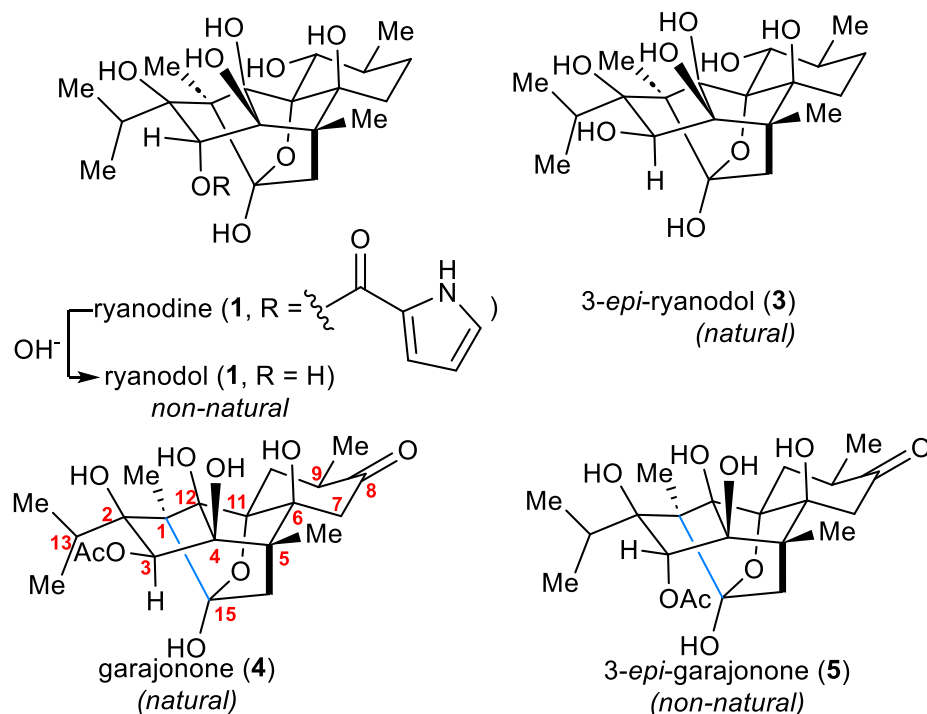


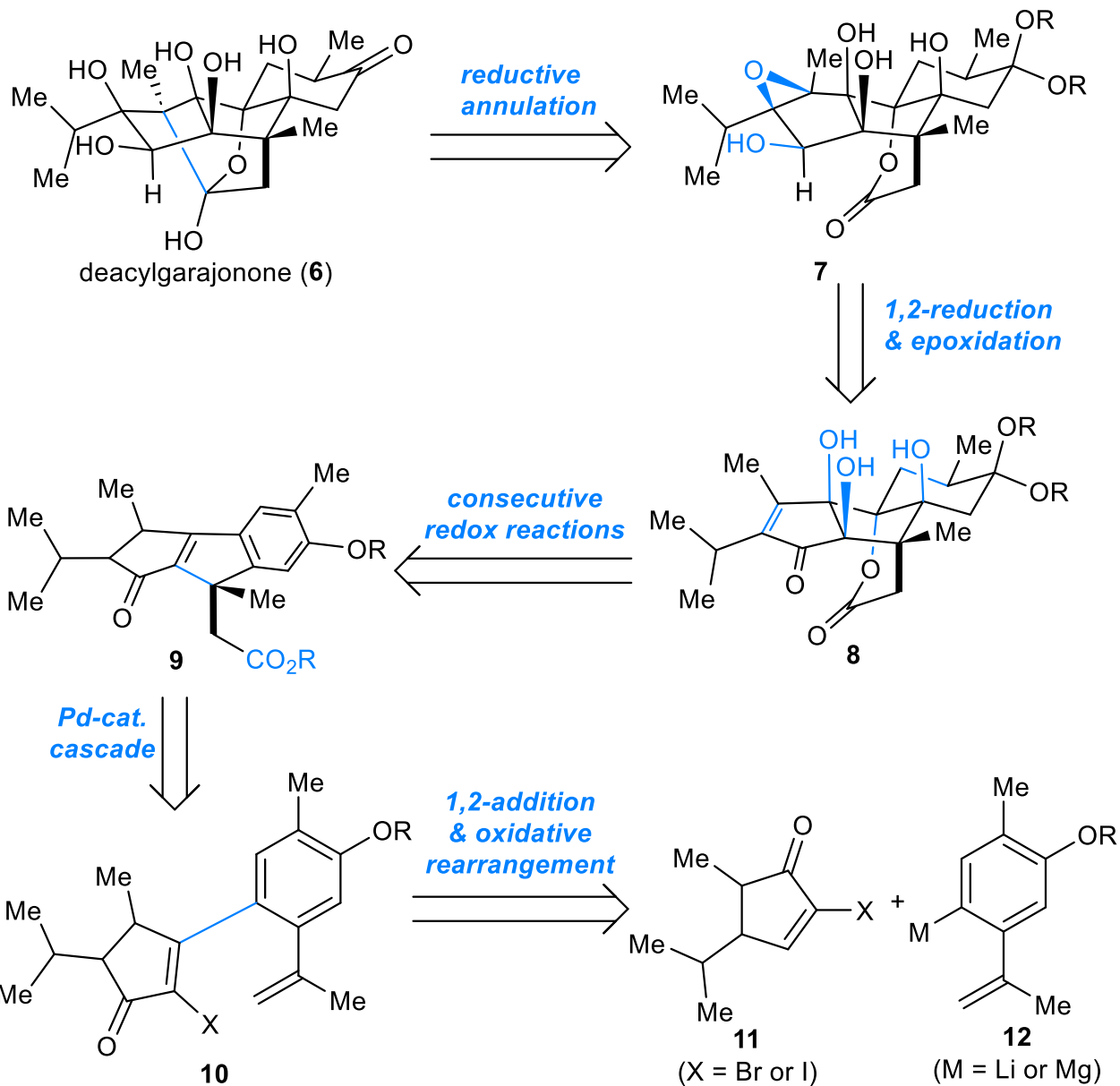
RESEARCH ARTICLE

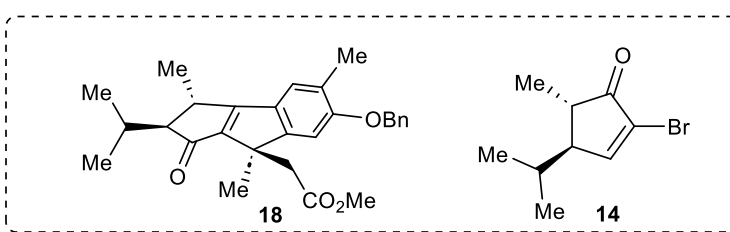
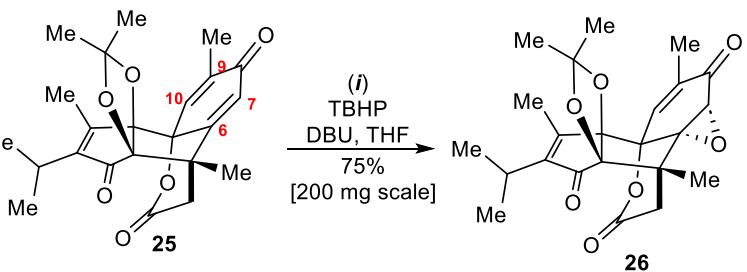
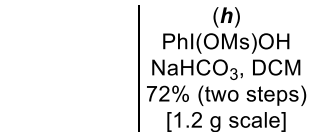
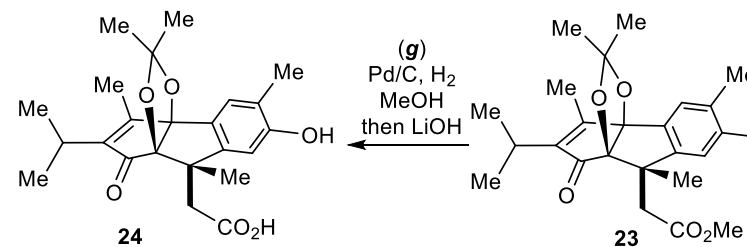
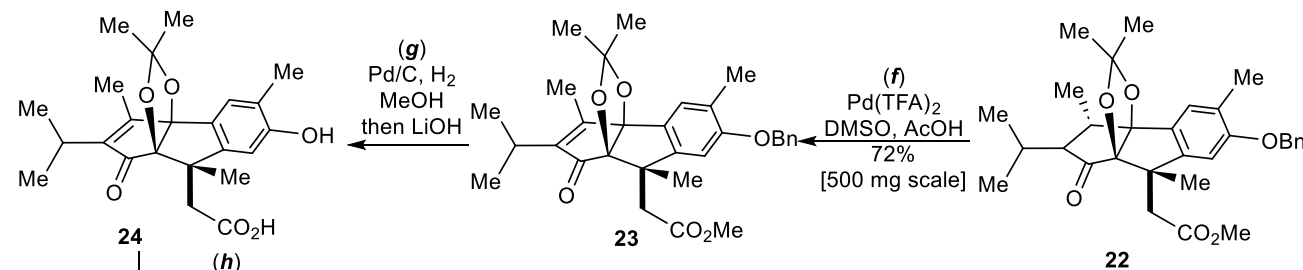
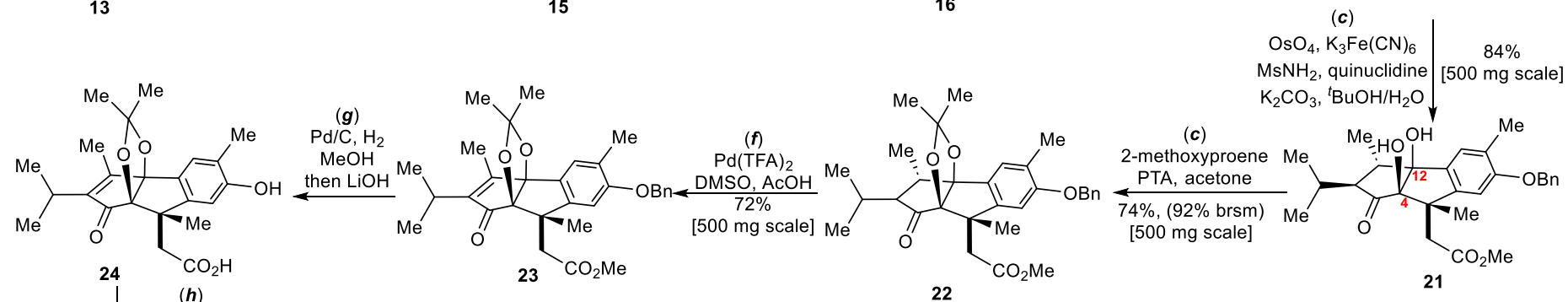
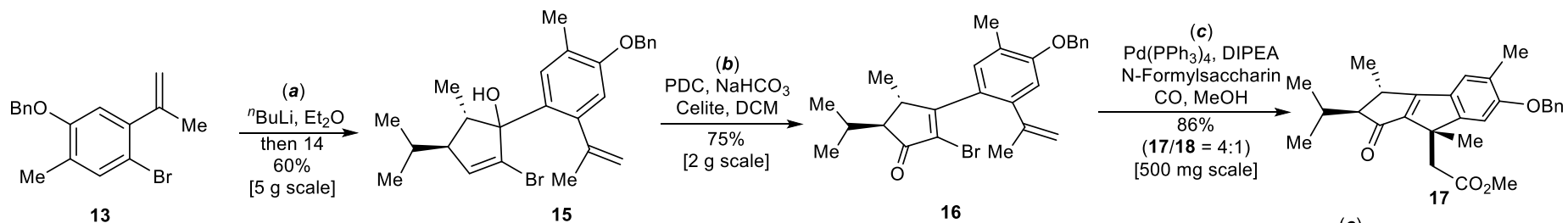
Total Synthesis of Ryanodane Diterpenoids Garajonone and 3-*epi*-Garajonone

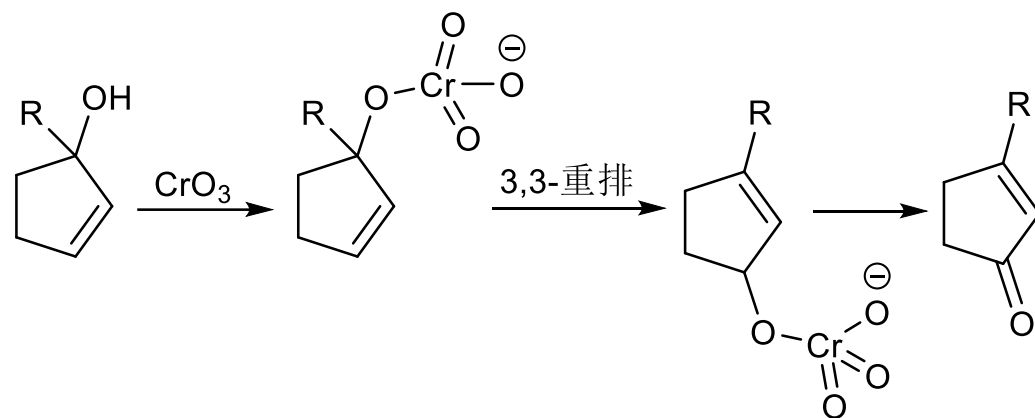
Jin-Bao Qiao,^[§.a] Long Meng,^[§.a] Jia-Yi Pei,^[a] Hui Shao,^[a] and Yu-Ming Zhao^[*,a]



DOI: 10.1002/ange.202417647.

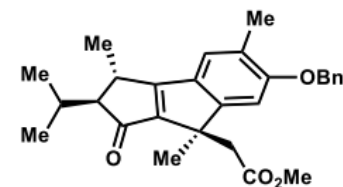




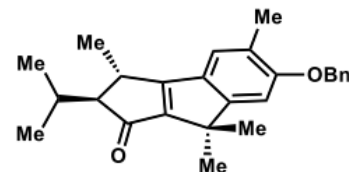




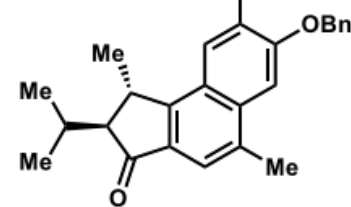
Entry	Catalyst	Ligand	Base	Additive	T (°C)	Combined yield (17/18, ratio)	Yield (19)	Yield (20)
1	Pd(dba) ₂	PCy ₃	K ₃ PO ₄	-	rt to 100	0	36%	60%
2	Pd(OAc) ₂	PPh ₃	NEt ₃	-	rt to 100	43% (2.5:1)	16%	35%
3	Pd(PPh ₃) ₄	-	NEt ₃	-	100	59% (3.7:1)	13%	20%
4	Pd(TFA) ₂	PPh ₃	NEt ₃	-	100	66% (2.6:1)	12%	18%
5	Pd ₂ (dba) ₃	PPh ₃	NEt ₃	-	100	52% (4.0:1)	27%	18%
6	Pd(TFA) ₂	PPh ₃	NEt ₃	S3	100	80% (2.9:1)	<5%	<5%
7	Pd(TFA) ₂	PPh ₃	NEt ₃	S4	100	65% (3:1)	11%	19%
8	Pd(TFA) ₂	PPh ₃	NEt ₃	S5	100	75% (3:1)	<5%	<5%
9	Pd(TFA) ₂	PPh ₃	Na ₂ CO ₃	S3	100	28% (3.6:1)	<5%	<5%
10	Pd(TFA) ₂	PPh ₃	DIPEA	S3	100	60% (3.1:1)	10%	25%
11 ^[d]	Pd(TFA) ₂	PPh ₃	NEt ₃	S3	100	0	0	28%
12	Pd(TFA) ₂	-	NEt ₃	S3	100	0	0	0
13	Pd(TFA) ₂	PPh ₃	-	S3	100	23% (3:1)	<5%	<5%
14 ^[d]	Pd(PPh ₃) ₄	-	NEt ₃	S3	100	0	0	24%
15	Pd(PPh ₃) ₄	-	NEt ₃	S3	100	84% (3.3:1)	<5%	<5%
16	Pd(PPh ₃) ₄	-	K ₃ PO ₄	S3	100	48% (4:1)	22%	13%
17	Pd(PPh ₃) ₄	-	DIPEA	S3	100	86% (4:1)	<5%	<5%



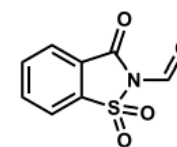
18 (minor diastereomer)



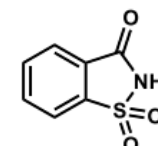
19



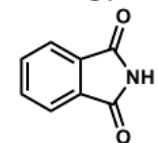
20 (via 6-endo-type Heck)



S3

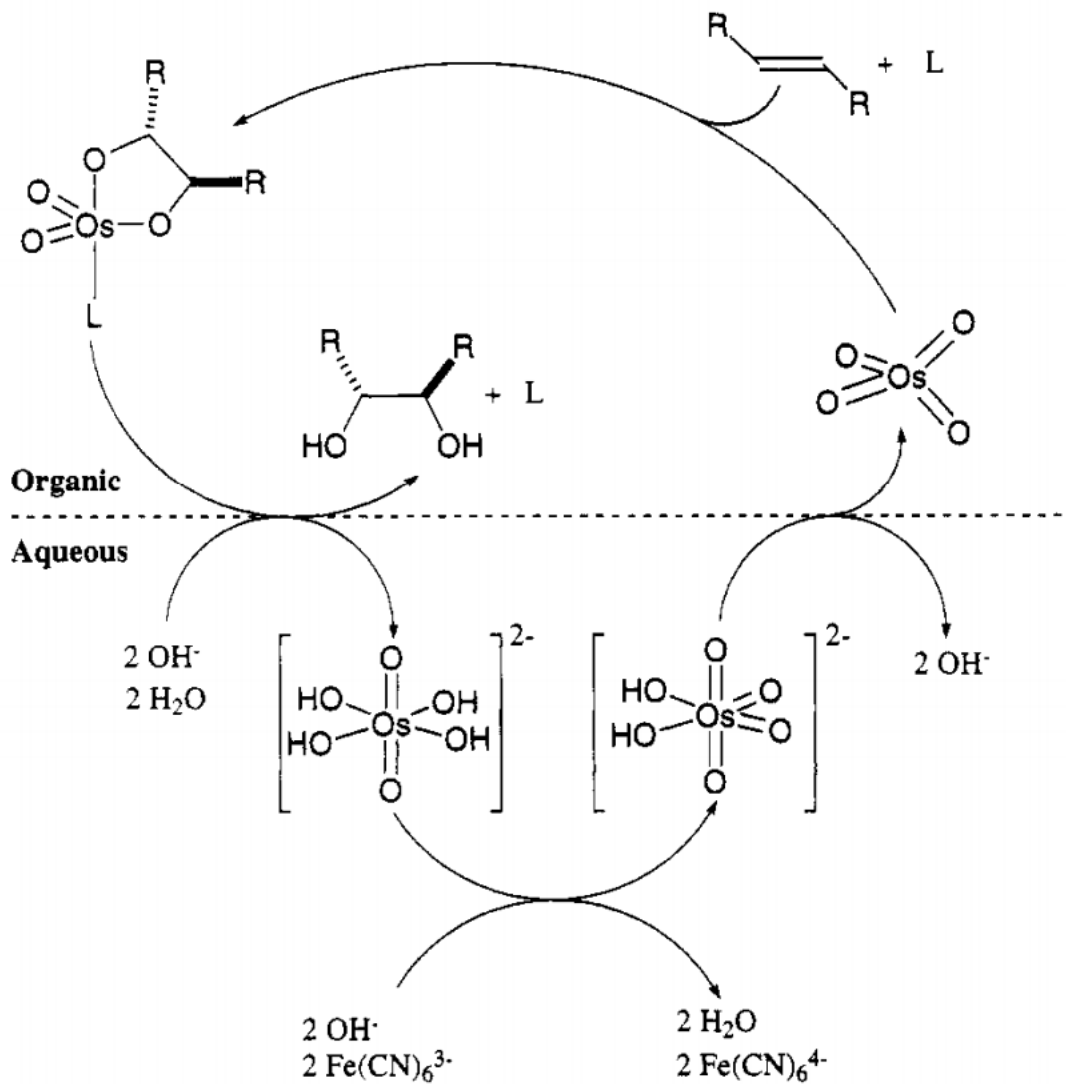


S4



S5

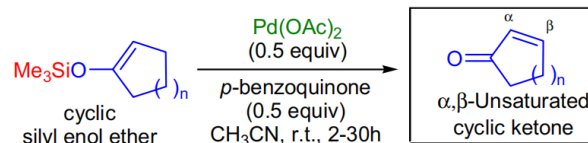
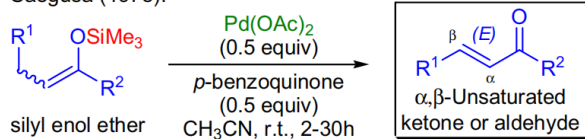
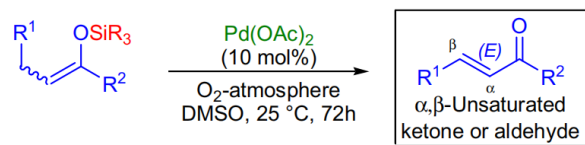
[a] All reactions were carried on 0.02 mmol, catalyst (10 mol%), ligand (20 mol%) in MeOH (2 mL) and were purged with carbon monoxide (CO balloon) for 20 seconds then placed into a preheated 100 °C oil bath. [b] Using the CH₂Br₂ as an internal standard to determine the yield of all reactions. [c] The ratio of stereoisomer was determined by H-NMR. [d] Replace CO source with S3.



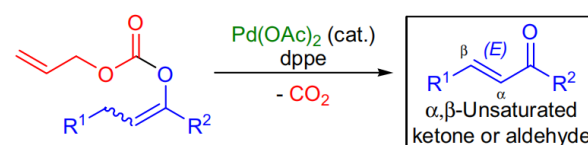
SAEGUSA OXIDATION

(References are on page 667)

Saegusa (1978):

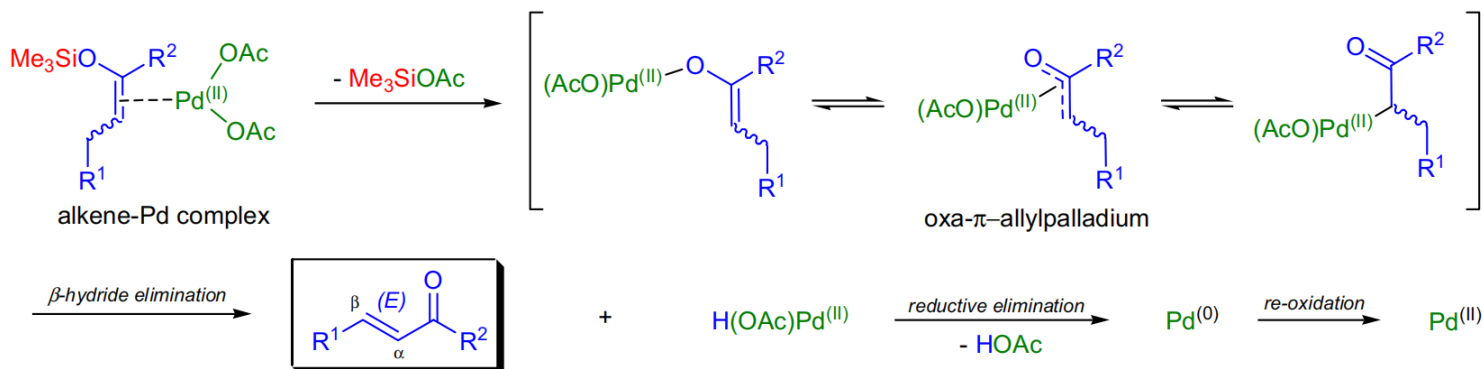
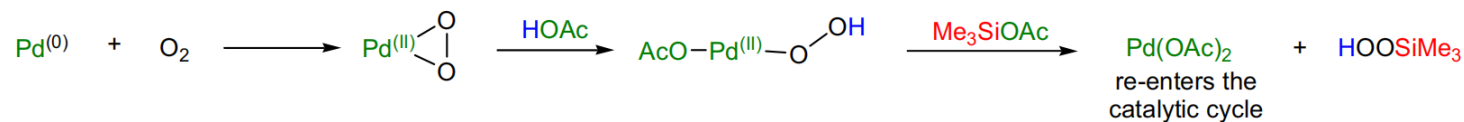
Catalytic process (*Larock modification*, 1995):

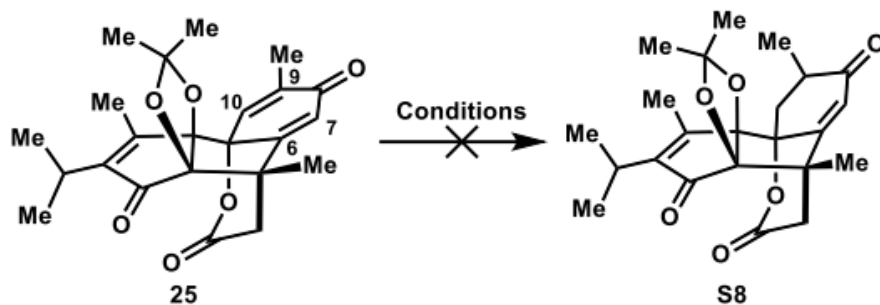
Allyl enol carbonate modification:



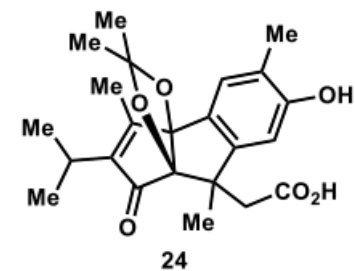
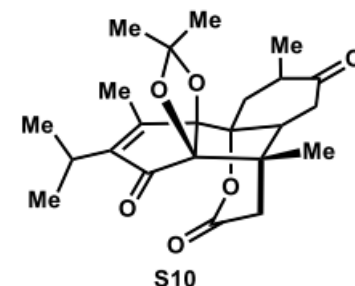
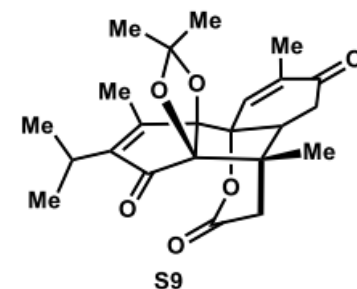
$\text{R}^{1-2} = \text{H, alkyl, aryl; SiR}_3 = \text{TMS, TBDMS; } n = 1-7$

Mechanism: ^{15,7}

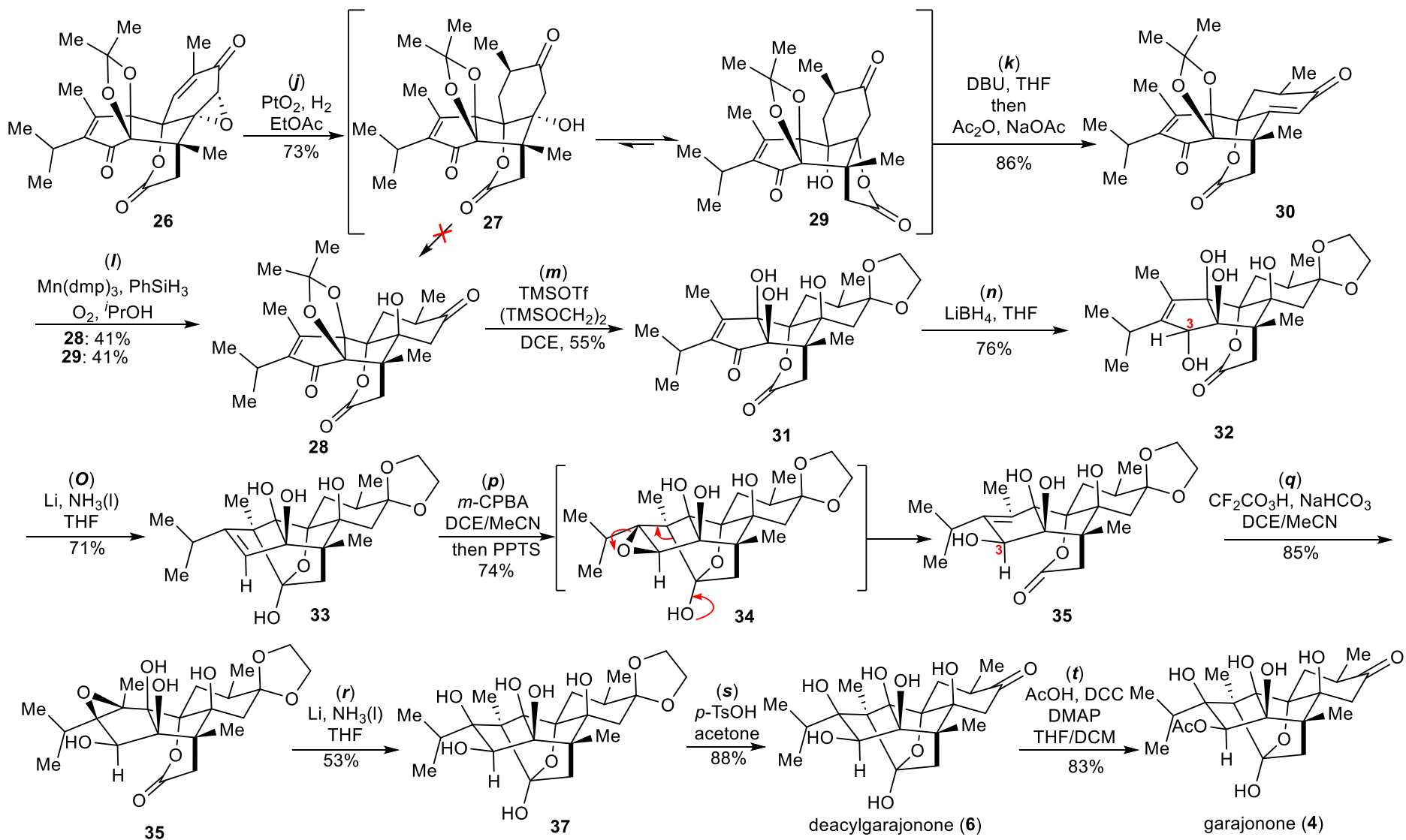
When substoichiometric/stoichiometric amounts of Pd(OAc)₂ is used:When the oxidation takes place under an oxygen atmosphere with catalytic amounts of Pd(OAc)₂:



Entry	Conditions	Result ^[a]
1	Ipr.HCl, CuCl, ^t BuONa, (EtO) ₃ SiH, PhMe, 23 °C	S9 (40%) + S10 (26%)
2	L-Selectride, THF, -78 °C	Decomp.
3	MAD, L-Selectride, PhMe, -78°C to 23 °C	NR
4	[(Ph ₃ P)CuH] ₆ , PhSiH ₃ , PhMe, 23 °C	S9 (58%) + S10 (24%)
5	RhCl(PPh ₃) ₃ , Et ₃ SiH, DCE, 0°C to 35 °C	S9 (70%) + S10 (13%)
6	RhCl(PPh ₃) ₃ , H ₂ balloon, DCM, 23 °C	S9 (68%) + S10 (10%)
7	Crabtree catalyst, H ₂ balloon, DCM, 23 °C	NR
8	Crabtree catalyst, H ₂ (3.0 Mpa), DCM, 23 °C	NR
9	KO ₂ CN=NCO ₂ K, HOAc, MeNO ₂ , 0 to 60 °C	NR
10	NiCl ₂ ·6H ₂ O, NaBH ₄ , MeOH, 0 °C	24 (86%)
11	Pd/C, H ₂ balloon, acetone, 0 to 23 °C	24 (90%)

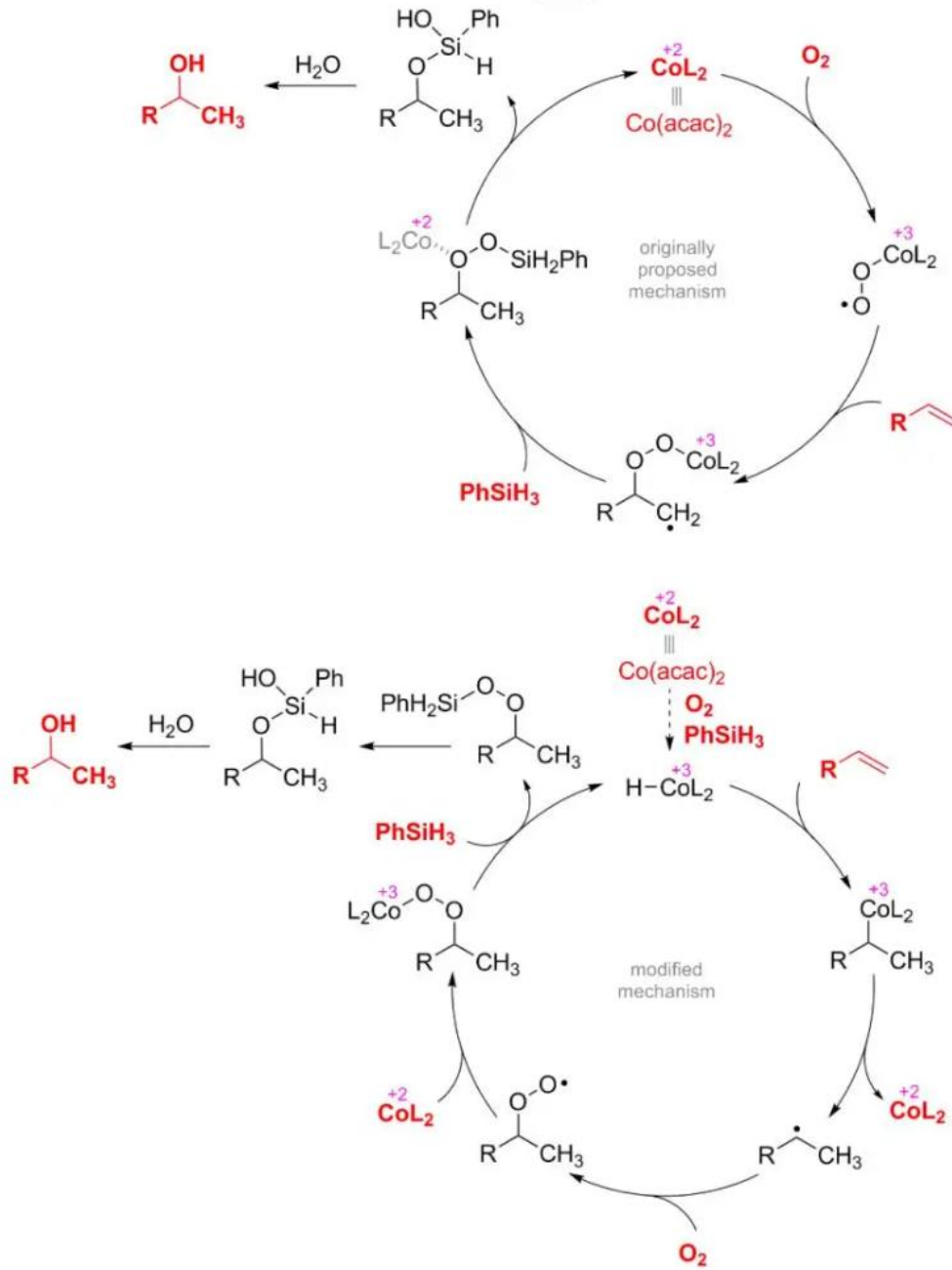


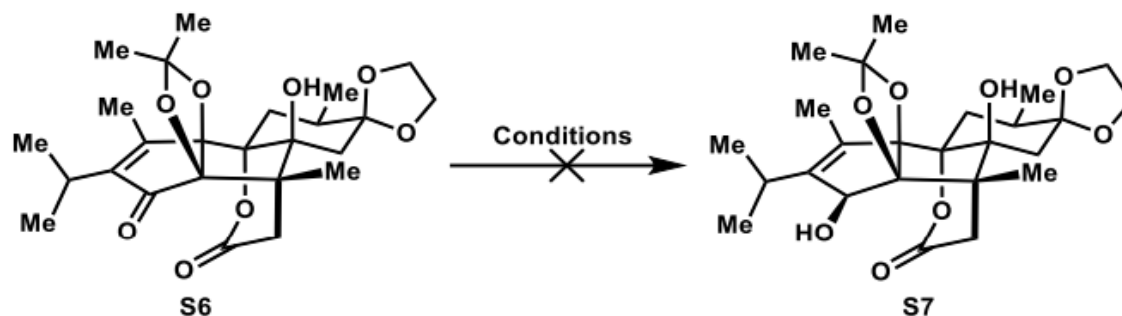
[a] NR = no reaction, MAD = methylaluminum bis(4-bromo-2,6-di(tert-butyl)phenoxide).



Mukaiyama Hydration

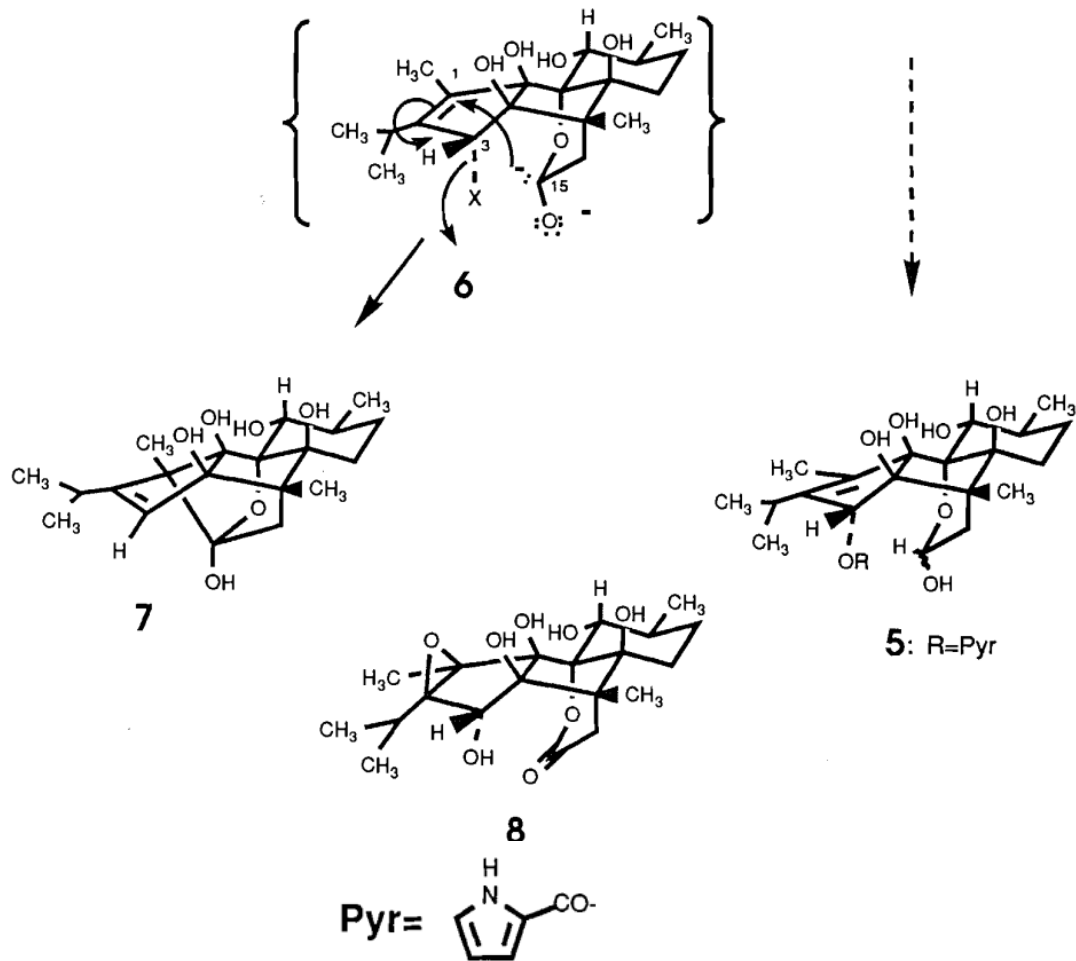
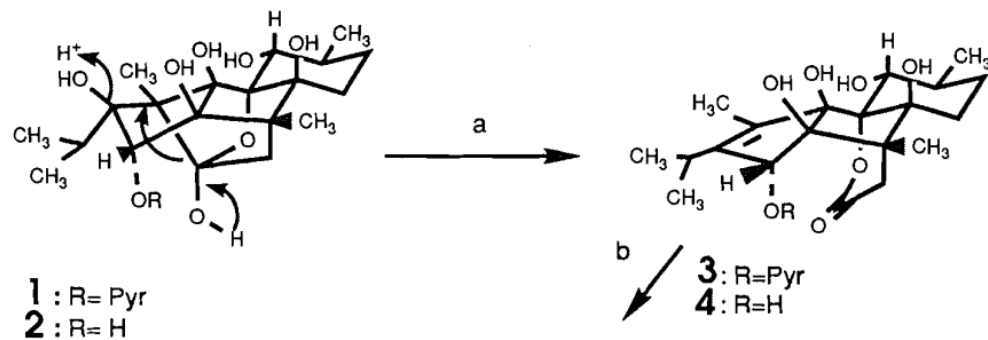
[1989]





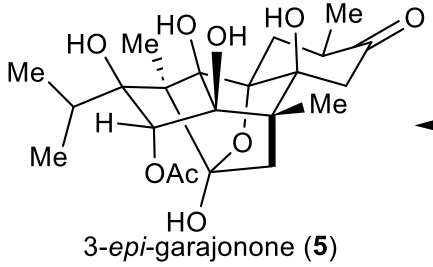
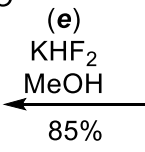
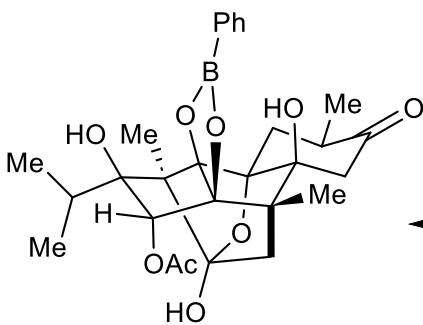
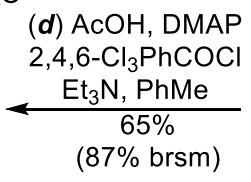
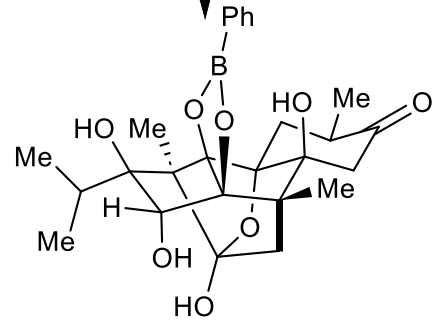
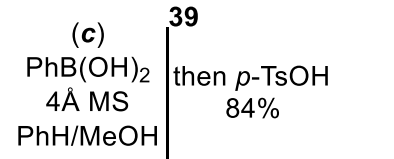
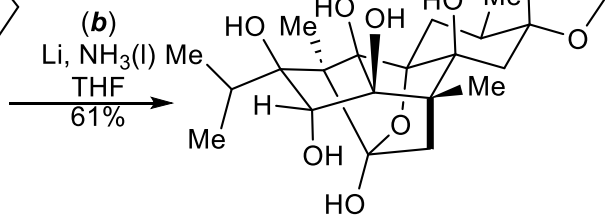
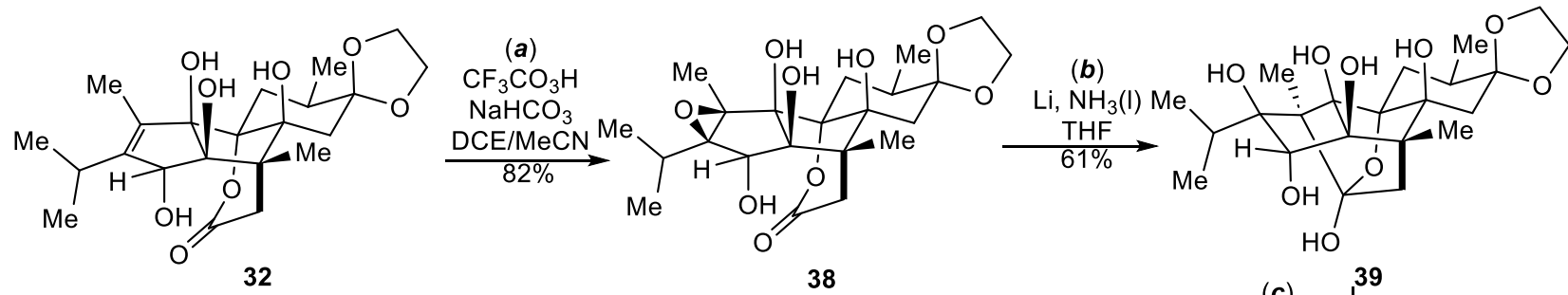
Entry	Conditions	Result ^[a]
1	LiBH ₄ , THF, -15 °C to 0°C	NR
2	NaBH ₄ , MeOH, 0 °C to 23°C	NR
3	NaBH ₄ , CeCl ₃ ·7H ₂ O, MeOH, 0 °C to 23°C	NR
4	L-selectride, THF, -15 °C to 0 °C	NR

[a] NR = no reaction



a: H₂SO₄, 2 N; b: Li, NH₃, THF, -78°C, 20–100 min

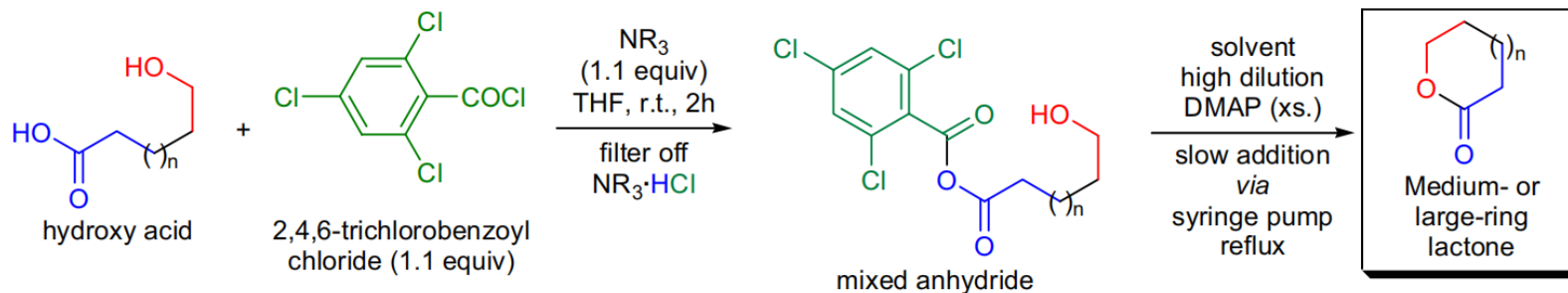
Can. J. Chem., 1993, 71, 634.



YAMAGUCHI MACROLACTONIZATION

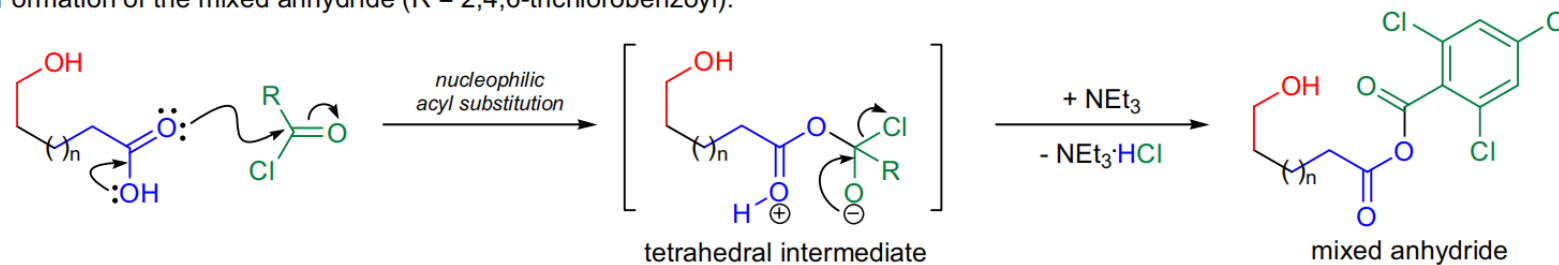
(References are on page 714)

Yamaguchi macrolactonization:



Mechanism:

Formation of the mixed anhydride (R = 2,4,6-trichlorobenzoyl):



Formation of the macrolactone (R = 2,4,6-trichlorobenzoyl):

