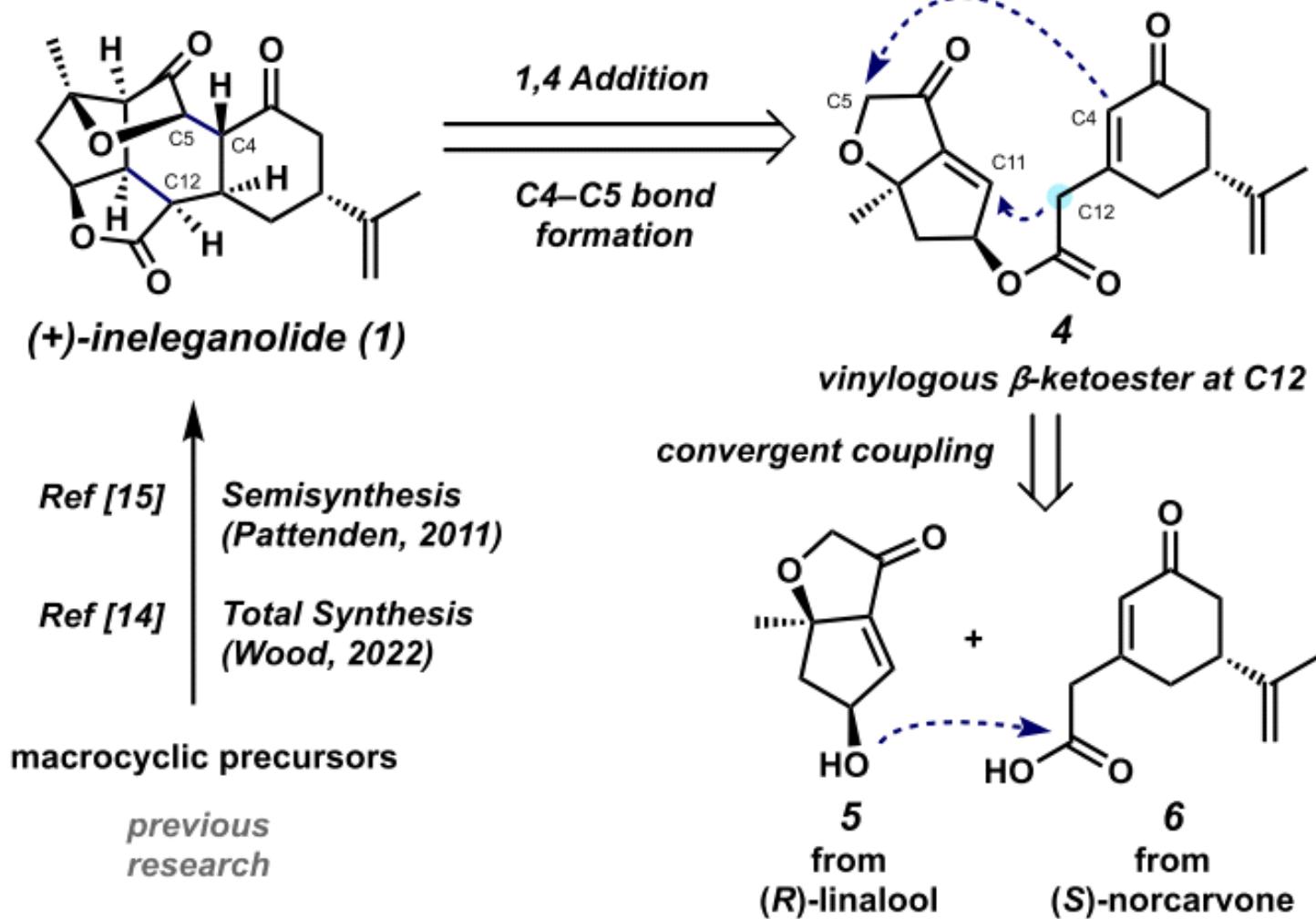


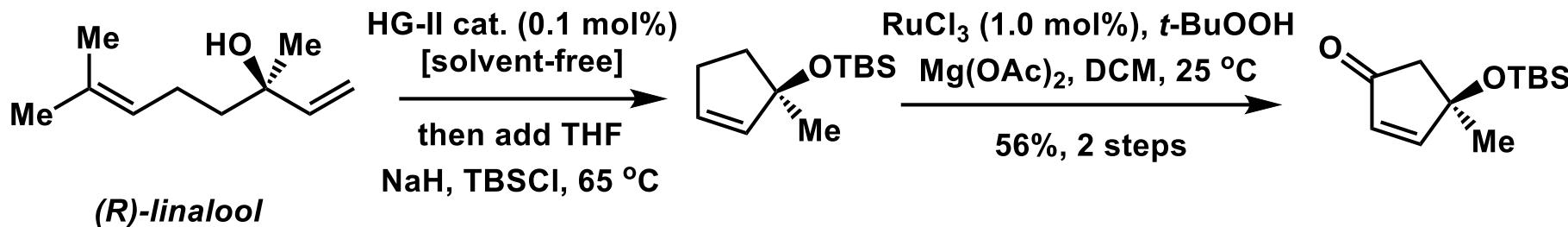
ineleganolide



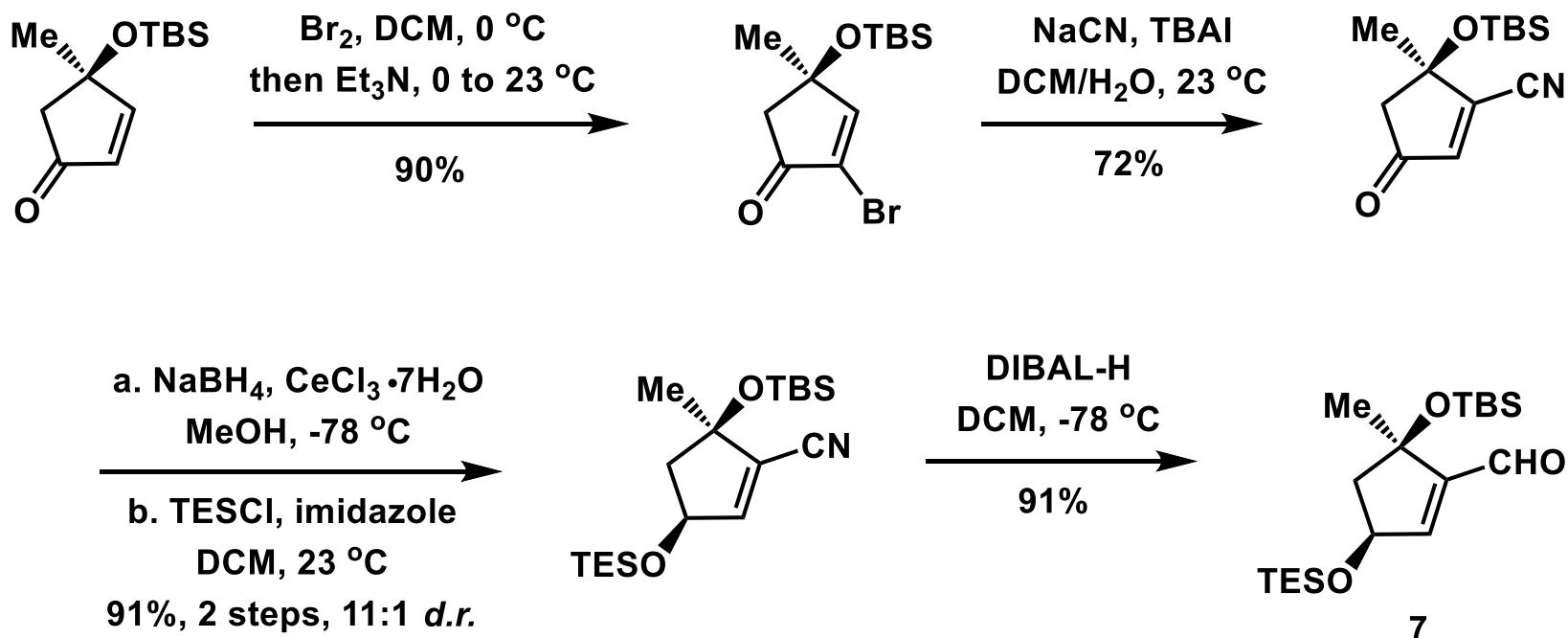
oxidants include: CAN, TBACN, Ag_2O , $\text{Mn}(\text{acac})_3$, $\text{Mn}(\text{hfacac})_3$, CrO_3 , $\text{Pb}(\text{OAc})_4$, NBS, I_2





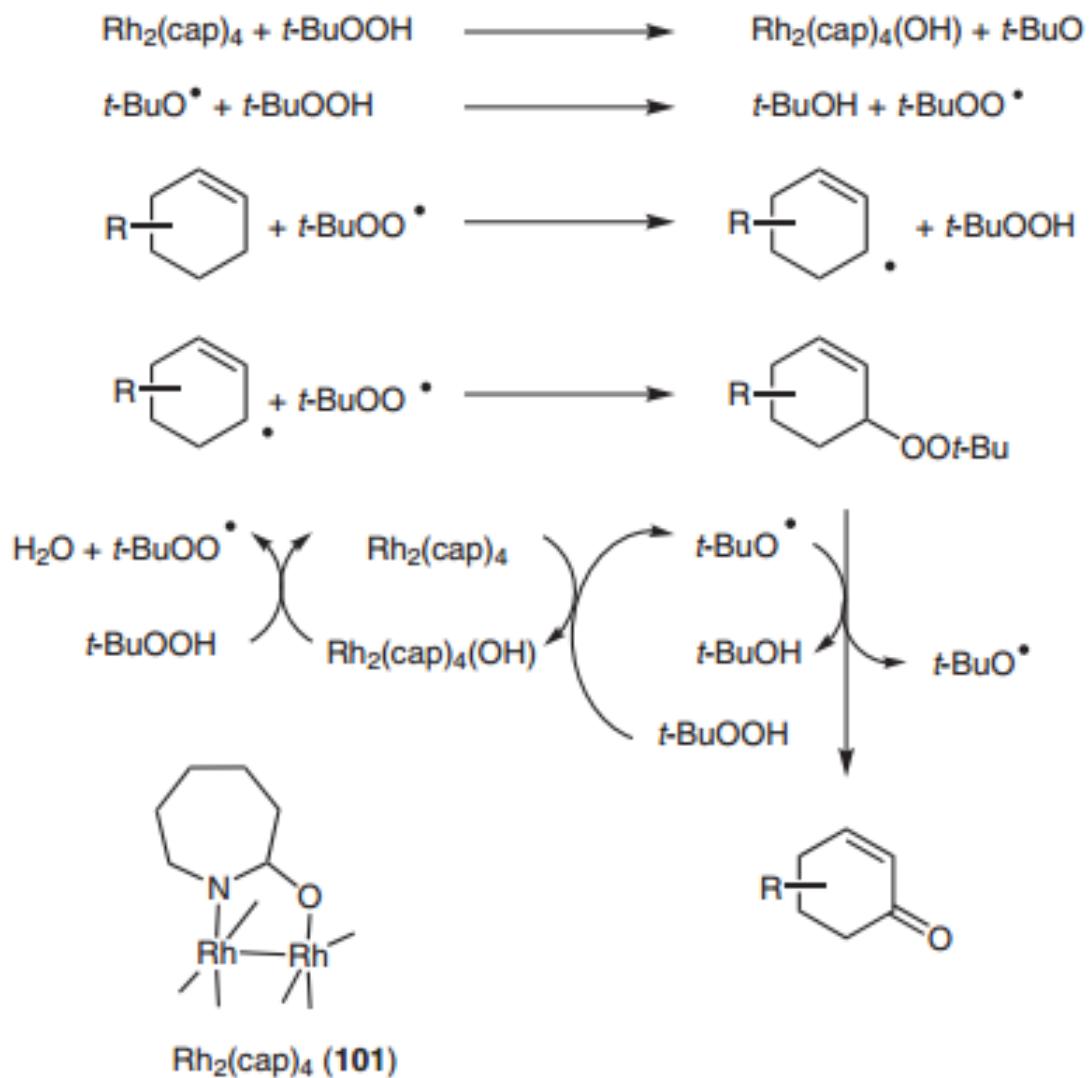


science, 2016, 352, 20232.

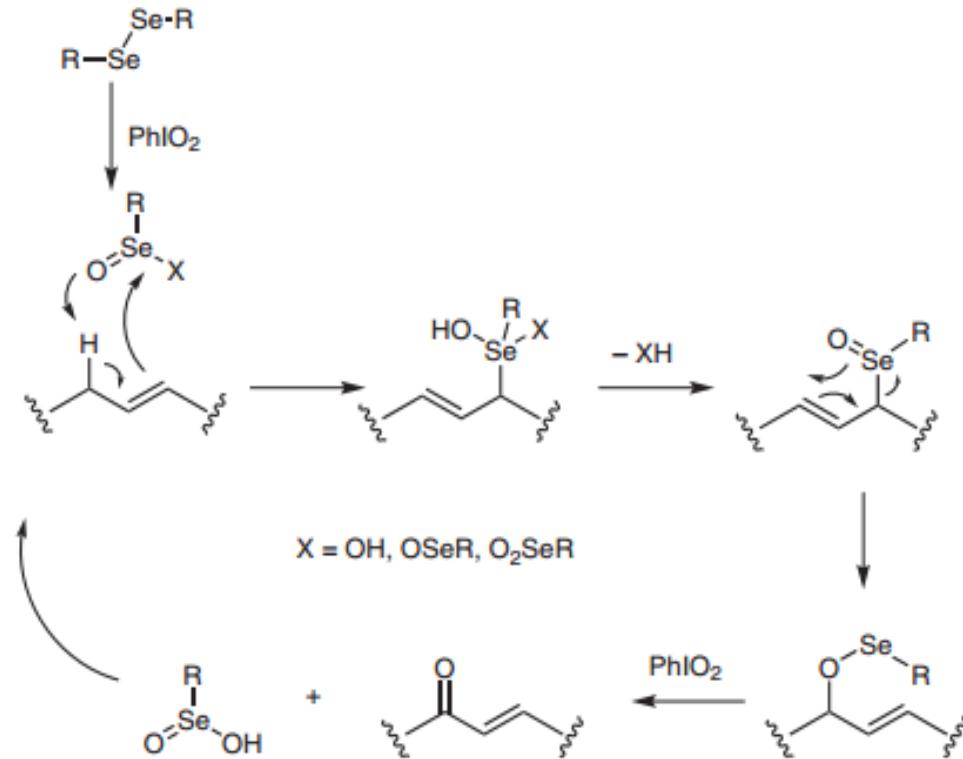


J. Am. Chem. Soc., 2022, 144, 20232.

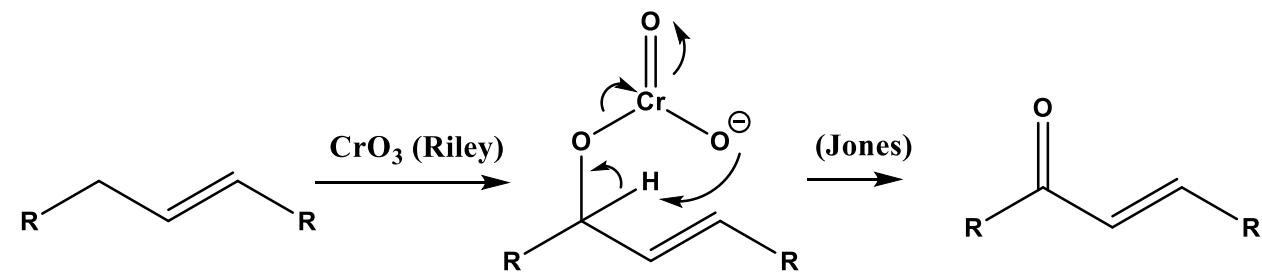
过渡金属催化过氧化物氧化烯丙位

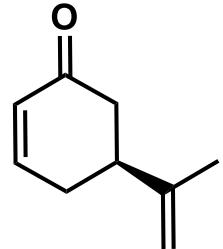


硒氧化烯丙位(Riley Oxidation)

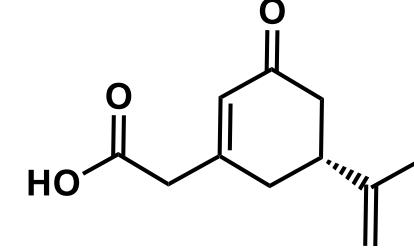
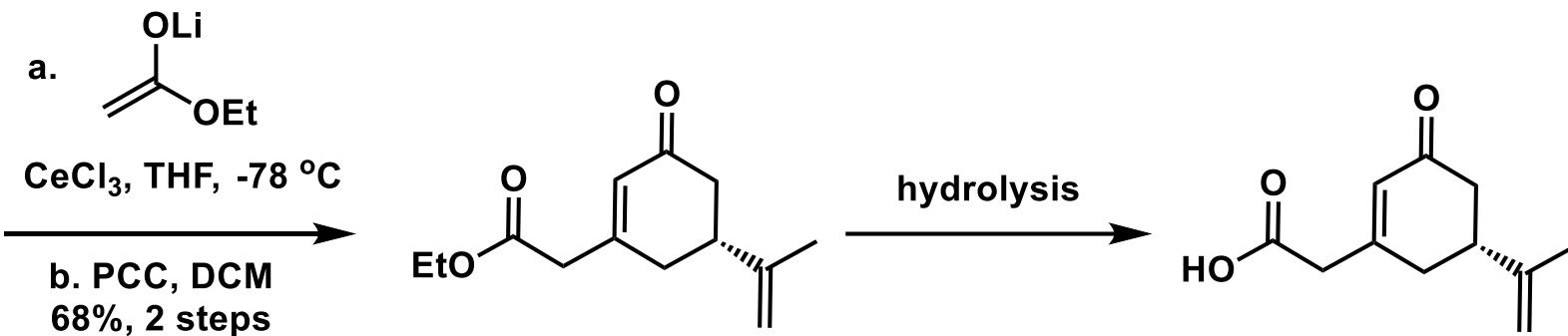


铬氧化烯丙位(Jones Oxidation)



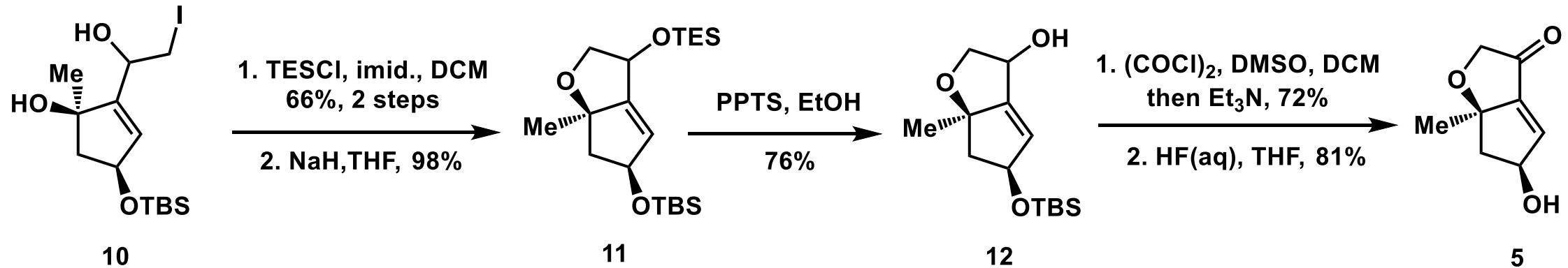
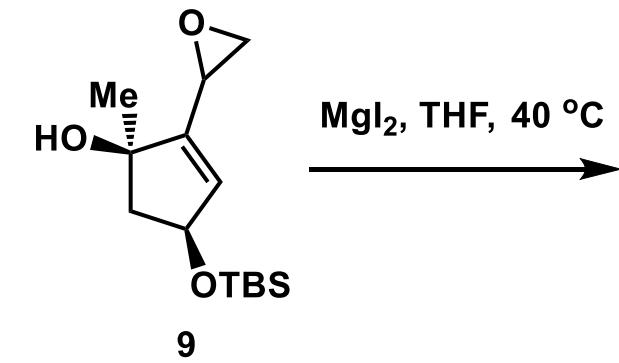
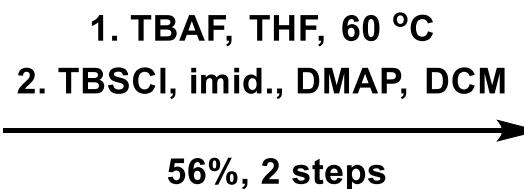
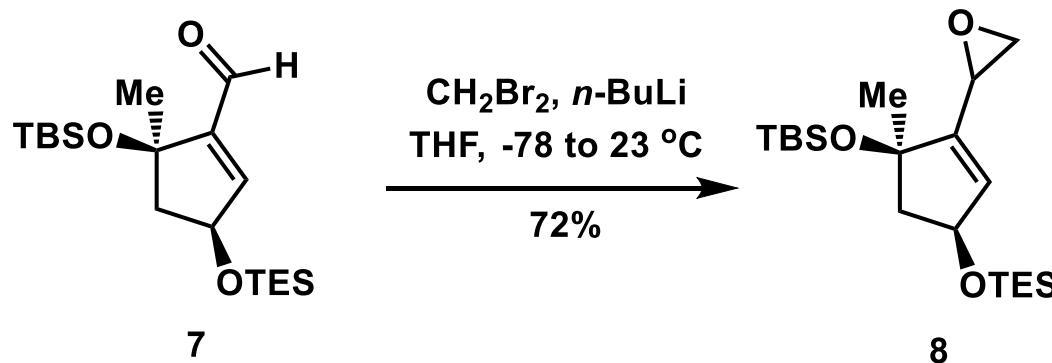


(S)-norarvone



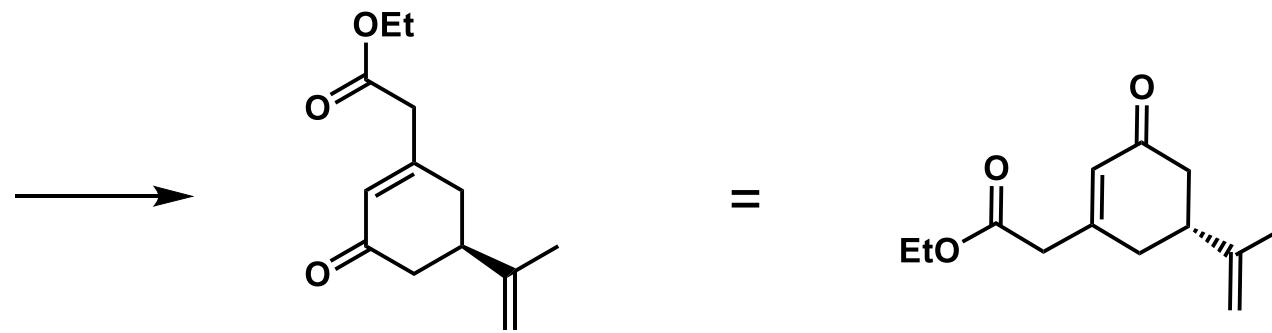
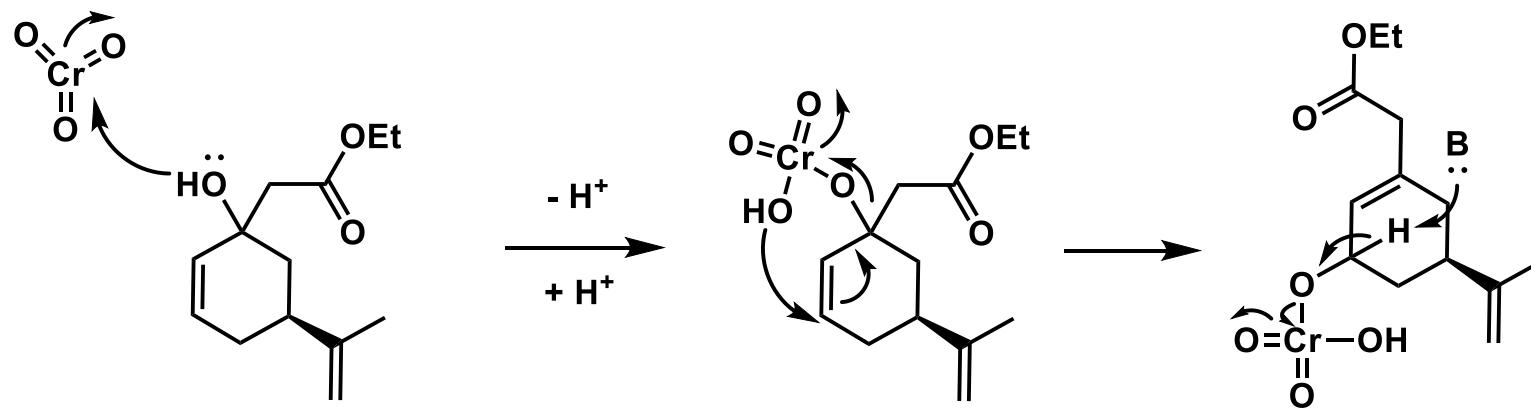
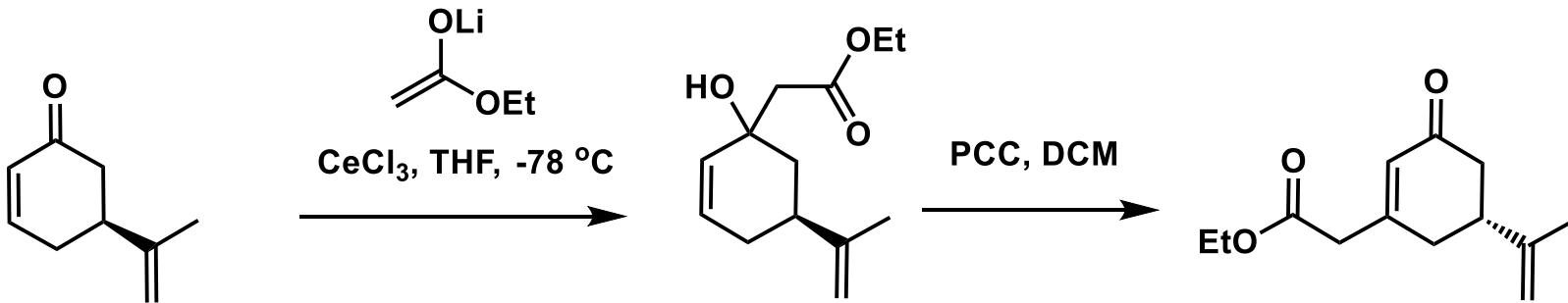
6

Chem. Sci., 2017, 8, 507.

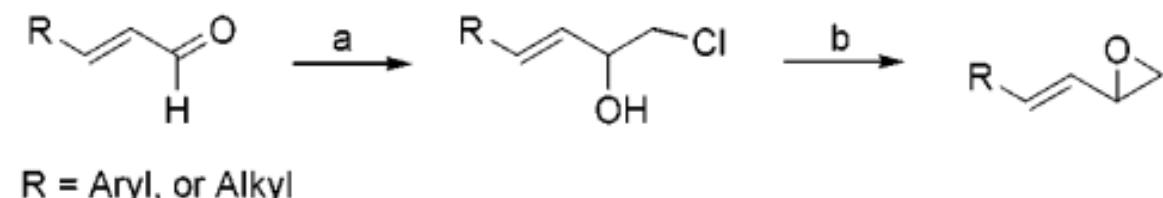


5

Babler–Dauben 氧化重排



Scheme 2. Approach to Synthesis of 2-Vinyloxiranes^a



^a (a) ClCH_2I (1.5 equiv), $n\text{-BuLi}$ (1.5 equiv), THF, -78°C , ~ 1 h; (b) NaH (95%, 1.1 equiv), NaI (10%), THF, 0°C , ~ 1 h.

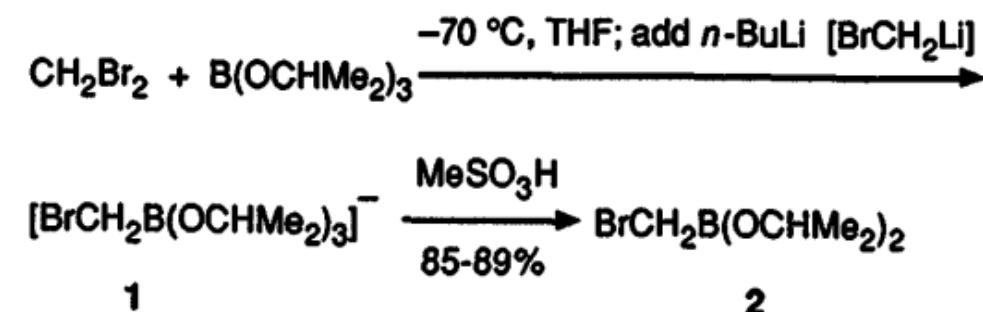
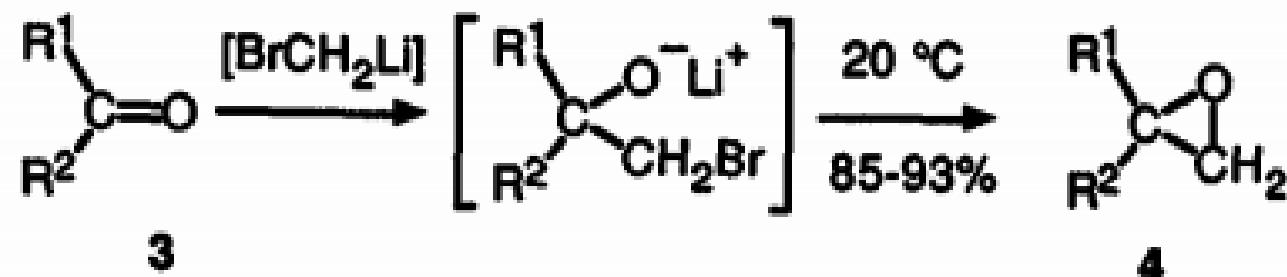


Table 1. Oxiranes (4) from $\text{R}^1\text{R}^2\text{C=O}$ (3), ClCH_2I or CH_2Br_2 , and BuLi .^a

Oxirane	bp, °C (mbar)	Yield from ClCH_2Li , %	from BrCH_2Li
4a	48-51 (0.2)	89	93
4b	60-63 (50)	88	91
4c	46-49 (0.3)	93	93
4d	57-65 (1.7)	40	93

^a Isolated yields, high purity indicated by 200-MHz ^1H NMR. Cpd. 4a-c are known.^{1,3} 4d has been partially characterized previously.⁸

Synlett, 1991, 9, 631.

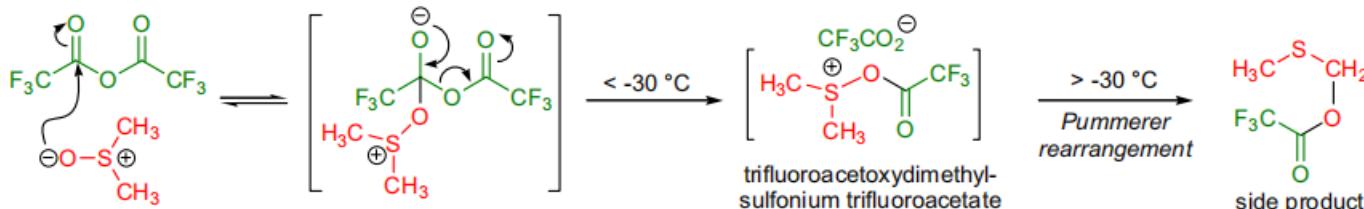
Org. Lett., 2002, 4, 1.

SWERN OXIDATION

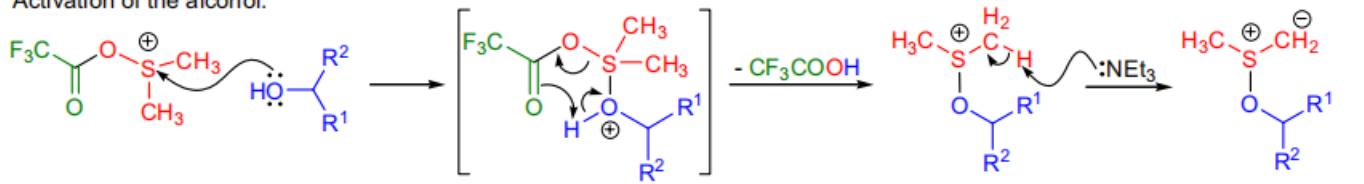
(References are on page 692)

Importance:[Seminal Publications¹⁻⁶; Reviews⁷⁻¹⁰; Modifications & Improvements¹¹⁻¹⁶]Mechanism:⁶⁻⁹

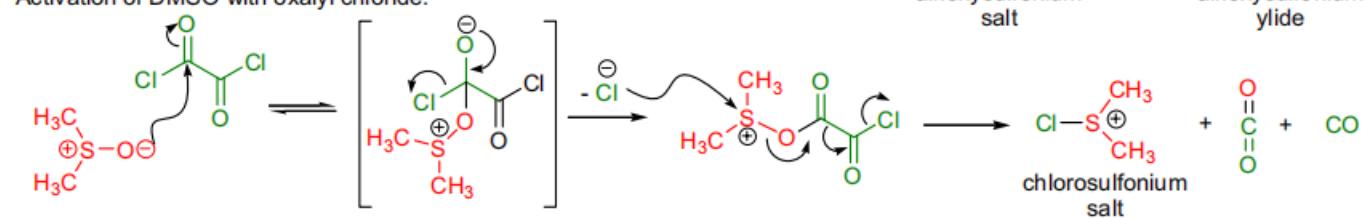
Activation of DMSO with TFAA:



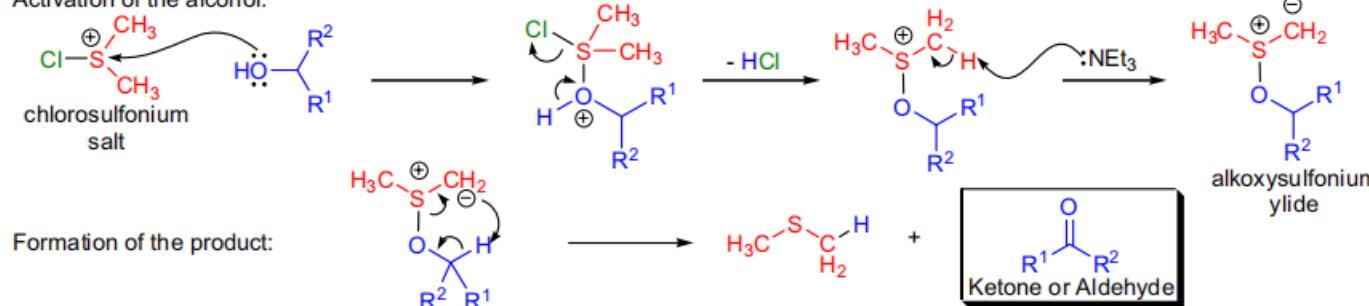
Activation of the alcohol:



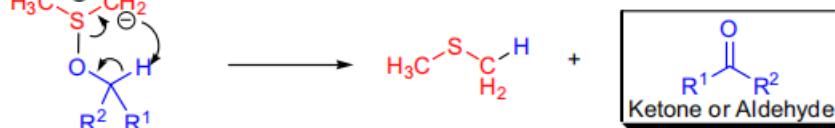
Activation of DMSO with oxalyl chloride:

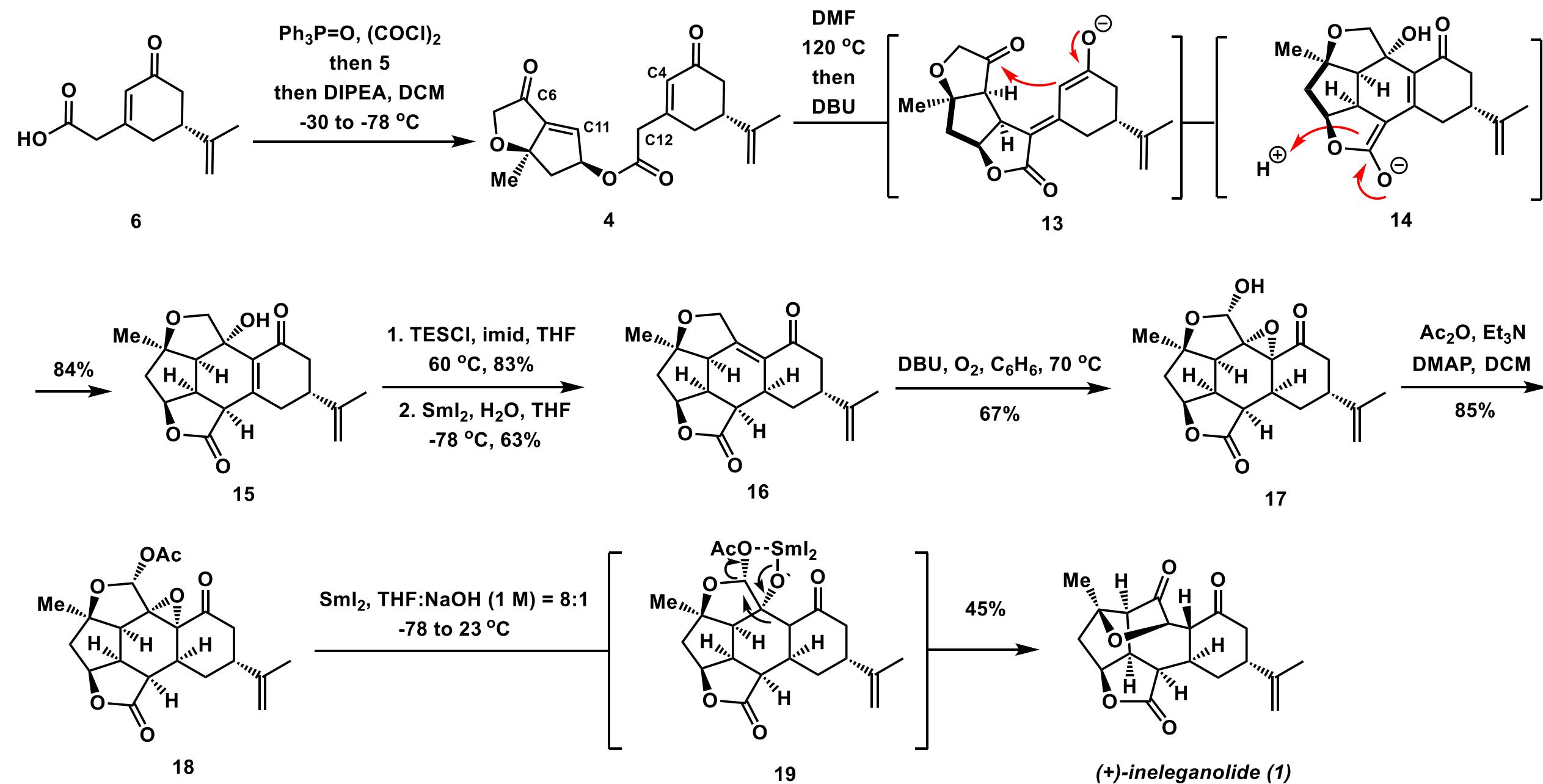


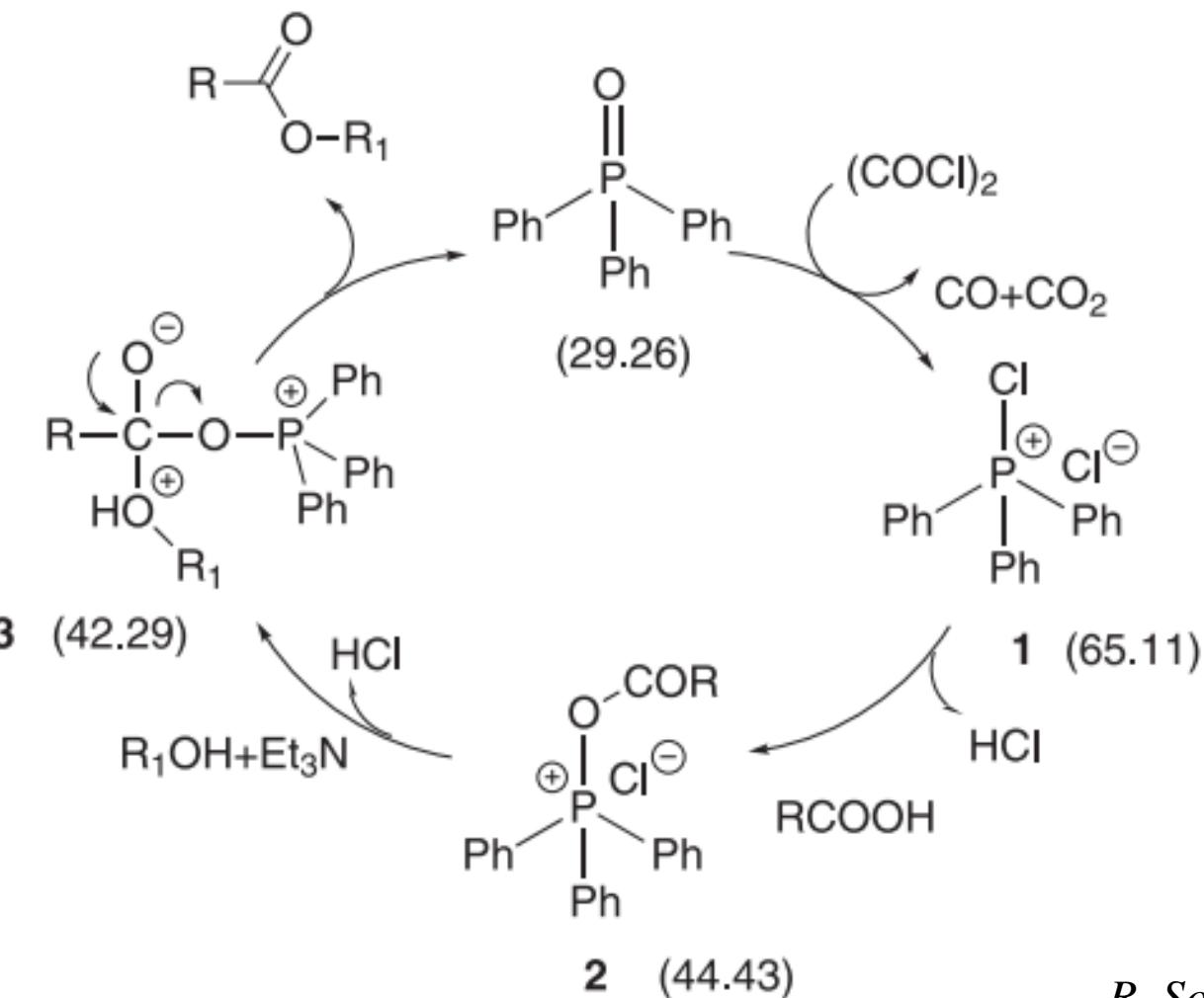
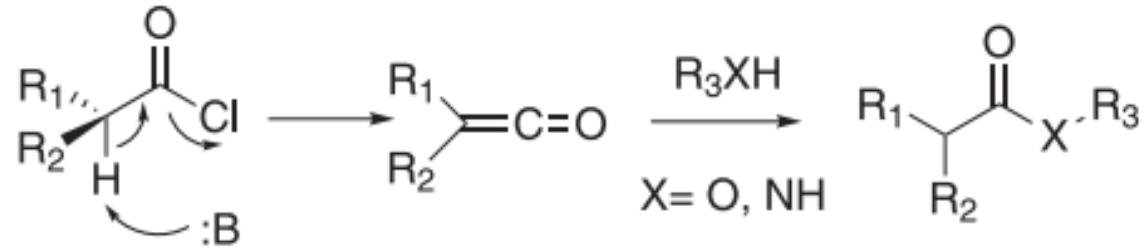
Activation of the alcohol:

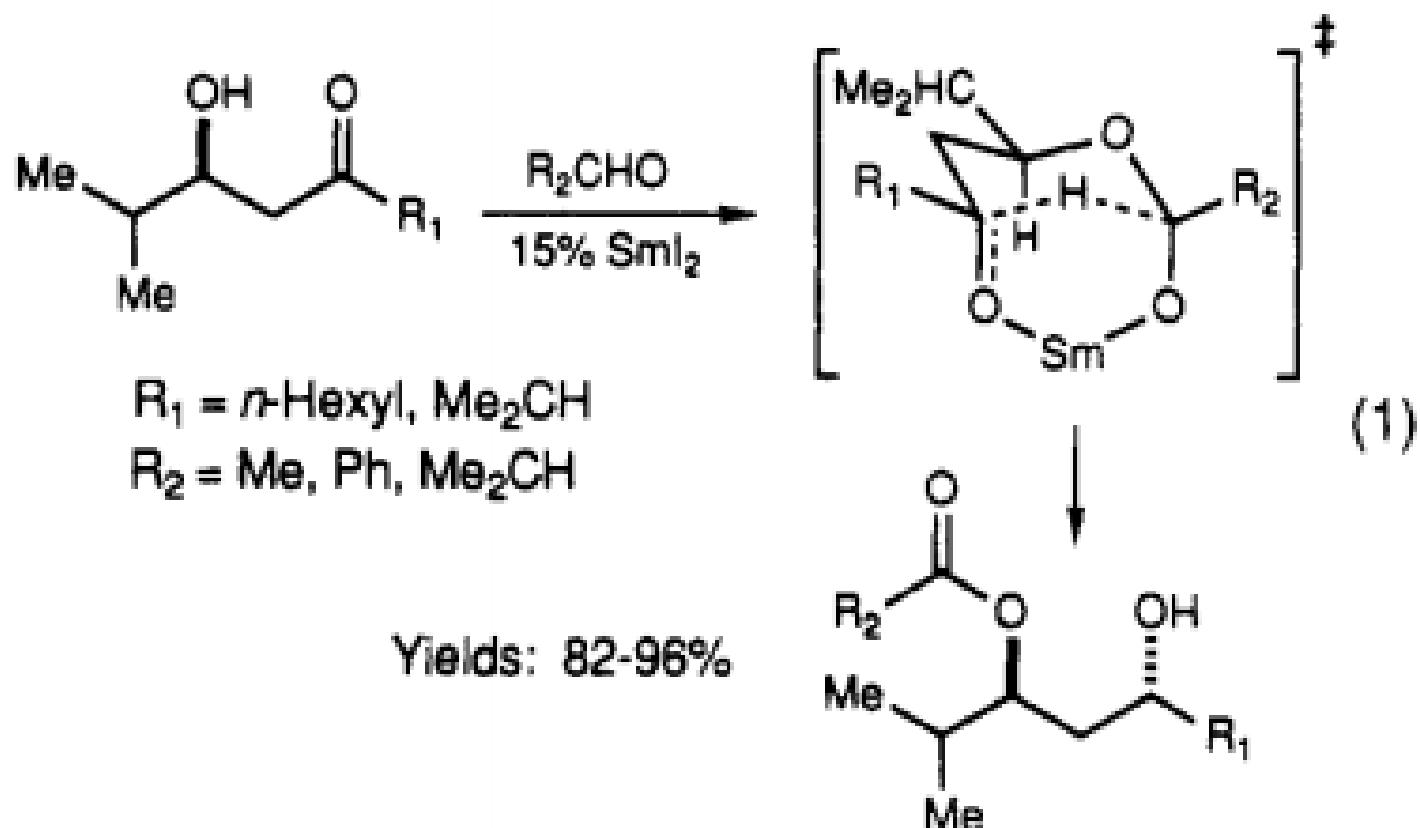
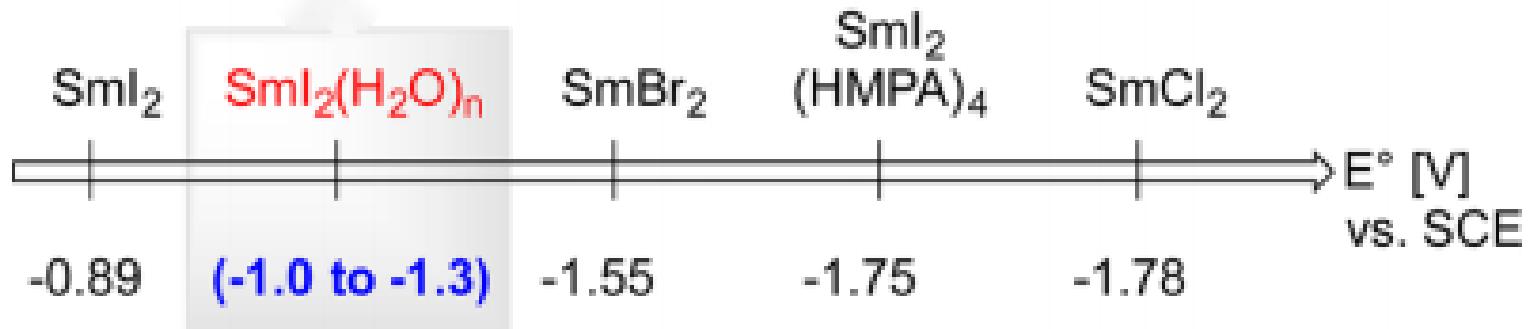


Formation of the product:









J. Org. Chem., 2014, 79, 2522.

J. Am. Chem. Soc., 1990, 112, 6447.

